Mineral Nutrition

1 INTRODUCTION

The basic needs of all living organisms are essentially the same. They require macro-molecules such as carbohydrates, proteins and fats, and water and minerals for their growth and development. This chapter focuses mainly on inorganic plant nutrition and the mechanism of biological nitrogen fixation.

2 Methods to Study the Mineral Requirements of Plants

In 1860, Julius von Sachs, a prominent German botanist demonstrated, for the first time, that plants could be grown to maturity in a defined nutrient solution in complete absence of soil. This technique of growing plants in a nutrient solution is known as hydroponics. By this method, essential elements were identified and their deficiency symptoms were discovered.

(3) Criteria for essentiality

- (a) The element must be absolutely necessary for supporting normal growth and reproduction.
- (b) The requirement must be specific and not replaceable by another element.
- (c) The element must be directly involved in the metabolism of the plant.

Based upon the above criteria only a few elements have been found to be absolutely essential for plant growth and metabolism. These elements are further divided into two broad categories based on their quantitative requirements, macronutrients and micronutrients.

4				
Macronutrients	Micronutrients			
 They are generally present in plant tissues in large amounts (in excess of 10 m mole kg¹ of dry matter) These include C, H, O, N, P, S, K, Ca, Mg 	in very small amoun			

Essential elements can also be grouped into four broad categories on the basis of their diverse functions. These categories are:

- (i) Components of biomolecules and structural (ii) elements of cells (C, H, O and N)
- (iii) Activate or inhibit certain enzymes, (Mg²⁺ activates RUBISCO and PEPcase, Zn²⁺ activate

(5) Role of Macro and Micronutrients

Essential elements perform several functions. They participate in various metabolic processes in the plant cells. Various forms and functions of essential nutrient elements are given below.

	ments are giv				
S.	Mineral	Absorbed	Required	Functions	
No.	element	as	in		
1.	Nitrogen	NO_3^- , NO_2^- or NH_4^+	All parts of the plants, particularly meristems	Major constituents of proteins, nucleic acids, vitamins and hormones	
2.	Phosphorus	H ₂ PO ₄ or HPO ₄ -	Developing fruits, seeds, storage organs, young meristems	Constituent of cell membrane, certain proteins, all nucleic acids	
3.	Potassium	K⁺	Meristems, buds, leaves and root tips	Maintain turgidity of cells, required for opening and closing of stomata	
4.	Calcium	Ca ²⁺	Meristems and differentiating tissues	Required for middle lamella, mitotic spindle and for certain enzymes. Accumulates in older leaves	
5.	Magnesium	Mg ²⁺	Seeds, leaves, growing areas of root and stem	Constituent of ring structure of chlorophyll and helps to maintain ribosome structure	
6.	Sulphur	SO ₄ ²⁻	Young leaves and meristems	Constituent of two amino acids cysteine and methionine and main constituents of several coenzymes and vitamins	
7.	Iron	Fe³⁺	All parts of plants	Main constituents of ferredoxin and cytochromes. It activates catalase enzyme and is essential for the formation of chlorophyll	
8.	Manganese	Mn²⁺	Leaves and seeds	The best defined function of manganese is in the splitting of water to liberate oxygen during photosynthesis	
9.	Zinc	Zn²⁺	All parts of the plants	Activates various enzymes especially carboxylases, also needed for auxin synthesis	
10.	Copper	Cu ²⁺	All parts of the plants	Associated with certain enzymes involved in redox reactions	
11.	Boron	BO ₃ ⁻ or B ₄ O ₇ ²⁻	Leaves and seeds	Required for uptake and utilisation of Ca²+, membrane functioning, pollen germination, cell elongation and carbohydrate translocation	
12.	Molybde- num	MoO ₂ ²⁺	All parts of plants and commonly in roots	Component of nitrogenase and nitrate reductase	
13.	Chlorine	CI	All parts of the plants	nitrate reductase Essential for water splitting reaction in photosynthesis and for anion and cation balance in cells	

alcohol dehydrogenase and Mo activate nitrogenase)

- Components of energy related chemical compounds in plants. (Mg in chlorophyll and P in ATP)
- r) Essential elements which alter osmotic potential (K).

6 Deficiency Symptoms of Essential Elements

The concentration of the essential element below which plant growth is retarded is termed as critical concentration. The element is said to be deficient when present below the critical concentration. Due to deficinecy of critical elements plants show deficiency symptoms. The parts of the plants that show the deficiency symptoms also depend on the mobility of the element in the plant.

7 Deficiency symptoms in plants include :

- (a) **Chlorosis**: Due to deficiency of N, K, Mg, S, Fe. Mn. Zn and Mo.
- (b) Necrosis: Due to deficiency of Ca, Mg, Cu, K
- (c) Inhibition of cell division: Due to deficiency of N, K, S, Mo
- (d) **Delay in flowering**: Due to deficiency of N, S, Mo.
- (e) Stunted plant growth.

Mobile Elements:

- Transported from older leaves to younger leaves.
- Deficiency symptoms first appear in older/senescent leaves. Example N, K, Mg Immobile Elements:
- Not transported out of the mature leaves.
- Deficiency symptoms tend to appear first in the young leaves. Example Ca.

8 Toxicity of micronutrients

The requirements of micronutrient is always in low amounts while their moderate decrease causes the deficiency symptoms and a moderate increase cause toxicity. Any mineral ion concentration it issues that reduces the dry weight of tissues by about 10 percent is considered toxic. Many a times, excess of an element may inhibit the uptake of another element.

For example, manganese competes with iron and Mg for uptake and with Mg for binding with enzymes. Mn also inhibit calcium translocation in shoot apex.

Apart from 17 essential elements there are 4 beneficial elements - Na, Si, Co, Se required by higher plants.

Biological

N_a fixation

Nitrosococcus⁶

(9) Mechanism of absorption of elements :

- Process of absorption can be demarcated into two phases.
- In the first phase, the rapid uptake of ions is into apoplast. It is passive.
 In the second phase the ions move into symplast. It requires energy; so it is active.

(Pseudomonas

(Thiobacillus)

Denitrification

Nitrobacter

Plant biomass

The movement of ions is called flux, the inward movement is influx and the outward movement is efflux.

(10) Translocation of solutes

Mineral salts are translocated through xylem along with the ascending stream of water, which is pulled up through plant by transpiration pull.

METABOLISM OF NITROGEN:

Nitrogen Cycle

Electrical

N_a fixation

(Uptake)

Atmospheric N₂

Nitrification

Soil 'N' Pool

Animal biomass

The nitrifying bacteria are chemoautotrophs.

Industrial

N. fixation

NH₃ Nitrosomonas NO₂

(Ammonification)

Decaying biomass

(11) Soil as Reservoir of Essential Elements

Soil consists of a wide variety of substances. Soil not only supplies minerals but also harbours nitrogen fixing bacteria.

(13) Biological nitrogen fixation

Only certain prokaryotic species are capable of fixing nitrogen. Reduction of nitrogen to ammonia by living organisms is called biological nitrogen fixation. Enzyme nitrogenase is exclusively present in prokaryotes.

$$N \equiv N \xrightarrow{\text{Nitrogenase}} NH_3$$

14 N, fixers	Bacteria	Nitrogen fixation	Aerobic/ anaerobic
	Azotobacter	Free living	Aerobic
	Beijerinckia	Free living	Aerobic
	Bacillus	Free living	Anaerobic
	Clostridium	Free living	Anaerobic
	Rhodospirillum	Free living	Anaerobic
	Anabaena	Symbiotic or	_
	Nostoc	free living	_
	Rhizobium	Symbiotic	-
	Frankia	Symbiotic	-

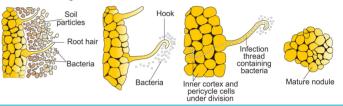
Rhizobium fixes nitrogen in legumes whereas Frankia in non-leguminous plants such as Alnus.

The ammonia synthesis by nitrogenase requires Product Substrate a very high input of energy 8 ATP for each NH₃ [nitrogen gas (N₂)] produced. The energy required thus, is obtained from the respiration of the host cells. of products Fnzvme Binding Free nitrogenase (nitrogenase) of substrate can bind another molecule of N Steps of conversion of atmospheric nitrogen to ammonia

15 Nodule formation

Nodule formation involves a sequence of multiple interactions between *Rhizobium* and roots of host plant. Principal steps in the nodule formation are as follows:

- Rhizobia multiply, colonise the surroundings of the roots and get attached to epidermal and root hair cells. Root hair curls and bacteria invade the root hair.
- An infection thread is produced carrying the bacteria into the cortex of the root where they initiate nodule formation in the cortex.
- The nodule thus formed, establishes a direct vascular connection with the host for exchange of nutrients.



- Nodule contains nitrogenase and leghaemoglobin.
- Nitrogenase is Mo-Fe protein and highly sensitive to the molecular oxygen, thus requires anaerobic conditions.
- To protect nitrogenase nodule contains an oxygen scavenger called laghaemoglobin.

Rhizobium and Frankia live as aerobes under free living condition but as symbionts during nitrogen fixing events they become anaerobic.

17) Fate of Ammonia

At physiological pH, the ammonia is protonated to form NH⁺₄ ions while most of the plants can assimilate nitrate as well as ammonia ions, the latter is quite toxic to plants and hence cannot accumulate in them.

NH₄ is used to synthesize amino acids in plants by two ways:

The two most important amides - Asparagine and glutamine found in plants are a structural part of proteins. They are formed from two amino acids namely aspartic and glutamic acid respectively.

Amides contain more nitrogen than amino acids, they are transported to other parts of the plant via xylem vessels. Nodules of some plants like sovabean export fixed nitrogen as ureides (High N : C ratio)