## CLASS – XI SUBJECT CHEMISTRY ASSIGNMENT NO. 13 TOPIC – HYDROGEN

- 1. Name the most abundant element in the universe?
- 2. Name the isotope of Hydrogen which finds use in nuclear reactor?
- 3. What is the importance of heavy water with regard to nuclear power generation?
- 4. Why does water have a high boiling point as compared to  $H_2S$ ?
- 5. What is the use of Hydrogen in manufacture of Vanaspati Ghee?
- 6. What is source of Sun's energy?
- 7. What is 'Syn' Gas, how it is produced, name the process of production of 'syngas'.
- 8. Name the catalyst used in manufacture of Methanol. give equation?
  9. Which isotope of hydrogen has no neutron?
- 10. Can we store H<sub>2</sub>O<sub>2</sub> in ordinary glass bottle? 11. What is tailing of Mercury?
- 12. H<sub>2</sub>O<sub>2</sub> is used to restore the colour of old paintings containing lead sulphide? Write a balanced equation for the reaction.
- 13. Name four negative catalysts that which retard the decomposition of H<sub>2</sub>O<sub>2</sub>?
- 14. Calculate molarity and normality of 20 volume of H<sub>2</sub>O<sub>2</sub>? What is its mass percentage?
- 15. What is the mass of H<sub>2</sub>O<sub>2</sub> present in 1 litre of a 2 M solu? Calculate the volume of oxygen at STB liberated upon the complete decomposition of 100 cm<sup>3</sup> of above solution?