DPP - Daily Practice Problems

Na	ime :	Date :			
Start Time :		End Time :			
	CHEMI				
SYLLABUS: Bimolecules - II: Fats and Lipids, Vitamins, Hormones and Nucleic acids					
Max	. Marks : 120	Time : 60 min			
•	The Daily Practice Problem Sheet contains 30 MCQ's. For each bubble in the Response Grid provided on each page. You have to evaluate your Response Grids yourself with the he Each correct answer will get you 4 marks and 1 mark shall be diffined bubble is filled. Keep a timer in front of you and stop important the sheet follows a particular syllabus. Do not attempt the sheet response to the syllabus sheet in the starting of the book for the syllabus.	deduced for each incorrect answer. No mark will be given/ deducted inmediately at the end of 60 min. heet before you have completed your preparation for that syllabus. Ous of all the DPP sheets.			
ques out Q.1	ECTIONS (Q.1-Q.24): There are 24 multiple choice stions. Each question has 4 choices (a), (b), (c) and (d), of which ONLY ONE choice is correct. The 'acid value' of an oil or fat is measured in terms of weight of (a) NH ₄ OH (b) NaOH (c) KOH (d) CH ₃ COOH The 'saponification value' of an oil or fat is measured in terms of (a) NH ₄ OH (b) NaOH	 Q.3 The 'iodine value' of an oil indicates (a) Its boiling point (b) Inflammability (c) Unsaturation present in acid contents (d) Solubility of salt in oils Q.4 The most important food reserves of animals and plants ar (a) Carbohydrates (b) Proteins (c) Vitamins (d) Fats Q.5 The energy change produced by the combustion of food it called the 'calorific value'. The highest calorific value it given by 			

RESPONSE GRID 1. (a) (b) (c) d 2. (a) (b) (c) d 3. (a) (b) (c) d 4. (a) (b) (c) d 5. (a) (b) (c) d

(c) KOH

(d) C_6H_5OH

(a) Proteins

(c) Carbohydrates

(b) Fats

(d) Vitamins

13.abcd

18.(a)(b)(c)(d)

14.abcd

19. (a) (b) (c) (d)

15. (a)(b)(c)(d)

20. (a)(b)(c)(d)

12. a b c d

17. (a) (b) (c) (d)

RESPONSE

GRID

11.abcd

16.a b c d

Q.21 Vitamin B₆ is known as

- (a) Pyridoxin
- (b) Thiamine
- (c) Tocopherol
- (d) Riboflavin

Q.22 Which of the following is not a lipid?

- (a) Oils
- (b) Fats
- (c) Waxes
- (d) Proteins

Q.23 Which of the following indicates the number of free –OH groups in an oil or fat?

- (a) Iodine value
- (b) Acid value
- (c) Acetyl value
- (d) Saponification value

Q.24 Which of the following is not glyceride?

- (a) Lipids (simple)
- (b) Phospholipids
- (c) Sphingolipids
- (d) All of these

DIRECTIONS (Q.25-Q.27): In the following questions, more than one of the answers given are correct. Select the correct answers and mark it according to the following codes:

Codes:

- 1,2 and 3 are correct (a)
- (b) I and 2 are correct
- 2 and 4 are correct
- (d) 1 and 3 are correct

Q.25 Which of the following statements about the assembly of nucleotides in a molecule of deoxyribose nucleic acid (DNA) is incorrect?

- (1) A pentose of one unit connects to a pentose of another
- (2) A pentose of one unit connects to the base of another
- (3) A phosphate of one units connects to the base of another
- (4) A phosphate of one unit connects to a pentose of another

Q.26 Which of the following is a constituent of RNA?

- (1) Ribose
- (2) Phosphate
- (3) Adenine
- (4) Pyridinc

Q.27 Which of the following compounds do not belong to lipids?

- (1) Fats
- (2) Amino acids
- (3) Phospholipids
- (4) Carbohydrates

RESPONSE GRID

21. a b c d 22. a b c d

23. (a) (b) (c) (d)

24. a b c d 25. a b c d

26.abcd 27. (a) (b) (c) (d) DIRECTIONS (Q. 28-Q.30): Each of these questions contains two statements: Statement-1 (Assertion) and Statement-2 (Reason). Each of these questions has four alternative choices, only one of which is the correct answer. You have to select the correct choice.

- (a) Statement-1 is True, Statement-2 is True; Statement-2 is a correct explanation for Statement-1.
- (b) Statement-1 is True, Statement-2 is True; Statement-2 is NOT a correct explanation for Statement-1.
- (c) Statement -1 is False, Statement-2 is True.
- (d) Statement -1 is True, Statement-2 is False.

- Q.28 Statement-1: Valine is an essential amino acid.
 - **Statement-2:** The lack of essential amino acids in the diet causes Kwashiorkor.
- Q.29 Statement-1: DNA as well as RNA molecules are found in the nucleus of a cell.
 - **Statement-2:** On heating, the enzymes do not lose their specific activity.
- Q.30 Statement-1: ATPmolecules are energyrich molecules.

Statement-2: ATP consists of a purine base adenine, pentose sugarribose and a string of three phosphate groups.

RESPONSE GRID

28.abcd

29. (a) b) c) d)

30.abcd

DAILY PRACTICE PROBLEM SHEET 57 - CHEMISTRY				
Total Questions	30	Total Marks	120	
Attempted		Correct		
Incorrect		Net Score		
Cut-off Score	44	Qualifying Score	64	
Success Gap = Net Score — Qualifying Score				
Net Score = (Correct × 4) – (Incorrect × 1)				

110 DPP/ C (57)

DAILY PRACTICE PROBLEMS

CHEMISTRY SOLUTIONS

(57)

- 1. (c) Acid value is the number of mg of KOH required to neutralize 1 gm of the fator oil.
- 2. (c) Saponification value is the number of mg of KOH required to neutralize the fatty acid resulting from the complete hydrolysis of 1 gm of oil or fat.
- 3. (c) Iodine number is the number of gms of I₂ which combine with 100 gm of oil or fat. It shows the degree of unsaturation of acids in fat or oil.
- 4. (d) Fats are called energy bank of the body.
- 5. (b) I gm carbohydrate on oxidation gives 17 kJ of energy while 1 gm fat provides 37 kJ of energy.
- 6. (b)

$$\begin{array}{c|cccc} CH_2OOCR & CH_2OH \\ & & & & \\ \hline \\ CHOOCR & + & \\ & & & \\ \hline \\ CH_2OOCR & CH_2OH \\ \hline \\ Lipid or & Glycerol or \\ Triglyceride & Trighydric alcohol \\ \end{array}$$

- 7. (a) Oleic acid, stearic acid and palmitic acid are produced by the hydrolysis of fats hence called fatty acids.
- (d) Oleic acid C₁₇H₃₃COOH, linoleic acid-C₁₇H₃₁COOH, linolenic acid C₁₇H₂₉COOH, palmitic acid C₁₅H₃₁COOH.
 Saturated monocarboxylic acids form a homologous

Saturated monocarboxylic acids form a homologous series which has a general formula $C_nH_{2n+1}COOH$ or $C_nH_{2n}O_2$. Only palmitic acid follows this.

- 9. (d) Nitrogen base + Sugar + Phosphate = Nucleotide

 Nucleoside
- (c) Adenine is a purine base common in both RNA and DNA.
- 11. (c) Insulin hormone which is secreted by pancreas.

Ascorbic acid (vitamin C)

- 13. (h) Adenine = Thymine, Guanine = Cytosine2 hydrogenbonds3 hydrogenbonds
- 14. (c) Vitamin B₁ is thiamine. Its main source is cereals.
- 15. (a) Gene is a part of the DNA molecule that codes for a specific protein.

- **16. (d)** Cortisone is not a sex hormone, it regulates metabolism of fats, carbohydrates, proteins etc.
- 17. (d) Thymine is present only in DNA while in RNA there is uracil.
- 18. (a) Mutation is a chemical change in the sequence of nitrogenous bases along the DNA strained which can lead to the synthesis of protein with altered amino acid sequence.
- 19. (b) Insulin is a hormone secreted by pancreas. It lowers blood glucose level by promoting the uptake of glucose by cells and the conversion of excess of glucose to glycogen by liver and skeletal muscles.
- **20. (c)** Nucleoside on hydrolysis gives an aldopentose and a heterocyclic base (purine and pyrimidine).
- (a) Vitamin B₆ is called pyridoxin. It is found in fruits, greenvegetables, milk etc. Due to its deficiency, anaemia disease is caused.
- 22. (d) Proteins are polymers of amino acids.
- 23. (b)
- 24. (c) Sphingolipids are not glycerides.
- **25.** (a) A pentose of one unit connects to a pentose of another through phosphate group.
- 26. (a) Pyridine is not a constituent of RNA.
- 27. (c)
- **28. (b)** Valine is an essential amino acid. The amino acids which the body cannot synthesize are called essential amino acids.
- 29. (d) DNA is found mainly in the nucleus of the cell and RNA occurs mainly in the cytoplasm of the cell. So assertion given is false.

Enzymes are very good biological catalysts in certain temperature range but they lose their specific activity on heating.

Hence reason is also a wrong statement.

30. (b) ATP has four negatively charged oxygen atoms very close to each other. So the repulsive forces between them is high. On hydrolysis of ATP, a H₂PO₄⁻ ion is eliminated and the number of negatively charged oxygen atoms decreases. Thus, repulsive forces decreases and a large amount of energy is set free. When ATP changes to ADP, which in turn changes into AMP, energy is released at each step. This is how ATP can act as a source of energy.