



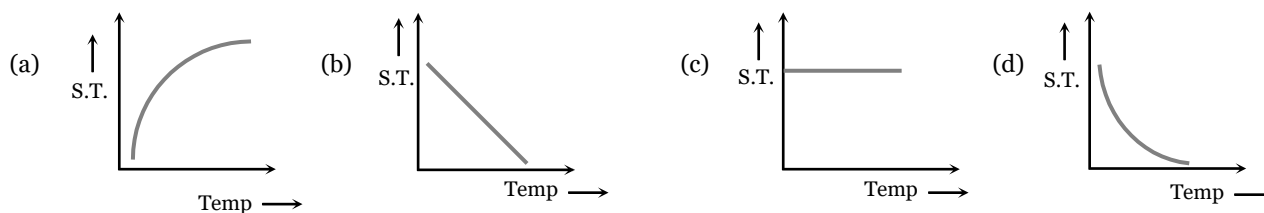
# Practice Problems

## Problems based on Cohesive and adhesive force

- Mercury does not wet glass, wood or iron because [MP PMT 1995; MP PET 1997]
  - Cohesive force is less than adhesive force
  - Cohesive force is greater than adhesive force
  - Angle of contact is less than  $90^\circ$
  - Cohesive force is equal to adhesive force
- The force of cohesion is [CPMT 1996]
  - Maximum in solids
  - Maximum in liquid
  - Same in different matters
  - Maximum in gases
- What enables us to write on the black board with chalk
  - Gravity
  - Cohesion
  - Adhesion
  - None of the above
- Intermolecular forces decrease rapidly as the distance between the molecules increases and do so much more
  - Slowly than demanded by the inverse square law of the distance
  - Rapidly than anticipated through the inverse square law of the distance
  - According to inverse square law
  - It actually remains the same for all the distances

## Problems based on Surface tension

- The spherical shape of rain-drop is due to [CPMT 1976, 90; CPMT 2001; NCERT 1982; AIIMS 1998; MHCET 2000; DCE 1999; AFMC 1999, 2001]
  - Density of the liquid
  - Surface tension
  - Atmospheric pressure
  - Gravity
- At which of the following temperatures, the value of surface tension of water is minimum [MP PMT/PET 1998]
  - $4^\circ C$
  - $25^\circ C$
  - $50^\circ C$
  - $75^\circ C$
- Force necessary to pull a circular plate of 5cm radius from water surface for which surface tension is 75 dynes/cm, is [MP PMT 1991]
  - 30 dynes
  - 60 dynes
  - 750 dynes
  - $750\pi$  dynes
- A square frame of side  $L$  is dipped in a liquid. On taking it out, a membrane is formed. If the surface tension of the liquid is  $T$ , the force acting on the frame will be [MP PMT 1990]
  - $2TL$
  - $4TL$
  - $8TL$
  - $10TL$
- Ball pen and fountain pen depend respectively upon the principle of
  - Surface tension and viscosity
  - Surface tension and gravity
  - Gravitation and surface tension
  - Surface tension and surface tension
- Which graph represents the variation of surface tension with temperature over small temperature ranges for water



- The material of a wire has a density of 1.4 g per  $cm^3$ . If it is not wetted by a liquid of surface tension 44 dyne per cm, then the maximum radius of the wire which can float on the surface of the liquid is
  - $\frac{1}{7}$  cm
  - 0.7 cm
  - $\frac{10}{14}$  cm
  - $\frac{10}{28}$  cm

## 70 Surface Tension

12. A water drop of  $0.05\text{cm}^3$  is squeezed between two glass plates and spreads into area of  $40\text{cm}^2$ . If the surface tension of water is  $70\text{ dyne/cm}$  then the normal force required to separate the glass plates from each other will be  
(a)  $90\text{ N}$  (b)  $45\text{ N}$  (c)  $22.5\text{ N}$  (d)  $450\text{ N}$
13. The main difference between a stretched membrane and the liquid surface is  
(a) The liquid surface has a tendency to contract but the stretched membrane does not  
(b) The surface tension does not depend on area but on the tension of the stretched membrane does  
(c) The surface tension increases with increases in area  
(d) Surface tension increases irregularly with temperature
14. On bisecting a soap bubble along a diameter, the force due to surface tension on any of its half part will be  
(a)  $4\pi RT$  (b)  $\frac{4\pi R}{T}$  (c)  $\frac{T}{4\pi R}$  (d)  $\frac{2T}{R}$
15. The addition of soap changes the surface tension of water to  $\sigma_1$  and that of sugar changes it to  $\sigma_2$ . Then  
(a)  $\sigma_1 = \sigma_2$  (b)  $\sigma_1 > \sigma_2$   
(c)  $\sigma_1 < \sigma_2$  (d) It is not possible to predict the above
16. A hollow disc of aluminum whose external and internal radii are  $R$  and  $r$  respectively, is floating on the surface of a liquid whose surface tension is  $T$ . The maximum weight of disc can be  
(a)  $2\pi(R+r)T$  (b)  $2\pi(R-r)T$  (c)  $4\pi(R+r)T$  (d)  $4\pi(R-r)T$

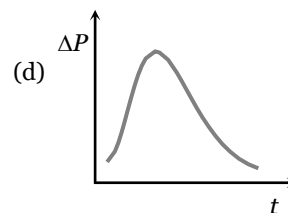
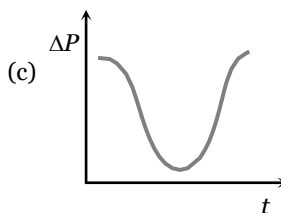
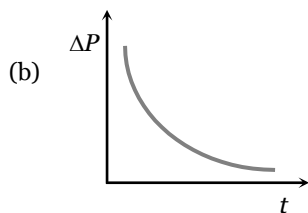
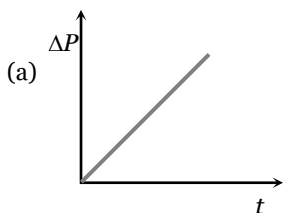
### Problems based on Surface energy

17. 8000 identical water drops are combined to form a big drop. Then the ratio of the final surface energy to the initial surface energy of all the drops together is  
(a)  $1:10$  (b)  $1:15$  (c)  $1:20$  (d)  $1:25$
18. 8 mercury drops coalesce to form one mercury drop, the energy changes by a factor of [DCE 2000]  
(a) 1 (b) 2 (c) 4 (d) 6
19. Which of the following statements are true in case when two water drops coalesce and make a bigger drop [Roorkee 1999]  
(a) Energy is released  
(b) Energy is absorbed  
(c) The surface area of the bigger drop is greater than the sum of the surface areas of both the drops  
(d) The surface area of the bigger drop is smaller than the sum of the surface areas of both the drops
20. An oil drop of radius  $1\text{cm}$  is sprayed into 1000 small equal drops of same radius. If the surface tension of oil drop is  $50\text{ dyne/cm}$  then the work done is [RPET 1990]  
(a)  $18\pi\text{ ergs}$  (b)  $180\pi\text{ ergs}$  (c)  $1800\pi\text{ ergs}$  (d)  $18000\pi\text{ ergs}$
21. If work  $W$  is done in blowing a bubble of radius  $R$  from a soap solution, then the work done in blowing a bubble of radius  $2R$  from the same solution is [MP PET 1990]  
(a)  $W/2$  (b)  $2W$  (c)  $4W$  (d)  $2\frac{1}{3}W$
22. A liquid drop of radius  $R$  is broken up into  $N$  small droplets. The work done is proportional to  
(a)  $N$  (b)  $N^{2/3}$  (c)  $N^{1/3}$  (d)  $N^0$
23. The work done in increasing the volume of a soap bubble of radius  $R$  and surface tension  $T$  by 700% will be  
(a)  $8\pi R^2 T$  (b)  $24\pi R^2 T$  (c)  $48\pi R^2 T$  (d)  $8\pi R^2 T^2 / 3$
24. 1000 drops of water all of same size join together to form a single drop and the energy released raises the temperature of the drop. Given that  $T$  is the surface tension of water,  $r$  the radius of each small drop,  $\rho$  the density of liquid,  $J$  the mechanical equivalent of heat. What is the rise in the temperature

- (a)  $T/Jr$  (b)  $10T/Jr$  (c)  $100T/Jr$  (d) None of these

### Problems based on Excess pressure

25. Two bubbles  $A$  and  $B$  ( $A > B$ ) are joined through a narrow tube. Then [UPSEAT 2001; Kerala (Med.) 2002]  
 (a) The size of  $A$  will increase (b) The size of  $B$  will increase  
 (c) The size of  $B$  will increase until the pressure equals (d) None of these
26. Excess pressure of one soap bubble is four times more than the other. Then the ratio of volume of first bubble to another one is [CPMT 1997; MH CET 2000]  
 (a) 1 : 64 (b) 1 : 4 (c) 64 : 1 (d) 1 : 2
27. The pressure of air in a soap bubble of 0.7 cm diameter is 8 mm of water above the pressure outside. The surface tension of the soap solution is [MP PET 1991; MP PMT 1997]  
 (a) 100 dyne/cm (b) 68.66 dyne/cm (c) 137 dyne/cm (d) 150 dyne/cm
28. An air bubble of radius  $r$  in water is at a depth  $h$  below the water surface at some instant. If  $P$  is atmospheric pressure,  $d$  and  $T$  are density and surface tension of water respectively, the pressure inside the bubble will be [Roorkee 1990]  
 (a)  $P + h dg - \frac{4T}{r}$  (b)  $P + h dg + \frac{2T}{r}$  (c)  $P + h dg - \frac{2T}{r}$  (d)  $P + h dg + \frac{4T}{r}$
29. A soap bubble is very slowly blown at the end of a glass tube by a mechanical pump which supplies a fixed volume of air every minute whatever the pressure against which it is pumping. The excess pressure  $\Delta P$  inside the bubble varies with time as shown by which graph



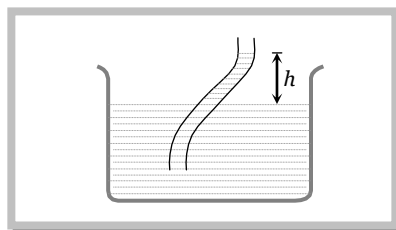
### Problems based on Angle of contact

30. A liquid does not wet the sides of a solid, if the angle of contact is [MP PAT 1990; AFMC 1988, MNR 1998, KCET 1998, Haryana CEE 1998; RPMT 1999; 2003]  
 (a) Zero (b) Obtuse (More than  $90^\circ$ ) (c) Acute (Less than  $90^\circ$ ) (d)  $90^\circ$
31. The meniscus of mercury in the capillary tube is [MP PET/PMT 1988]  
 (a) Convex (b) Concave (c) Plane (d) Uncertain
32. The angle of contact between glass and mercury is [MP PMT 1987]  
 (a)  $0^\circ$  (b)  $30^\circ$  (c)  $90^\circ$  (d)  $135^\circ$
33. When the temperature is increased the angle of contact of a liquid  
 (a) Increases (b) Decreases  
 (c) Remains the same (d) First increases and then decreases
34. For those liquids which do not wet the solid surface, the ratio of cohesive force and adhesive force will be  
 (a) Greater than  $\frac{1}{\sqrt{2}}$  (b) Greater than  $\sqrt{2}$  (c) Lesser than  $\frac{1}{\sqrt{2}}$  (d) Lesser than  $\sqrt{2}$
35. The water proofing agent makes an angle of contact  
 (a) From acute angle to obtuse angle (b) From obtuse angle to acute angle  
 (c) From obtuse angle to right angle (d) From acute angle to right angle

36. A glass plate is partly dipped vertically in the mercury and the angle of contact is measured. If the plate is inclined, then the angle of contact will
- (a) Increase (b) Remain unchanged (c) Increase or decrease (d) Decrease

### **Problems based on Capillarity**

37. The surface tension for pure water in a capillary tube experiment is [MH CET 2002]
- (a)  $\frac{\rho g}{2hr}$  (b)  $\frac{2}{hr\rho g}$  (c)  $\frac{r\rho g}{2h}$  (d)  $\frac{hr\rho g}{2}$
38. If capillary experiment is performed in vacuum then for a liquid there
- (a) It will rise (b) Will remain same (c) It will fall (d) Rise to the top
39. A surface tension experiment with a capillary tube in water is repeated in an artificial satellite. Which is revolving around the earth, water will rise in the capillary tube upto a height of [Roorkee 1992]
- (a) 0.1 m (b) 0.2 m (c) 0.98 m (d) Full length of the capillary tube
40. When a capillary is dipped in water, water rises to a height  $h$ . If the length of the capillary is made less than  $h$ , then [MP PAT 1990]
- (a) The water will come out (b) The water will not come out (c) The water will not rise (d) The water will rise but less than height of capillary
41. A long cylindrical glass vessel has a small hole of radius ' $r$ ' at its bottom. The depth to which the vessel can be lowered vertically in the deep water bath (surface tension  $T$ ) without any water entering inside is [MP PMT 1990]
- (a)  $4T/\rho rg$  (b)  $3T/\rho rg$  (c)  $2T/\rho rg$  (d)  $T/\rho rg$
42. Water rises to a height of 10cm in capillary tube and mercury falls to a depth of 3.112cm in the same capillary tube. If the density of mercury is 13.6 and the angle of contact for mercury is  $135^\circ$ , the ratio of surface tension of water and mercury is [MP PET/PMT 1988]
- (a) 1 : 0.15 (b) 1 : 3 (c) 1 : 6 (d) 1.5 : 1
43. Water can rise to a height  $h$  in a capillary tube lowered vertically into water. If the height of tube above the surface of water be  $l$  and  $l < h$ , then water will rise in the capillary to a height
- (a)  $h$  (b)  $l$  (c)  $l - h$  (d)  $l + h$
44. The height upto which water will rise in a capillary tube will be
- (a) Maximum when water temperature is  $4^\circ\text{C}$  (b) Maximum when water temperature is  $0^\circ\text{C}$  (c) Minimum when water temperature is  $4^\circ\text{C}$  (d) Same at all temperatures
45. The exact expression for surface tension of liquid which rises up in the capillary tube is
- (a)  $T = rhdg / 2$  (b)  $T = rhdg / 2 \cos \theta$  (c)  $T = \frac{r(h + r/3)dg}{2}$  (d)  $T = \frac{r(h + r/3)dg}{2 \cos \theta}$
46. If a wax coated capillary tube is dipped in water, then water in it will
- (a) Rise up (b) Depress (c) Sometimes rise and sometimes fall (d) Rise up and come out as a fountain
47. Capillaries made from various materials but having the same bore are dipped in the same liquid, then
- (a) Liquid will not rise in any of them (b) Liquid will rise in all upto same height (c) Liquid will not rise in all upto same height (d) Liquid will rise in all and height of liquid columns will be inversely proportional to the density of material used
48. A straight capillary tube is immersed in water and the water rises to 5cm. If the capillary is bent as shown in figure then the height of water column will be
- (a) 5cm (b) Less than 5cm (c) Greater than 5cm (d)  $4 \cos \alpha$

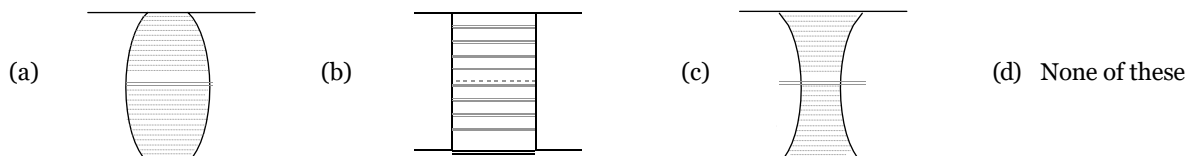


49. Water rises in a capillary tube through a height  $h$ . If the tube is inclined to the liquid surface at  $30^\circ$ , the liquid will rise in the tube upto its length equal to

- (a)  $\frac{h}{2}$  (b)  $h$  (c)  $2h$  (d)  $4h$

### Problems (Miscellaneous)

50. If a water drop is kept between two glass plates, then its shape is



51. When two soap bubbles of radius  $r_1$  and  $r_2$  ( $r_2 > r_1$ ) coalesce, the radius of curvature of common surface is [MP PMT 1996]

- (a)  $\frac{r_2 - r_1}{2}$  (b)  $\frac{r_2 - r_1}{r_1 r_2}$  (c)  $\frac{r_1 r_2}{r_2 - r_1}$  (d)  $\frac{r_2 + r_1}{2}$

52. Two soap bubbles of radius  $1\text{ cm}$  and  $2\text{ cm}$  coalesce to form a single drop under isothermal conditions. The total energy possessed by them if surface tension is  $30\text{ dyne cm}^{-1}$ , will be

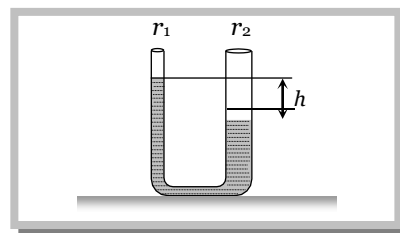
- (a)  $400\pi\text{ ergs}$  (b)  $600\pi\text{ ergs}$  (c)  $1000\pi\text{ ergs}$  (d)  $1200\pi\text{ ergs}$

53. In the above question, the radius of the bigger drop will be

- (a)  $\sqrt{3}\text{ cm}$  (b)  $\sqrt{5}\text{ cm}$  (c)  $\sqrt{7}\text{ cm}$  (d)  $\sqrt{8}\text{ cm}$

54. In a  $U$ -tube the radii of two columns are respectively  $r_1$  and  $r_2$  and if a liquid of density  $d$  filled in it has level difference of  $h$  then the surface tension of the liquid is

- (a)  $T = \frac{hdg}{r_2 - r_1}$   
 (b)  $T = \frac{(r_2 - r_1)hdg}{2}$   
 (c)  $T = \frac{(r_1 + r_2)hdg}{2}$   
 (d)  $T = \frac{hdg}{2} \frac{(r_1 r_2)}{r_2 - r_1}$





## Answer Sheet (Practice problems)

<b>1.</b>	<b>2.</b>	<b>3.</b>	<b>4.</b>	<b>5.</b>	<b>6.</b>	<b>7.</b>	<b>8.</b>	<b>9.</b>	<b>10.</b>
b	a	c	b	b	d	d	c	c	b
<b>11.</b>	<b>12.</b>	<b>13.</b>	<b>14.</b>	<b>15.</b>	<b>16.</b>	<b>17.</b>	<b>18.</b>	<b>19.</b>	<b>20.</b>
a	b	b	a	c	a	c	c	a, d	c
<b>21.</b>	<b>22.</b>	<b>23.</b>	<b>24.</b>	<b>25.</b>	<b>26.</b>	<b>27.</b>	<b>28.</b>	<b>29.</b>	<b>30.</b>
c	c	b	d	a	a	b	b	b	b
<b>31.</b>	<b>32.</b>	<b>33.</b>	<b>34.</b>	<b>35.</b>	<b>36.</b>	<b>37.</b>	<b>38.</b>	<b>39.</b>	<b>40.</b>
a	d	b	b	a	b	d	a	d	b
<b>41.</b>	<b>42.</b>	<b>43.</b>	<b>44.</b>	<b>45.</b>	<b>46.</b>	<b>47.</b>	<b>48.</b>	<b>49.</b>	<b>50.</b>
c	c	b	c	d	b	c	a	c	c
<b>51.</b>	<b>52.</b>	<b>53.</b>	<b>54.</b>						
c	d	b	d						