# **CGPET 2022**

# **Solved Paper**

# **Question 1**

Two rods A and B of same dimension but different materials are welded together as shown in figure. Their thermal conductivities are K and 2K. The thermal conductivity of the composite rod will be



**Options:** 

A.  $\frac{2}{3}K$ 

B.  $\sqrt{2}K$ 

C. 3*K* 

D.  $\frac{4}{3}K$ 

Answer: D

## Solution:

**Solution:** In series, equivalent thermal resistance of series combination is  $R_S = R_1 + R_2$  $\Rightarrow \frac{l_1 + l_2}{K_s A} = \frac{l_1}{K_1 A} + \frac{l_2}{K_2 A}$  $\leftarrow I \rightarrow \leftarrow I \rightarrow$ K 2K

In given case,  $l_1 = l_2 = l$  and  $K_1 = K$ ,  $K_2 = 2K$  So, we have  $\frac{2l}{K_s A} = \frac{l}{KA} + \frac{l_2}{2KA}$   $\Rightarrow \frac{2}{K_s} = \frac{1}{K} + \frac{1}{2K}$   $\Rightarrow \frac{2}{K_s} = \frac{3}{2K}$ or  $K_s = \frac{4K}{3}$ 

-----

# **Question 2**

A rod of length L and mass M is bent into form of a semi-circular ring (as shown in figure). The moment of inertia about XY-axis is



#### **Options:**

A.  $\frac{ML^2}{2\pi^2}$ 

- B.  $\frac{ML^2}{\pi^2}$
- C.  $\frac{ML^2}{4\pi^2}$

D. 
$$\frac{2ML^2}{\pi^2}$$

## Answer: A

# Solution:

**Solution:** Moment of inertia of a semi-circular ring (using standard results) is  $I = \frac{1}{2}MR^2 \dots (i)$ As rod of length L is bent into semi-circular ring. Then,  $\pi R = L$  or  $R = \frac{L}{\pi}$ So, moment of inertia is given by  $I = \frac{1}{2}M \cdot (\frac{L}{\pi})^2$ or  $I = \frac{ML^2}{2\pi^2}$ 

\_\_\_\_\_

# **Question 3**

The average life 7 and the decay constant  $\lambda$  of a radioactive nucleus are related by the expression

## **Options:**

C

A.  $T\lambda = 1$ B.  $T = \frac{0.693}{\lambda}$ C.  $\frac{T}{\lambda} = 1$ D.  $T = \frac{C}{\lambda}$ 

#### **Answer:** A

## Solution:

```
Solution:
Decay constant \lambda and average life T are reciprocal of each other, i.e.
\lambda = \frac{1}{T}
\Rightarrow \lambda T = 1
```

# **Question 4**

A bar magnet of length / and magnetic dipole moment M is bent in the form of an arc as shown in figure. The new magnetic dipole moment will be



#### **Options:**

A.  $\frac{M}{2}$ 

B. *M* 

C.  $\frac{3M}{\pi}$ D.  $\frac{2M}{\pi}$ 

#### Answer: C

## Solution:

#### Solution:

Initially, magnetic dipole moment of magnet is  $M = \mu \cdot I$  $\leftarrow I \rightarrow$ 

After bending  $M' = \mu . I' ... (ii)$ 



Here, 
$$l = \frac{\pi}{3} \cdot r$$
 or  $\pi = \frac{3l}{\pi} \dots (iii)$   
 $\therefore M = \mu \times 2r \sin 30^{\circ}$   
 $= \mu \times 2 \times \frac{3l}{\pi} \times \frac{1}{2} \dots [From Eqs(iii)]$   
 $\frac{3\mu}{\pi} = \frac{3M}{\pi} \dots [From Eqs(i)]$ 



# **Question 5**

B - H curve of two different materials P and Q are shown in figure. These materials are used to make magnets for electric generator, transformer and electromagnetic core.

Then, it is proper to use



## **Options:**

- A. *P* for electric generators and transformer
- B. P for electromagnets and Q for electric generators
- C. P for transformers and Q for electric generators or electromagnets
- D. Q for electromagnets and transformers

#### **Answer: D**

## Solution:

Solution:

In B - H curve of material P has large retentivity and coercivity, so it can be used for making permanent magnet. Material Q has less retentivity and coercivity, so it can be used for making core of electromagnetic and transformers.

\_\_\_\_\_

# **Question 6**

For the angle of minimum deviation of a prism to be equal to its refracting angle, the prism must be made of a material whose refractive index

A. lies between  $\sqrt{2}$  and 1

B. lies between 2 and  $\sqrt{2}$ 

C. is less than 1

D. is greater than 2

Answer: B

## Solution:

Solution: Given,  $\delta_m = A$ Now, refractive index of material of prism is  $n = \frac{\sin\left(\frac{A + \delta_m}{2}\right)}{\sin\left(\frac{A}{2}\right)} = \frac{\sin\left(\frac{2A}{2}\right)}{\sin\left(\frac{A}{2}\right)}$   $= \frac{\sin A}{\sin\frac{A}{2}} = \frac{2\sin\frac{A}{2}\cos\frac{A}{2}}{\sin\frac{A}{2}}$ or  $n = 2\cos\frac{A}{2}$ As,  $0 \le \cos\frac{A}{2} \le 1$  $\Rightarrow 0 \le n \le 2$ 

\_\_\_\_\_

# **Question** 7

Three masses of 2 kg, 4 kg and 4 kg are placed at the three points (1, 0, 0), (1, 1, 0) and (0, 1, 0), respectively. The position vector of its centre of mass is

A.  $\frac{3}{5}\hat{i} + \frac{4}{5}\hat{j}$ B.  $3\hat{i} + \hat{j}$ C.  $\frac{2}{5}\hat{i} + \frac{4}{5}\hat{j}$ D.  $\frac{1}{5}\hat{i} + \frac{4}{5}\hat{j}$ 

**Answer:** A

## Solution:

Solution: 7. (a) Coordinates by centre of mass are given by  $\overline{x} = \frac{m_1 x_1 + m_2 x_2 + m_3 x_3}{m_1 + m_2 + m_3}$   $= \frac{2 \times 1 + 4 \times 0 + 4 \times 1}{2 + 4 + 4} = \frac{6}{10} = \frac{3}{5}$ and  $\overline{y} = \frac{m_1 y_1 + m_2 y_2 + m_3 y_3}{m_1 + m_2 + m_3}$   $= \frac{2 \times 0 + 4 \times 1 + 4 \times 1}{2 + 4 + 4} = \frac{8}{10} = \frac{4}{5}$ Also,  $\overline{z} = \frac{m_1 z_1 + m_2 z_2 + m_3 z_3}{m_1 + m_2 + m_3}$   $= \frac{2 \times 0 + 4 \times 0 + 4 \times 0}{2 + 4 + 4} = 0$ Position vector of Centre of mass is  $r = x^{\frac{1}{2}} + y^{\frac{1}{2}} + z^{\frac{1}{2}} = 35^{\frac{1}{2}} + \frac{4}{5}^{\frac{1}{2}}$ 

# **Question 8**

## The wavelength of X-rays is the order of

#### **Options:**

A. centimetre

- B. micron  $(10^{-6}m)$
- C. angstrom  $(10^{-10}m)$
- D. metre

```
Answer: C
```

## Solution:

Solution: X-rays are electromagnetic waves with wavelength range, 0.1Å to 100Å.

# **Question 9**

# Photoelectric cell is a device which converts

#### **Options:**

- A. light energy into electrical energy
- B. chemical energy into electrical energy
- C. electrical energy into light energy
- D. magnetic energy into electrical energy

#### Answer: A

## Solution:

**Solution:** A photoelectric cell or a photocell converts light energy into electrical energy.

-----

# **Question 10**

# The angular speed of a flywheel making 120 rev / min is

#### **Options:**

A.  $2\pi$  rad / s

B.  $4\pi^2$  rad / s

C.  $\pi\,\mathrm{rad}$  / s

D.  $4\pi$  rad / s

Answer: D

Solution:

```
Solution:
Angular speed
= 120 \text{ rpm}
= \frac{120 \times 2\pi}{60} rad / s = 4\pi rad / s
```

\_\_\_\_\_

# **Question 11**

The capacity of a parallel plate capacitor depends on

- A. nature of the plates
- B. distance between the plates
- C. thickness of the plates
- D. potential difference between the plates

Answer: B

## Solution:

```
Solution:
Capacity of a parallel plate capacitor is
C = \frac{\varepsilon_0 A}{d}where,
A =
area of plates and
d =
distance between plates.
So,
C \approx A
and
C \approx \frac{1}{d}
```

\_\_\_\_\_

# **Question 12**

A point mass *m* is placed inside a spherical shell of radius *R* and mass *M* at a distance  $\frac{R}{2}$  from the centre of the shell. The gravitational force exerted by the shell on the point mass is

#### **Options:**

A. 
$$\frac{GMm}{R^2}$$
  
B. 
$$-\frac{GMm}{R^2}$$

C. zero

D. 4  $\frac{GMm}{R^2}$ 

#### Answer: C

## Solution:

#### Solution:

Inside a hollow sphere, gravitational field is zero. So, force exerted on a mass placed anywhere inside shell is always zero.

-----

# **Question 13**

A ray of light travelling in a direction  $\frac{1}{2}(\hat{i} - \sqrt{3}\hat{j})$  is incident on a plane mirror. After reflection, it travels along the direction  $\frac{1}{2}(\hat{i} - \sqrt{3}\hat{j})$ . The angle of incidence is

#### **Options:**

A. 30 °

- B. 45 °
- C. 60 °
- D. 120°

#### Answer: C

## Solution:

Solution:

θ

Let  $\theta$  is angle between incident ray vector and reflected ray vector. Then, using  $\mathbf{A} \cdot \mathbf{B} = AB \cos \theta$ We have,  $\frac{1}{2}(\mathbf{\hat{i}} + \sqrt{3}\mathbf{\hat{j}}) \cdot \frac{1}{2}(\mathbf{\hat{i}} - \sqrt{3}\mathbf{\hat{j}})$   $= \frac{1}{2}(\sqrt{1^2 + 3}) \cdot \frac{1}{2}(\sqrt{1^2 + 3}) \cdot \cos \theta \Rightarrow \frac{1}{4}(1 - 3) = \frac{1}{4} \times 4 \times \cos \theta$   $\Rightarrow \cos \theta = \frac{-1}{2}$ or  $\theta = 120^{\circ}$ So, angle of incidence  $= \frac{\theta}{2}$  $60^{\circ}$ 

\_\_\_\_\_

# Question 14

The energy related per fission of a  $_{92}$ U<sup>235</sup> nucleus is nearly

#### **Options:**

A. 200 eV

B. 20 eV

C. 200 MeV

#### Answer: C

## Solution:

**Solution:** Energy released per fission of  ${}^{235}_{92}$ U is around 200 MeV.

\_\_\_\_\_

# **Question 15**

If the number of holes and electrons are  $n_h$  and  $n_e$  respectively, then the correct relation for pure semiconductor

#### **Options:**

A.  $n_h > n_e$ 

- B.  $n_h < n_e$
- C.  $n_h = n_e$
- D.  $n_h^2 = n_e^2$

Answer: C

## **Solution:**

#### Solution:

For a pure semiconductor number of holes and electrons are equal at every temperature.  $\therefore n_h = n_e$ 

\_\_\_\_\_

# **Question 16**

# The distance travelled by a particle starting from rest and moving with an acceleration $\frac{4}{3}$ m / $s^2$ . In the third second is

#### **Options:**

A.  $\frac{10}{3}$ m B.  $\frac{19}{3}m$ 

C. 6m

D. 4m

#### Answer: A

# Solution:

Solution: Given, u = 0,  $a = 4 / 3 \text{ms}^{-2}$ Distance travelled by particle in first 3s is  $s_3 = ut + \frac{1}{2}at^2$   $= \frac{1}{2} \times \frac{4}{3} \times 3^2 = 6\text{m}$ and distance travelled in first 2s is  $s_2 = \frac{1}{2} \times \frac{4}{3} \times 2^2 = 8 / 3\text{m}$ So, distance travelled in 3rd second  $= s_3 - s_2 = 6 - \frac{8}{3} = \frac{10}{3}\text{m}$ 

\_\_\_\_\_

# **Question 17**

# A spring of force constant 800N / m has an extension of 5 cm. The work done in extending it from 5 cm to 15 cm is

**Options:** 

A. 16J

B. 8J

C. 32J

D. 24J

**Answer: B** 

## Solution:

```
Solution:
For a spring work of extension is given by
W = \frac{1}{2}k(x_f^2 - x_i^2)Here, k =  spring constant =  800N / m
x_i =  initial extension =  5 cm
=  5 \times 10^{-2}m
x_f =  final extension =  15 cm
=  15 \times 10^{-2}m
So, work done,
W = \frac{1}{2} \times 800[(15 \times 10^{-2})^2 - (5 \times 10^{-2})^2]=  8]
```

\_\_\_\_\_

# **Question 18**

The dimensional formula of Boltzmann's constant is

- A.  $[ML^2T^{-2}\theta^{-1}]$
- B. [ML<sup>2</sup>T<sup>-2</sup>]
- C. [ML<sup>0</sup>T<sup>-2</sup> $\theta^{-1}$ ]
- D.  $[ML^{-2}T^{-1}\theta^{-1}]$

#### Answer: A

## Solution:

#### Solution:

Energy of a gas molecule per degree of freedom is  $E = \frac{1}{2}k_BT$ So, or  $k_B = \frac{2E}{T}$  $[k_B] = \frac{[E]}{[T]} = \frac{[ML^2T^{-2}]}{[\theta]}$  $= [ML^2T^{-2}\theta^{-1}]$ 

-----

# **Question 19**

The magnetic flux linked with a loop is  $\varphi = 6t^2 + 7t + 1$ , where  $\varphi$  is in milli weber and t is in second. What will be emf induced at t = 2s?

#### **Options:**

A. -49 mV

B. - 39 mV

 $C. -31 \, mV$ 

D. –19 mV

Answer: C

## Solution:

Solution: Induced emf is given by  $E = -\frac{d}{dt}\varphi_B$   $\Rightarrow E = -\frac{d}{dt}(6t^2 + 7t + 1) = -[12t + 7]$ So, E(at t = 2s) $= -(12 \times 2 + 7) = -31 \text{ mV}$ 

-----

# **Question 20**

A body of mass 1000 kg is moving horizontally with a velocity 50m / s. A mass of 250 kg is added. Find the final velocity.

#### **Options:**

A. 40m / s

B. 20m / *s* 

C. 30√2m / *s* 

D. 50m / *s* 

Answer: A

## Solution:

**Solution:** Momentum is conserved in the process of adding mass, so  $m_i v_i = m_f v_f$   $\Rightarrow 1000 \times 50 = (1000 + 250) v_f$   $\Rightarrow v_f = \frac{1000 \times 50}{1250}$  $= 40 \text{ms}^{-1}$ 

# **Question 21**

The mass M oscillates in simple harmonic motion with amplitude A as shown in figure. The amplitude of point P is



**Options:** 

A.  $\frac{A}{2}$ 

B. 2A

C.  $\frac{A}{3}$ 

D.  $\frac{2A}{3}$ 

#### Answer: D

## Solution:

#### Solution:

Let  $x_1$  and  $x_2$  are displacements of springs, then  $A = x_1 + x_2 \dots (i)$ Also, force on both springs is same  $\Rightarrow k_1 x_1 = k_2 x_2$   $\Rightarrow k x_1 = 2k x_2$  or  $x_2 = \frac{x_1}{2}$ Substituting in Eq.(*i*), we have  $x_1 = \frac{2}{3}A$ 

# **Question 22**

\_\_\_\_\_

Same force acts on two bodies of different masses 3 kg and 5 kg initially at rest. The ratio of times required to acquire same final velocity is

#### **Options:**

A. 5:3

B. 25:9

C. 9:25

D. 3:5

**Answer: D** 

#### Solution:

**Solution:** Let force on bodies is F, then accelerations of bodies are  $a_1 = \frac{F}{3}$  and  $a_2 = \frac{F}{5}$ Final velocities are given by v = u + at with u = 0; v = at $\therefore v_1 = a_1 t_1$ and  $v_2 = a_2 t_2$ As final velocities are given equal  $\Rightarrow v_1 = v_2$ or  $a_1 t_1 = a_2 t_2$  $\Rightarrow \frac{t_1}{t_2} = \frac{a_2}{a_1} = \frac{F/5}{F/3} = \frac{3}{5}$ 

\_\_\_\_\_

# **Question 23**

## The quantity which remains unchanged in a transformer is

#### **Options:**

A. voltage

B. current

C. frequency

D. None of these

Answer: C

# Solution:

C

**Solution:** In a transformer, voltage or current can be increased or decreased but input frequency is always equals to output frequency.

\_\_\_\_\_

# **Question 24**

# Electron-volt is the unit of

#### **Options:**

A. charge

B. potential

C. momentum

D. energy

Answer: D

## Solution:

Solution: Electron-volt is unit of energy,  $1 \text{ eV} = 1.6 \times 10^{-19} \text{ J}$ 

-----

# **Question 25**

A particles starts from rest. Its acceleration (a) versus time (t) graph is shown in the figure. The maximum speed of the particle will be



#### **Options:**

A. 110m / s

B. 55m/s

C. 550m/s

D. 660m/s

Answer: B

# Solution:

Solution: Final speed of particle  $= \frac{1}{2}at^2 = \text{Area under } a - t \text{ graph}$  $= \frac{1}{2} \times 10 \times 11 = 55 \text{m} / s$ 

\_\_\_\_\_

# **Question 26**

The number of electrons in 1C charge are

#### **Options:**

A.  $5.46 \times 10^{29}$ 

B.  $1.6 \times 10^{-19}$ 

C.  $625 \times 10^{18}$ 

D.  $9.5 \times 10^{-16}$ 

#### Answer: C

## Solution:

```
Solution:

Using Q = ne,

Q = 1C, e = 1.6 \times 10^{-19}C

We have,

n = \frac{Q}{e} = \frac{1}{1.6 \times 10^{-19}}

= 6.25 \times 10^{18} electrons
```

\_\_\_\_\_

# **Question 27**

A train moving at a speed of 220m/s towards a stationary object, emits a sound of frequency 1000 Hz. Some of the sound reaching the object gets reflected back to the train as echo. The frequency of the echo as detected by a person in the train is (velocity of sound in air = 330m/s)

#### **Options:**

A. 3500 Hz

 $B.\ 4000\,\text{Hz}$ 

 $C.\ 5000\,Hz$ 

D. 3000 Hz

**Answer: C** 

## Solution:

Frequency of reflected sound observed by an observer on moving train is  $f_2 = f_0(\frac{\nu + \nu_s}{\nu - \nu_s})$ Here,  $f_0$  = emitted sound's frequency  $\nu_s$  = speed of scource  $\nu$  = speed of sound So, frequency of reflected sound in given case will be  $f_2 = 1000(\frac{330 + 220}{330 - 220}) = 1000 \times \frac{550}{110}$ = 5000 Hz

# **Question 28**

A thermodynamic system undergoes cyclic process *ABCDA* as shown in figure. The work done by the system in the cycle is



## **Options:**

A.  $p_0 V_0$ B.  $2p_0 V_0$ C.  $\frac{p_0 V_0}{2}$ 

D. Zero

## Answer: D

## Solution:

#### Solution:

Work done in a cyclic process = Area under p - V graph.

Also, clockwise cycle means work done is positive and anti-clockwise cycle means work done is negative. In given cycle



# **Question 29**

# The lower end of a capillary tube of radius r is placed vertically in a water of density $\rho$ and surface tension S. The rise of water in the capillary tube is upto height h, then heat evolved is

**Options:** 

A. +  $\frac{\pi^2 r^2 h^2 \rho g}{J}$ B. +  $\frac{\pi r^2 h^2 \rho g}{2J}$ C. -  $\frac{\pi^2 r^2 h^2 \rho g}{2J}$ D. -  $\frac{\pi r^2 h^2 \rho g}{J}$ 

**Answer: B** 

## Solution:

Solution:

Heat evolved = Work done by force of surface tension Now, force due to surface tension, F = surface tension × circumference of tub  $\Rightarrow F = 2\pi r \times T$  ... [r = radius of tube] where, surface tension, T =  $\frac{h\rho gr}{2}$ Also, displacement of Centre of mass of liquid =  $\frac{h}{2}$ So, work done by force of surface tension W = force × displacement =  $2\pi r \times T \times \frac{h}{2}$ =  $2\pi r \times \frac{h\rho gr}{2} \times \frac{h}{2}$ 

 $\Rightarrow W = \frac{\pi r^2 h^2 \rho g}{2}$ Converting into units of heat,  $Q = \frac{\pi r^2 h^2 \rho g}{2J} \dots [\because Q = \frac{W}{J}]$ 

\_\_\_\_\_

# **Question 30**

# The current gain of transistor in CE mode is 49. What will be the current gain of the same transistor in CB mode?

#### **Options:**

A. 49

B. 0.98

C. 4.9

D. 98

#### **Answer: B**

## Solution:

**Solution:** Relation between current gain in CE mode ( $\beta$ ) and current gain in CB mode ( $\alpha$ ) is  $\alpha = \frac{\beta}{1+\beta}$ Here, given  $\beta = 49$  $\therefore \alpha = \frac{49}{1+49} = \frac{49}{50} = 0.98$ 

#### \_\_\_\_\_

# **Question 31**

## In the given circuit, currents $|_1$ and $|_2$ are respectively



#### **Options:**

A.  $\frac{8}{7}A$ ,  $\frac{12}{5}A$ B.  $\frac{8}{7}A$ ,  $\frac{12}{7}A$ 

C. 
$$\frac{10}{7}A$$
,  $\frac{12}{5}A$   
D.  $\frac{10}{7}A$ ,  $\frac{12}{7}A$ 

#### Answer: C

## Solution:

Solution:



By KVL in loop ABEFA, we have  $\Sigma V = 0$   $\Rightarrow 6 + 4 - 7I_1 = 0$   $\Rightarrow I_1 = \frac{10}{7}A$ In loop BCDEB,  $\Rightarrow 8 + 4 - 5I_2 = 0$  $I_2 = \frac{12}{5}A$ 

-----

# **Question 32**

The root mean square velocity of hydrogen molecules at 300K is 1930m / *s*. The rms velocity of oxygen molecules at 1200K will be

#### **Options:**

A. 765m / *s* 

- B. 1065m / s
- C. 965m / s
- D. 865m / s

**Answer: C** 

## Solution:

Solution:  $v_{\text{rms}} \propto \frac{\sqrt{T}}{M}$   $\Rightarrow \frac{v_2}{v_1} = \frac{\sqrt{T_2}}{T_1} \times \frac{\sqrt{M_1}}{M_2}$ Here,  $v_1 = 1930 \text{m} / sT_1 = 300 \text{K}$ ,  $M_1 = 2 \text{ unit}$ ,  $T_2 = 1200 \text{K}$ ,  $M_2 = 32 \text{ unit}$  $\therefore v_2 = v_1 \sqrt{\left(\frac{T_2}{T_1} \times \frac{M_1}{M_2}\right)}$ = 1930 ×  $\sqrt{\left(\frac{1200}{300} \times \frac{2}{32}\right)}$ = 1930 ×  $\sqrt{\frac{8}{32}}$  $=\frac{1930}{2}=965 \text{m}/s$ 

# **Question 33**

A thin convex lens made from crown glass ( $\mu = \frac{3}{2}$ ) has focal length f. When it is measured in two different liquids having refractive indices  $\frac{4}{2}$ and  $\frac{5}{3}$ , it has the focal lengths  $f_1$  and  $f_2$ , respectively. The correct relation between the focal lengths, is

#### **Options:**

A.  $f_2 > f_1 f_1$  become negative

B.  $f_1$  and  $f_2$  both become negative

C.  $f_1 > f$  and  $f_2$  become negative

D.  $f_1 = f_2 < f$ 

#### **Answer: C**

## Solution:

#### Solution:

If lens is dipped in a medium of higher refractive index, then its nature is changed. In given case,  $>\frac{3}{2}$ 

So, convex lens behaves like a concave lens in second medium. Hence,  $f_2$  is negative. Also, when lens is in first medium ( $n = \frac{4}{3}$ ), its power reduces or  $f_1 > f$ .

# **Question 34**

The escape velocity of a body from the surface of the earth is  $v_e$ . The escape velocity of the same body from a height equal to 7*R* from earth's surface will be

#### **Options:**

A.  $\frac{v_e}{\sqrt{2}}$ B.  $\frac{v_e}{2}$ V۵

C. 
$$\frac{e}{2\sqrt{2}}$$

D. 
$$\frac{v_e}{4}$$

#### **Answer: C**

## Solution:

#### Solution:

 $=\frac{1}{2\sqrt{2}}\cdot v_e$ 

Let M = mass, R<sub>e</sub> = radius of earth, then  $v_e = \frac{\sqrt{2GM}}{R_e}$ Now, when object is at a height of 7R from earth's surface, let escape speed needed is  $v_e$ , then by energy conservation, PE + KE at separation ' 8R' from earth's Centre = Total energy at infinity  $\Rightarrow \frac{-GMm}{8R_e} + \frac{1}{2}m(v_e)^2 = 0$  $\Rightarrow \frac{1}{2}m(v_e)^2 = \frac{GMm}{8R_e}$  $\Rightarrow v_e = \frac{1}{2\sqrt{2}} \cdot \left(\sqrt{\frac{2GM}{R_e}}\right)$ 

\_\_\_\_\_

# **Question 35**

The speed of light (c), gravitational constant (G) and Planck's constant (*h*) are taken as the fundamental units in a system. The dimension of time in this new system should be

**Options:** 

A. 
$$\left[G^{\frac{1}{2}}h^{\frac{1}{2}}c^{\frac{-5}{2}}\right]$$
  
B.  $\left[G^{\frac{-1}{2}}h^{\frac{1}{2}}c^{\frac{1}{2}}\right]$ 

C. 
$$\begin{bmatrix} G^{\frac{1}{2}}h^{\frac{1}{2}}c^{\frac{-3}{2}} \end{bmatrix}$$
  
D.  $\begin{bmatrix} G^{\frac{1}{2}}h^{\frac{1}{2}}c^{\frac{1}{2}} \end{bmatrix}$ 

**Answer:** A

## Solution:

Solution: We have,  $T = G^{a} \cdot h^{b} \cdot c^{c}$   $\Rightarrow T = [M^{-1}L^{3}T^{-2}]^{a}[ML^{2}T^{-1}]^{b}[LT^{-1}]^{c}$ Equating dimensions of similar physical quantities, we have  $-a + b = 0 \dots (i)$   $3a + 2b + c = 0 \dots (ii)$   $-2a - b - c = 1 \dots (iii)$ Adding Eqs. (*ii*) and (*iii*), we have  $a + b = 1 \dots (iv)$ From adding Eqs. (*i*) and (*iv*), we have  $b = \frac{1}{2}$   $\Rightarrow a = \frac{1}{2}$ Hence,  $c = \frac{-5}{2}$ So, we can write  $[T] = [G^{1/2}h^{1/2}c^{-5/2}]$ 

\_\_\_\_\_

# **Question 36**

At two points *P* and *Q* on screen in Young's double slit experiment, wave from slits *S*<sub>1</sub> and *S*<sub>2</sub> have a path difference of 0 and  $\frac{\lambda}{4}$ , respectively. The ratio of intensities at *P* and *Q* will be

#### **Options:**

A. 3:2

B. 2:1

C. √2:1

D. 4:1

Answer: B

## Solution:

**Solution:** Intensity in YDSE is given by  $I_1 + I_2 + 2\sqrt{I_1}\sqrt{I_2}\cos\varphi$ where,  $\varphi =$  phase difference. Here at point P,  $I_1 = I_2 = I_0$  and  $\varphi = 0$ 

$$\Rightarrow I_{P} = 4I_{0}$$
  
At point Q ,  
$$I_{1} = I_{2} = I_{0}$$
$$\Delta L = \frac{\lambda}{4} \text{ So } \varphi = \frac{2\pi}{4} = \frac{\pi}{2}$$
$$\Rightarrow I_{Q} = 2I_{0}$$
$$\therefore \frac{I_{P}}{I_{Q}} = \frac{4}{2} = \frac{2}{1}$$

# **Question 37**

## 3 moles of oxygen are mixed with 2 moles of helium. What will be the ratio of specific heats at constant pressure and constant volume for the mixture?

#### **Options:**

A. 6.7

B. 1.5

C. 4

D. None of these

#### **Answer: B**

## Solution:

#### Solution:

Solution: For a gas mixture,  $C_{\rho_{mix}} = \frac{n_1 C_{\rho_1} + n_2 C_{\rho_2}}{n_1 + n_2}$ and  $C_{V_{mix}} = \frac{n_1 C_{V_1} + n_2 C_{V_2}}{n_1 + n_2}$ Here,  $n_1 = 3, n_2 = 2$   $C_{V_1} = \frac{f}{2}R = \frac{5}{2}R; C_{V_2} = \frac{3}{2}R$   $C_{\rho_1} = (\frac{f}{2} + 1)R = \frac{7}{2}R;$   $C_{\rho_2} = \frac{5}{2}R$  $C_{p_{1}} = \frac{C_{2}}{2} + \frac{1}{2} + \frac{1}{2}$ 

# **Question 38**

Three conducting rods of same material and cross-section are shown in figure. Temperature of A, D and C are maintained at 20°C, 90°C and

 $\mathbf{0}\,^\circ\,\mathbf{C}$  , respectively. The ratio of the lengths of BC and BD , if there is no flow in *AB* is



## **Options:**

- A.  $\frac{2}{9}$
- B. <u>9</u>2
- C.  $\frac{2}{7}$
- D.  $\frac{7}{2}$

# Answer: C

# Solution:

Solution:

$$\begin{array}{c} (20^{\circ}\text{C}) \\ B & \swarrow & l_1 \longrightarrow \\ A(20^{\circ}\text{C}) \\ & & & \\ A(20^{\circ}\text{C}) \\ & & & \\ I_2 \\ & & \\ D(90^{\circ}\text{C}) \end{array}$$

----

As there is no heat flow in AB this means junction B is at 20 °C. Hence, heat in flow from DB = heat out flow from BC.  $\Rightarrow \frac{H_{DB} = H_{BC}}{I_2}$   $\Rightarrow \frac{kA(\theta_D - \theta_B)}{I_2} = \frac{kA(\theta_B - \theta_C)}{I_1}$ As rods are of same material and area,  $\frac{\theta_D - \theta_B}{I_2} = \frac{\theta_B - \theta_C}{I_1}$   $\Rightarrow \frac{I_1}{I_2} = \frac{\theta_B - \theta_C}{\theta_D - \theta_B} = \frac{20 - 0}{90 - 20} = \frac{2}{7}$ 

\_\_\_\_\_

# **Question 39**

# The force on a charge q placed in a uniform electric field E will be

#### **Options:**

A.  $\frac{E}{q}$ B. qEC. E

D.  $\frac{q}{E}$ 

Answer: B

## Solution:

Solution: As,  $E = \frac{F}{q}$  $\Rightarrow F = qE$ 

-----

# **Question 40**

# Wattless current means

**Options:** 

- A. current is zero
- B. the average power consumed in a cycle is zero
- C. emf is zero
- D. the phase difference between current and potential difference is zero

Answer: B

## Solution:

#### Solution:

Wattless current means current of inductor or capacitor due to which there is no power loss. Average power consumed in an inductor or capacitor is zero.

-----

# **Question 41**

# The hydrogen atom is in d-state. For this state, values of m are

#### **Options:**

A. 2, 1, 0

B. -1,0,1

C. -2, -1, 0, 1, 2

D. -1, -2, -3, -4

#### Answer: C

## Solution:

#### Solution:

```
In d-state, n = 3
Hence, for a hydrogen atom is in d-state, I = 2 and values of magnetic quantum number m are
m = (2/+1) = 2(2) + 1 = 5
\therefore m = -2, 1, 0, 1, 2
```

\_\_\_\_\_

# **Question 42**

A uniform cylindrical rod of length L, cross-sectional area A and Young's modulus Y is acted upon by the forces as shown in figure. The elongation of the rod is



#### **Options:**

- A.  $\frac{2FL}{3AY}$
- B.  $\frac{3FL}{5AY}$
- C.  $\frac{3FL}{8AY}$
- D.  $\frac{8FL}{3AY}$

Answer: A

## Solution:

**Solution:** (Note *Q* does not match with diagram given)



Net force on portion of length  $\frac{L}{3} = 2F$ So, Young's modulus is

$$Y = \frac{\frac{2F}{A}}{\Delta L / (L/3)} = \frac{2}{3} \cdot \frac{FL}{A \Delta L}$$
$$\Rightarrow \Delta L = \frac{2}{3} \cdot \frac{F \cdot L}{A \Delta L}$$

\_\_\_\_\_

# **Question 43**

If the self-inductance of two coils are  $L_1$  and  $L_2$ , respectively. The maximum possible mutual inductance is

#### **Options:**

A.  $L_1 L_2$ B.  $\frac{L_1}{L_2}$ C.  $\frac{\sqrt{L_1}}{L_2}$ 

D.  $\sqrt{L_1 L_2}$ 

## Answer: D

# Solution:

Solution:

For two <u>coils</u>, mutual inductance and self-inductances are related as  $M = k\sqrt{L_1 \cdot L_2}$ where, k = coefficient of magnetic coupling. Now, M is <u>maximum</u> when k = 1 $\Rightarrow M_{max} = \sqrt{L_1 \cdot L_2}$ 

# **Question 44**

A particle of mass *m* is executing oscillations about the origin on the *X*-axis. Its potential energy is  $U(x) = k|x|^3$ , where *k* is a positive constant. If the amplitude of oscillation is *a*, then the time period *T* is

#### **Options:**

A. proportional to  $\frac{1}{\sqrt{a}}$ 

B. independent of a

C. proportional to  $\sqrt{a}$ 

D. proportional to  $a^{3/2}$ 

**Answer:** A

## Solution:

#### Solution:

Maximum potential energy,  $U_{max} = ka^3$  (..Given) But  $U_{max} = \frac{1}{2}m\omega^2a^2$ Equating, we have  $\frac{1}{2}m\omega^2a^2 = ka^3$   $\Rightarrow \omega^2 = \frac{2ka}{m} \Rightarrow \frac{4\pi^2}{T^2} = \frac{2ka}{m}$  $\Rightarrow T = \sqrt{\frac{4\pi^2m}{2ka}}$  or  $T \propto \frac{1}{\sqrt{a}}$ 

\_\_\_\_\_

# **Question 45**

A heavy uniform chain lies on a horizontal table-top. If the coefficient of friction between the chain and table surface is 0.25, then the maximum fraction of length of the chain, that can hang over one edge of the table is

Options:	
A. 20 %	
B. 25 %	
C. 35 %	
D. 15 %	
Answer: A	

# Solution:

#### Solution:

Let  $(\frac{1}{n})$  th part of chain hangs in equilibrium. Then,  $(1 - \frac{1}{n})$  th part lies on table. In equilibrium, downward gravitational pull = Force of friction.  $\Rightarrow (\frac{1}{n}) \cdot L \cdot x \cdot g = (1 - \frac{1}{n}) \cdot L \cdot x \cdot g \cdot \mu$ Here, x = mass per unit length, L = length of complete chain. So,  $n - 1 = \frac{1}{\mu}$   $\Rightarrow n = 1 + \frac{1}{\mu} = 1 + \frac{1}{0.25} = 5$  $\therefore \frac{1}{5}$  th part of chain = 20% of total length can over hang.

**Question 46** 

A mass *m* is performing linear simple harmonic motion, then which of the following graph represents correctly the variation of acceleration *a* corresponding to linear velocity v ?

**Options:** 



D.



#### Answer: D

#### Solution:

#### Solution:

In SHM , velocity,  $v = \omega \sqrt{A^2 - y^2}$   $v^2 = \omega^2 (A^2 - y^2) \dots (i)$ and acceleration,  $a = -\omega^2 y$   $\Rightarrow a^2 = \omega^4 y^2 \dots (ii)$ From Eqs.(*i*) and(*ii*), we have  $\Rightarrow v^2 = \omega^2 A^2 - \omega^2 y^2$   $\Rightarrow v^2 = \omega^2 A^2 - \omega^2 a^2$   $\Rightarrow v^2 = -\omega^2 a^2 + \omega^2 A^2$ Comparing with y = mx + cAbove equation is a straight line with negative slope and positive intercept.

# **Question 47**

If mass of a body, *M* on the surface of the earth, then the mass of the same body on the surface of the moon is

**Options:** 

A.  $\frac{M}{6}$ 

B. Zero

C. *M* 

D. None of these

**Answer: C** 

## Solution:

Solution:

Mass is quantity of matter and its value doesnot change from place-to-place.

\_\_\_\_\_

\_\_\_\_\_

# **Question 48**

A magnet is cut in three equal parts by cutting it perpendicular to its length. The time period of original magnet is  $T_0$  in a uniform magnetic field B.

Then, the time period of each part in the same magnetic field is

C

#### **Options:**

- A.  $\frac{T_0}{2}$ B.  $\frac{T_0}{3}$ C.  $\frac{T_0}{4}$
- D. *T*<sub>0</sub>

## Answer: D

# Solution:

#### Solution:

Let magnetic moment of magnet is M and its moment of inertia is I. Then, magnetic moment of each part will be  $\frac{M}{3}$ , and moment of inertia of each part will be  $\frac{I}{3}$ .



-----

# **Question 49**

For a satellite, if the time of revolution is *T*, then its kinetic energy is proportional to

## **Options:**

A. 
$$\frac{1}{=}$$
  
B.  $\frac{1}{0}$   
C.  $\frac{1}{0}$   
D.  $T^{-\frac{2}{3}}$ 

#### Answer: D

## Solution:

Solution: We have,  $E_{k} = \frac{1}{2}mv^{2}$ where,  $v = \frac{2\pi r}{T}$   $\therefore E_{k} = \frac{1}{2}m(\frac{2\pi r}{T})^{2}$   $\Rightarrow E_{k} \propto \frac{r^{2}}{T^{2}}$ As for a satellite,  $T^{2} \propto r^{3}$   $\Rightarrow r^{2} \propto T^{\frac{4}{3}}$   $\Rightarrow E_{k} \propto \frac{T^{\frac{4}{3}}}{T^{2}} = T^{\frac{4}{3}-2}$   $\Rightarrow E_{k} \propto T^{-\frac{2}{3}}$ 

\_\_\_\_\_

# **Question 50**

If g is acceleration due to gravity on the surface of the earth, its value at the height equal to double the radius of the earth is

#### **Options:**

A. g

- B.  $\frac{g}{2}$
- a
- C.  $\frac{g}{3}$
- D.  $\frac{g}{9}$

#### Answer: D

## Solution:

**Solution:** Acceleration of gravity at height *h* above earth's surface is  $g_h = g(\frac{R}{R+h})^2$ At h = 2R,  $g_h = g(\frac{R}{R+2R})^2$  $\Rightarrow g_h = \frac{g}{9}$ 

# **Question 51**

# Correct order of electronegativity is

\_\_\_\_\_

## **Options:**

A. F > O > CI = N

B. F > CI > O > N

C. CI > F > O < N

D. None of the above

**Answer:** A

## Solution:

#### Solution:

Left to right group 1 to 17 electronegativity increase while top to bottom decreases. Fluorine in group 17 at positioned in  $II^{nd}$  period. Hence, correct order of electronegativity is F > O > CI = N

# **Question 52**

## Correct order of electronegativity is

\_\_\_\_\_

#### **Options:**

A. F > O > CI = N

B. F > CI > O > N

C. CI > F > O < N

D. None of the above

#### Answer: A

## Solution:

#### Solution:

Copper (Cu) and mercury (Hg) have positive electrode potential,  $E^{\circ} = positive$ . Hence, shows property of auto-reduction. Auto-reduction process in which sulphone ores of lesselectropositive metals like Hg, Pb, Cu etc. are heated in the air so as to convert part of the oreinto oxide or sulphate.

\_\_\_\_\_

# **Question 53**

## A de-polariser used in dry cell is

#### **Options:**

A. manganese dioxide

- B. potassium hydroxide
- C. ammonium chloride

D. zinc chloride

**Answer:** A

## Solution:

#### Solution:

The term 'de-polariser' has been used todenote a substance used in a primary cell toprevent a buid up of hydrogen gas bubbles. Abattery de-polariser takes up electrons duringdischarge of the cell, therefore it is always anoxidizing agent (Manganese dioxide) Manganesedioxide is used as de-polarizes in dry cell.  $2NH_4^+ + 2Mn_2O_2 \rightarrow Mn_2O_3 + H_2O + 2NH_3$ 

\_\_\_\_\_

# **Question 54**

## Which of following is an anionic detergent?

#### **Options:**

A. Sodium stearate

B. Sodium lauryl sulphate

- C. Cetyltrimethyl ammonium bromide
- D. Glycerol oleate

**Answer: B** 

## Solution:

#### Solution:

Sodium lauryl sulphate is an anionicdetergent. It has anion at the soluble end of thechain, it is a sodium salt of the sulphonated longchain of lauryl alcohol.

\_\_\_\_\_

# **Question 55**

# Which statement is correct for dinitrogen pentoxide?

## **Options:**

A. In the gas phase it decomposes into  $NO_2$ , NO and  $O_2$ .

- B. Solid  $N_2O_5$  is ionic.
- C. Covalent in solution and gas phase.
- D. Coloured deliquescent solid.

## Answer: B
#### Solution:

In gaseous form  $N_2O_5$  dissociate as:  $2N_2O_5 \rightarrow 4NO_2 \rightarrow O_2$  hence (a) is incorrect. In solid form  $N_2O_5$  is ionic, hence (b) is correct. There are both covalent and coordinate covalentbonds, hence (c) is incorrect. It is a colourless compound, hence (d) isincorrect. So, statement (b) is correct for dinitrogen petoxide.

-----

# **Question 56**

Which compound would give 5-keto, 2-methyl hexanal upon ozonolysis?

### **Options:**

A.





















# **Question 57**

### Anhydrone is

### **Options:**

A.  $KCIO_4$ 

B.  $NaClO_4$ 

C. Mg(ClO<sub>4</sub>)<sub>2</sub>

D.  $HCIO_4 + H_2SO_4$ 

Answer: C

### Solution:

### Solution:

Magnesium perchlorate  ${\rm Mg(CIO_4)_2}$  is dehydrating agent and known as anhydrase.

------

# **Question 58**

# In case of following Bronsted base, the correct order of basic strength is

### **Options:**

A.  $CIO_4^- > CIO_3^- > CIO_2^- > CIO_2^-$ B.  $CIO_2^- > CIO_2^- > CIO_3^- > CIO_4^-$  C.  $CIO_{3}^{-} > CIO_{2}^{-} > CIO_{4}^{-} < CIO^{-}$ 

D.  $CIO_2^- > CIO_3^- > CIO_4^- > CIO_4^-$ 

#### **Answer: B**

### Solution:

#### Solution:

According Bronsted concept of acid and base, strongest acid form weakest conjugate base.  $HCIO_4$  is strongest and HOCI is weakest acid.

 $CIO_4^- < CIO_3^- < CIO_2^- < CIO_-^-$  is the correct increasing order of Bronsted base strength with an increase in the number of oxygen atoms, the extent of delocalization of negative charge by  $p\pi - d\pi$  bonding increases.

-----

# **Question 59**

# When 10 mL of 10M solution of H<sub>2</sub>SO<sub>4</sub> and 100 mL of 1M solution of NaOH are mixed, the resulting solution will be

#### **Options:**

A. acidic

B. neutral

C. alkaline

D. Cannot be predicted

#### **Answer:** A

### Solution:

#### Solution:

One mole of  $H_2SO_4$  will give = 2 moles of H <sup>+</sup> 10 mole of  $H_2SO_4$  will give = 2 × 10 moles of H <sup>+</sup> 10 mole  $H_2SO_4$  will give = 2 × 10 × 10 moles of H <sup>+</sup> = 200 moles of H <sup>+</sup> One mole of NaOH will give = 1 mole of OH <sup>-</sup> 100 mL of NaOH will give = 1 × 100 moles of OH <sup>-</sup> In neutralisation H <sup>+</sup> will remains in solution, hence solution will be acidic.

-----

# **Question 60**

### HIO<sub>3</sub> on heating at 170°C give

#### **Options:**

A. I<sub>2</sub>

B. I<sub>2</sub>O<sub>5</sub>

C. I<sub>2</sub>O<sub>4</sub>

D. HIO<sub>4</sub>

### Answer: B

Solution:

**Solution:** On heating  $HIO_3$  at 170 °C  $2HIO_3 \rightarrow I_2O_5 + H_2O$ 

# **Question 61**

Among the following compounds that is both paramagnetic and coloured

### **Options:**

A. K<sub>2</sub>Cr<sub>2</sub>O<sub>7</sub>

B.  $(NH_4)_2[TiCl_6]$ 

C. VOSO<sub>4</sub>

D.  $K_3[Cu(CN)_4]$ 

**Answer: C** 

### Solution:

**Solution:** In VOSO<sub>4</sub> oxidation state of vanadium is +4. = [Ar] $3d^34s^2$   $V^{4+} = [Ar]3d^14s^0$ Therefore, it is paramagnetic. In K<sub>2</sub>Cr<sub>2</sub>O<sub>7</sub> oxidation state of Cr = +6, unpaired electron = 0 In (NH<sub>4</sub>)<sub>2</sub>[TiCl<sub>6</sub>] oxidation state of Ti = +4, unpaired electron = 0 In K<sub>3</sub>[Cu(CN)<sub>4</sub>] oxidation state of Cu = +1, unpaired electron = 0 In K<sub>2</sub>Cr<sub>2</sub>O<sub>7</sub>, (NH<sub>4</sub>)<sub>2</sub>[TiCl<sub>6</sub>] and K<sub>3</sub>[Cu(CN)<sub>4</sub>] in coloured but diamagnetic but VOSO<sub>4</sub> paramagnetic and coloured.

\_\_\_\_\_

\_\_\_\_\_

# **Question 62**

An element A has face centered cubic structure with edge length of **361** pm. The radius of atom A is

#### **Options:**

- B. 180.5 pm
- C. 160.5 pm
- D. 64 pm

**Answer:** A

# Solution:

**Solution:** For face centered cubic structure  $a\sqrt{2} = 4r$ a = edge length



 $r = \frac{a \times \sqrt{2}}{2}$  $a = \frac{4}{4}$  $= \frac{361 \times 1.41}{4}$  $=\frac{4}{127.6}$  pm

# **Question 64**

# IUPAC name of the complex $K_3[A|(C_2O_4)_3]$ is

### **Options:**

- A. potassium alumino oxalate
- B. potassium trioxalatoaluminate (III)
- C. potassium aluminium (III) oxalate
- D. potassium trioxalato aluminate (VI)

### Answer: B

### Solution:

**Solution:** IUPAC name of  $K_3[AI(C_2O_4)_3]$  is potassiumtrioxalatoaluminate (III).

\_\_\_\_\_

# **Question 65**

Number of electrons transferred in each case when  $KMnO_4$  acts as an oxidising agent to give  $MnO_2$ ,  $Mn^{2+}$ ,  $Mn(OH)_3$ ,  $MnO_4^{2-}$  are respectively

### **Options:**

A. 3, 5, 4, 1

B. 4, 3, 1, 5

C. 1, 3, 4, 5

D. 5, 4, 3, 1

**Answer:** A

**Solution:** 

Solution:



Hence, 3541 is correct.

\_\_\_\_\_

# **Question 66**

# In following compounds, which one contains least oxidation state of iron?

### **Options:**

A.  $K_3[Fe(OH)_6]$ 

B. K<sub>2</sub>[FeO<sub>4</sub>]

C.  $FeSO_4(NH_4)_2SO_4 \cdot 6H_2O$ 

D.  $K_3[Fe(CN)_6]$ 

Answer: C

### Solution:

### Solution:

 $\begin{array}{l} \mathsf{K}_3[\mathsf{Fe}(\mathsf{OH})_6] \text{ oxidation state of }\mathsf{Fe}=+3\\ \mathsf{K}_2[\mathsf{FeO}_4] \text{ oxidation state of }\mathsf{Fe}=+6\\ \mathsf{FeSO}_4(\mathsf{NH}_4)_2\mathsf{SO}_4\cdot\mathsf{6H}_2\mathsf{O} \text{ oxidation state of }\mathsf{Fe}=+2\\ \mathsf{K}_3[\mathsf{Fe}(\mathsf{CN})_6] \text{ oxidation state }\mathsf{Fe}=+3\\ \mathsf{Hence}, \ \mathsf{FeSO}_4(\mathsf{NH}_4)_2\mathsf{SO}_4\cdot\mathsf{6H}_2\mathsf{O} \text{ contains least oxidation state of iron.} \end{array}$ 

\_\_\_\_\_

# **Question 67**

### The set of quantum number not possible for an electron in an atom is

#### **Options:**

A.  $n = 1, l = 1, m = 1, s = +\frac{1}{2}$ B.  $n = 1, l = 0, m = 0, s = +\frac{1}{2}$ C.  $n = 1, l = 0, m = 0, s = -\frac{1}{2}$ D.  $n = 2, l = 0, m = 0, s = +\frac{1}{2}$ 

### Answer: A

### Solution:

#### Solution:

Azimuthal quantum number (/) have 0 to (n - 1) values. where, n is principal quantum number. When n = 1, / will always be 0 Hence, set a is not possible.

# **Question 68**

### In the dichromate dianion $(Cr_2O_7)^2$

\_\_\_\_\_

### **Options:**

- A. 4 Cr O bonds are equivalent
- B. 6 Cr O bonds are equivalent
- C. All Cr O bonds are equivalent
- D. All Cr O bonds are non-equivalent

#### **Answer: B**

### Solution:

Solution: In  $(Cr_2O_7)^2$  – dichromate dianion. 0-0 0-0

In  $Cr_2O_7$ , 6Cr - O bonds are equivalent and exhibit double bond character Cr - O - Cr and bond is different.

\_\_\_\_\_

# **Question 69**

### $\alpha$ – D( + ) glucose and $\beta$ -D ( + ) glucose are

#### **Options:**

A. conformers

- B. epimers
- C. anomers
- D. enantiomers

**Answer: C** 

### **Solution:**

#### Solution:

 $\alpha - D(+)$  glucose and  $\beta - D(+)$  glucose are anomers. An anomer of a saccharide only differs in its structure at the anomeric carbon. Anomeric carbon being the functional group of the carbohydrate, which is usually the carboxyl group attached to it.

-----

# **Question 70**

When  $CH_2 = CH - COOH$  is reduced with,  $LiAIH_4$  the compound obtained will be

#### **Options:**

- A. CH<sub>3</sub>CH<sub>2</sub>COOH
- B.  $CH_2 = CH CH_2 OH$
- C.  $CH_3 CH_2 CH_2 OH$
- D.  $CH_3 CH_2 CHO$

### Answer: B

### Solution:

#### Solution:

Due to reduction by  $\text{LiAlH}_4 - \text{COOH}$  group will reduce into  $-\text{CH}_2\text{OH}$  without affect carbon-carbon double bond.  $\text{CH}_2 = \text{CH} - \text{COOH} + 4\text{H} \xrightarrow{\text{LiAlH}_4} \text{CH}_2 = \text{CH} - \text{CH}_2\text{OH} + \text{H}_2\text{O}$ 

-----

# **Question 71**

### A substance forms Zwitter ion. It can have functional groups

### **Options:**

A. -NH<sub>2</sub>, -COOH

B.  $-NH_2$ ,  $-SO_3H$ 

C. Both (a) and (b)

D. None of the above

### Answer: C

### Solution:

### Solution:

Those compound which have one acidic and other basic group in their structure can form Zwitter ion. When one hydrogen change their position from acidic group to basic, Zwitter ion form. Hence,  $-NH_2$ , -COOH and  $-NH_2$ ,  $-SO_3H$  functional groups can form Zwitter ion.

\_\_\_\_\_

# **Question** 72

One mole of  $Co(NH_3)_5 Cl_3$ , gives 3 moles of ions on dissolution in water. One mole of the same complex reacts with two moles of  $AgNO_3$  solution to yield two moles of AgCl. The structure of the complex is

### **Options:**

```
A. [Co(NH<sub>3</sub>)<sub>5</sub>Cl]Cl<sub>2</sub>
```

- B.  $[Co(NH_3)_3Cl_3] \cdot 2NH_3$
- C.  $[Co(NH_3)_4Cl_2]CI \cdot NH_3$
- D.  $[Co(NH_3)_4CI]CI_2 \cdot NH_3$

### Answer: A

### Solution:

### Solution:

 $Co(NH_3) Cl_3$  complex compound gives 2 moles of AgCl by adding excess of AgNO<sub>3</sub> It exhibit only 2Cl<sup>-</sup> in present at primary valency other Cl<sup>-</sup> is present as ligand, hence molecular formula will be  $[Co(NH_3)_5Cl]Cl_2$ 

# **Question 73**

### $N_2(g) + 2H_2(g) \rightleftharpoons 2NH_3(g);$ [ $\Delta H = -93.6$ k]] The yield of ammonia does not Increase when

### **Options:**

- A. pressure is increased
- B. pressure is decreased
- C. temperature is lowered
- D. volume of reaction vessel is decreased

### Answer: B

### Solution:

#### Solution:

 $N_2(g) + 3H_2(g) \rightleftharpoons 2NH_3(g); \Delta H = -93.6 \text{ kJ}$  In above reaction, number of molecules are reducing from left to right. According to Le-Ch atelier when we create any boundation. Then reaction will shift in a direction to neutralize the effect of boundation. When we reduce the pressure in above reaction then reaction will shift into backward direction where number of molecules are increasing. Hence, the yield of ammonia does not increase when pressure is decreased.

-----

# **Question 74**

### Which pair of the following will not show common ion effect?

### **Options:**

A.  $H_2S + HNO_3$ 

B.  $H_2S + HCI$ 

C.  $NH_4 CI + NH_4 OH$ 

D. KCN + HCN

### Answer: A

### **Solution**:

#### Solution:

 $HNO_3$  is oxidizing agent, hence if will not suitable to create common ion effect. Therefore,  $H_2S + HNO_3$  combination will not exhibit common ion effect.

**Common Ion effect** When a weak electrolyte is ionise with any strong electrolyte having one ion common than ionisation of weak electrolyte in further depressed due to presence of common ion of strong electrolyte.

\_\_\_\_\_

# **Question** 75

C

### The binding energy of the hydrogen nucleus is

### **Options:**

A. 13.6 MeV

B. 200 MeV

C. 24 MeV

D. Zero MeV

**Answer:** A

### Solution:

**Solution:** Binding energy  $= mc^2 = \Delta mc^2$  $\Delta m =$  mass defect c = velocity of light For hydrogen nucleus, it 13.6 MeV. Binding energy  $= \Delta m \times 931.48$  MeV. Mass defect in hydrogen nucleus in 0.0146  $\times 931.48$  MeV = 13.6 MeV

\_\_\_\_\_

# **Question 76**

The question contains statement I and statement II of the four choices given after the statements, choose the one that best describe the two statements.

Statement I In strongly acidic solution, aniline becomes more reactive towards electrophile.

Statement II The amino group being completely protonated in strongly acidic solution, the lone pair of electrons on the nitrogen is no longer available for resonance.

### **Options:**

A. Statement I is false and statement II is true.

B. Statement I is true and statement II is true; statement II is correct explanation of statement I

C. Statement I is true and statement II is true; statement  $\mid$  is not a correct explanation of Statement I.

D. Statement I is true; statement II is false.

Answer: A

### Solution:



In strongly acidic solution aniline is not more reactive for electrophile.Due to gaining H + from acidic solution,  $NH_2$  is protonated, hence not available for resonance.Hence statement 1 is wrong and 2 is right.

\_\_\_\_\_

# **Question** 77

# Pair of species having identical shape for molecule is

### **Options:**

A.  $CF_4$ ,  $SF_4$ 

B.  $BF_3$ ,  $PCI_3$ 

C.  $XeF_2$ ,  $CO_2$ 

D.  $PF_5$ ,  $IF_5$ 

Answer: C

### Solution:

Solution: 0 = 0 = 0-Xe--F Linear Linear

------

# **Question 78**

# When mercuric iodide is added to aqueous solution of potassium iodide

### **Options:**

A. freezing point is raised

- B. freezing point is depressed
- C. freezing point remains constant
- D. freezing point may raised or depressed

Answer: A

**Solution:** By Hgl<sub>2</sub> is added in KI solution. Hgl<sub>2</sub> + 2 KI  $\rightarrow$  K<sub>2</sub>[Hgl<sub>4</sub>] the number of ions decreases, thus decreasing the van't Hoff  $\leftarrow$  factor  $\Delta T_f = i \times K_f \times m$ Since, *i* has decreased, it results in less depression in the freezing point causing the freezing point to rise.

\_\_\_\_\_

# **Question 79**

When the electrons of hydrogen atoms return to L-shell from shell of higher energy, the series of lines in the spectrum are of

### **Options:**

A. Pfund series

B. Lyman series

C. Brackett series

D. Balmer series

### Answer: D

### Solution:

### Solution:



**K L M N O P** When electron returns to n = 2 i.e. L shell than Balmer series of spectrum exhibit.

\_\_\_\_\_

# **Question 80**

Benzene reacts with  ${\sf H}_2$  at 150  $^\circ{\sf C}$  at 30  ${\sf atm}$  pressure in the presence of Ni catalyst to give

### **Options:**

A. cyclohexene

B. cyclohexane

C. *n*-hexane

D. No reaction occur

### **Answer: B**

### Solution:

#### Solution:



In pressure of Ni due to complete hydrogenation of benzene gives cyclohexane.

# **Question 81**

### The reaction



### can be classified as

### **Options:**

- A. Williamson's ether synthesis reaction
- B. alcohol formation reaction
- C. dehydration reaction
- D. Williamson alcohol synthesis reaction

### Answer: A

### Solution:

#### Solution:

Answer is Williamson's a ether synthesis reaction, where is first step -O - H group in change into  $\overline{O} - Na^+$  group changes into  $-O - CH_3^-$ .

\_\_\_\_\_

# **Question 82**

An organic compound, containing C = 39.13%, H = 2.69% and remaining % were oxygen. Its vapour density was found to be 46. What will be its molecular formula?

**Options:** 

A. C<sub>3</sub>H<sub>8</sub>O<sub>3</sub>

B.  $C_3H_6O$ 

C.  $C_6H_6O$ 

D.  $C_7H_6O_2$ 

Answer: A

Solution:

Solution: C = 39.13 H = 269 O = 100 - (39.13 + 269) = 100 - 41.82 = 58.18 % Divided in atomic mass : C =  $\frac{39.13}{12}$  H =  $\frac{2.69}{1}$  O =  $\frac{58.18}{16}$ = 3, 3, 3 Vapour density = 46 Molecular mass

 $= 2 \times 46 = 92$ Molecular formula  $= C_3H_8O_3$ = 36 + 8 + 48 = 92Compound have 3 carbons and 3 oxygens.

-----

# **Question 83**

<sup>235</sup><sub>92</sub> $\cup$  belongs to IIIrd B group of periodic table. It loses one  $\alpha$ -particle and two  $\beta$ -particles, the new element belongs to the group.

### Answer: C

#### Solution:

When radioactive elements emits  $\alpha$ -particle then it shifts 2 groups in left hand side and by emitting are  $\beta$ -particle it shift one group right. IB – IIB – IIIB Continuous release of one  $\alpha$  and two  $\beta$ -particles. It will returns to its original group, therefore new elements also belongs to III B group

\_\_\_\_\_

# **Question 84**

### Identify the gas which is adsorbed by activated charcoal.

#### **Options:**

A.  $H_2$ 

B. N<sub>2</sub>

C.  $O_{2}^{2}$ 

D. SO<sub>2</sub>

Answer: D

### Solution:

Solution:

Hydrogen, oxygen and nitrogen are not adsorb at activated charcoal. SO<sub>2</sub> will be adsorbed at activated charcoal.

\_\_\_\_\_

# **Question 85**

# If a is the initial concentration of the reactant and n be the order of the reaction, then half-life of the reaction will be

### **Options:**

A.  $a^{1-n}$ 

B.  $a^{n-1}$ 

C. *a*<sup>*n*</sup>

D.  $a^{n+1}$ 

### Answer: A

### Solution:

Solution:

# **Question 86**

**Consider following statements:** 

A. The absolute value of internal energy cannot be evaluated.

**B.** Entropy is a state function and extensive property.

C. Temperature is an extensive property.

**D.** The standard enthalpy of formation of an element ( $\Delta H_f$ ) is always considered to zero.

E. For an adiabatic process,  $\Delta S > 0$ . Wrong statements are

### **Options:**

A. A and B

B. A and D

C. C and E

D. B and D

Answer: C

Solution:

### Solution:

Temperature is an intensive property and for adiabatic process  $\Delta S$  = 0. Hence, wrong statements are C and E.

\_\_\_\_\_

# **Question 87**

2 moles of  $\mathsf{PCI}_5$  is decomposed in 2L closed vessel. At equilibrium the percentage decomposition of  $\mathsf{PCI}_5$  is 40%. The value of equilibrium constant is

# Options: A. 0.267 B. 0.0267 C. 2.67

D. 26.7

D. 20.7

Answer: A

Solution:  $\begin{array}{l} PCl_{5} \rightleftharpoons PCl_{3} + Cl_{2} \\ (a-x) \times \chi \\ \hline V & VV \\ \hline K_{C} = \frac{\chi^{2}}{(a-x)V} \\ \therefore x = 40\% \\ \therefore 2 \text{ mole} = 80\% \\ V = 2L \\ = \frac{0.8 \times 0.8}{(2-0\cdot8) \times 2} \\ = \frac{0.64}{1\cdot2\times2} = \frac{0.64}{2.4} \\ \Rightarrow K_{C} = 0\cdot267 \end{array}$ 

------

# **Question 88**

The dissociation energy of  $CH_4$  is 360 kcal mol<sup>-1</sup> and that of ethane is 620 kcal mol<sup>-1</sup>. The bond energy of C - C bond is

### **Options:**

A. 260 kcal

B. 980 kcal

C. 90 kcal

D. 80 kcal

**Answer: D** 

### Solution:

```
Solution:

In CH<sub>4</sub> there are 4C – H bond.

Bond energy = 360

Hence, bond energy of one C – H bond

= \frac{360}{4} = 90

In C<sub>2</sub>H<sub>6</sub> there are 6C – H bond and one C – C

620 = 6 × 90 + C – C

620 - 540 = C – C

Hence, C – C = 80 kcal
```

-----

# **Question 89**

Which of the following structure represents neoprene polymer?

### **Options:**

A.

$$-(-CH_2 - C = CH - CH_2 -)_n$$

В.



C.

D.



### Answer: A

### Solution:

Solution:  $-(CH_2-C=CH-CH_2)_n$  ClNeoprene polymer

\_\_\_\_\_

# **Question 90**

This question contains statement I and statement II, of the four choices given after the statements. Choose the one that best describes the two statements.

Statement I Aromatic acids do not undergo Friedel-Crafts reaction. Statement II COOH is *m*-directing group.

### **Options:**

A. Statement I is false, statement II is true.

B. Statement I is true, statement | is true, statement II is correct explanation of statement I.

C. Statement I is true, statement II is true, statement II is not a correct explanation of statement I.

D. Statement I is true, statement II is false.

#### Solution:

-COOH group is m-directing it is explained on the basis of resonance there is positive charge at o and p. Hence, electrophilic attack is not possible at o and p only m position remains for electrophilic substitution.



Aromatic acid do not undergo Friedel-Crafts reaction.

# **Question 91**

How many structural isomers are possible for  $\mathsf{C_4H_9Br}$  ?

### **Options:**

A. 2

- B. 3
- C. 4
- D. 5

### **Answer: C**

### Solution:

Solution: 4 isomers are possible for  $C_4H_9Br$ 1)  $CH_3CH_2CH_2CH_2Br$ ;  $CH_3$ —CH— $CH_2CH_3$ ; Br  $CH_3$ —CH— $CH_2Br$ ;  $CH_3$   $CH_3$   $CH_3$ — $CH_3$   $CH_3$   $CH_3$ — $CH_3$  $CH_3$ 

# **Question 92**

### The ionisation potential of which element is highest?

\_\_\_\_\_

### **Options:**

A. H

B. He

C. Ar

D.F

**Answer: B** 

### Solution:

#### Solution:

He has highest ionization potentialIP increases left to rightIP decreases top to bottom. He is present extremely right and upper positionin periodic table.

\_\_\_\_\_

# **Question 93**

 $\ensuremath{\text{pH}}$  of a solution is 10 . The number of hydrogen ions present per millilitre of solution is

**Options:** 

A.  $6.023 \times 10^{23}$ 

B. 6.023 × 10<sup>13</sup>

C.  $6.023 \times 10^{10}$ 

D.  $6.023 \times 10^7$ 

Answer: B

### Solution:

#### Solution:

pH = 10 of a solution means  $[H^+] = 1 \times 10^{-10}$  mole per litre 1 mole number of  $H^+ = 6.023 \times 10^{23}$   $10^{-10}$  mole =  $H^+ = 6.023 \times 10^{23} \times 10^{-10}$ = 6.023 × 10<sup>13</sup>

------

# **Question 94**

## Among the following which are mismatched?

	Order	Unit of k
(A)	Zero order	$mol L^{-1}$ time $^{-1}$
(B)	First order	$mol L^{-1}$
(C)	Second order	$\mathrm{Lmol}^{-1}\mathrm{time}^{-1}$
(D)	Third order	$L^2 mol^{-1} time^{-1}$
(E)	nth order	$\text{mol}^{1-n} L^{n-1} \text{time}^{-1}$

### **Options:**

A. A and  $\mathsf{C}$ 

B. B and D

 $C. \ A \ and \ E$ 

D. C and E

**Answer: B** 

### Solution:

**Solution:** Unit of  $k = mol^{(1-h)}L^{(n-1)}$  time <sup>-1</sup> For I<sup>st</sup> order reaction  $= mol^{(1-h)}L^{1-1}$  time <sup>-1</sup>  $k = time^{-1}$ Hence, B is mismatched. For III<sup>rd</sup> order reaction,  $= mol^{(1-3)}L^{3-1}$ time<sup>-1</sup>  $= mol^{-2}L^{2}$ time<sup>-1</sup> Hence, D is mismatched.

-----

# **Question 95**

### Which one of the following will act as the best protective colloid?

#### **Options:**

A. Gelatin

- B. Starch
- C. gum arabic
- D. Egg albumin

**Answer:** A

### Solution:

#### Solution:

Lower the gold number better is protecting power of hydrophilic colloid. Gelatin has minimum value of gold number. So, gelatin act as the best protective colloid.

\_\_\_\_\_

# **Question 96**

 $Na_2(g) + 3H_2(g) \rightleftharpoons 2NH_3(g)$  then  $\Delta H$  is equal to

### **Options:**

A.  $\Delta U + 2RT$ 

- B.  $\Delta U 2RT$
- C.  $\Delta U + RT$

D.  $\Delta U - RT$ 

#### Answer: B

### Solution:

Solution:  $N_2 + 3H_2 \rightleftharpoons 2NH_3$   $\therefore \Delta H = \Delta E + nRT$ where,  $\Delta H$  = Change in enthalpy  $\Delta E$  = Change in external energy  $= \Delta U + (-2)RT = \Delta U - 2RT$ 

\_\_\_\_\_

# **Question 97**

Match List-I with List-II and select the correct answer using following codes: Substance
A. Crystals of tartaric acid
B. Fe<sub>2</sub>O<sub>3</sub>
C. Rochelle salt
D. KH<sub>2</sub>PO<sub>4</sub>
E. Fe<sub>3</sub>O<sub>4</sub>
Properties
1. Ferrimagnetism
2. Ferroelectricity
3. Piezoelectricity
4. Anti-ferromagnetism
5. Pyroelectricity
Codes

### **Options:**

A. 1 2 3 4 5

B. 54321

C. 4 5 1 2 3

D. 32541

**Answer: B** 

### Solution:

#### Solution:

- A. Crystals of tartaric acid  $\rightarrow$  5. Pyroelectricity
- B.  $Fe_2O_3 \rightarrow 4$ . Anti-ferro magnetism
- C. Rochelle salt  $\rightarrow$  3. Piezoelectricity
- D.  $KH_2PO_4 \rightarrow 2$ . Ferroelectricity
- E.  $Fe_3O_4 \rightarrow 1$ . Ferrimagnetism

# **Question 98**



#### **Options:**

A. molarity

B. molality

C. normality

D. formality

**Answer: B** 

### Solution:

Solution:  $\therefore \Delta T_{f} = \frac{K_{f} \times W_{B} \times 1000}{M_{B} \times W_{A}}$ where,  $W_{B} = \text{ amount of solute in gram}$   $W_{A} = \text{ amount of solvent in gram}$   $M_{B} = \text{ molecular mass of solute}$   $\Delta T_{f} = \text{ depression in freezing point}$   $K_{f} = \text{ molar depression constant}$   $\frac{\Delta T_{f}}{K_{f}} = \frac{W_{B} \times 1000}{M_{B} \times W_{A}}$   $\frac{\Delta T_{f}}{K_{f}} = \text{ molality}$ 

# **Question 99**

In the reaction,



# the product E is

### **Options:**

### A.



## Β.



### C.



### D.



Answer: C





# **Question 100**

The process of converting hydrated alumina AI(OH)<sub>3</sub> into anhydrous alumina is called

### **Options:**

### A. roasting

B. smelting

C. dressing

D. calcination

### **Answer: D**

### Solution:

### Solution:

In the process of converting hydrated alumina  $AI(OH)_3$  into anhydrous alumina  $AI_2O_3$  is called calcination. 2  $AI(OH)_3 \xrightarrow{\Delta} AI_2O_3 + 3H_2O$ 

# **Question 101**

The angle of elevation of the top of a tower at any point on the ground is 30° and that at a point 10m nearer to tower from earlier is 60°. Then, the height of the tower is

### **Options:**

- A. 15m
- B. 10m
- C. 5√3m
- D. None of these

### Answer: C

### Solution:

#### Solution:



Let AB be the tower of height hm and C and D are the points where the angle of elevation of the top of the tower be 30° and 60°. Also, CD = 10m and let AD be xm

# Question 102

# If [.] denotes the greatest integer function, then $\int_{0}^{1.5} [x^2] dx =$

### **Options:**

A. 2 +  $\sqrt{2}$ 

B. 2 –  $\sqrt{2}$ 

C.  $1 + \sqrt{2}$ 

D. None of these

### Answer: B

### Solution:

Solution:  
Let I = 
$$\int_{0}^{1.5} [x^2] dx$$
  
=  $\int_{0}^{1} [x^2] dx + \int_{1}^{\sqrt{2}} [x^2] dx + \int_{\sqrt{2}}^{1.5} [x^2] dx$   
=  $\int_{0}^{1} 0 dx + \int_{1}^{\sqrt{2}} 1 dx + \int_{\sqrt{2}}^{1.5} 2 dx$   
=  $0 + [x]_{1}^{\sqrt{2}} + 2[x]_{\sqrt{2}}^{1.3}$   
=  $(\sqrt{2} - 1) + 2(1.5 - \sqrt{2})$   
=  $\sqrt{2} - 1 + 3 - 2\sqrt{2} = 2 - \sqrt{2}$ 

\_\_\_\_\_

# **Question 103**

Argument of the difference of two complex numbers 3 + 4i and -3 - 4i is

### **Options:**

A. 
$$\frac{\pi}{2}$$
  
B.  $\tan^{-1}\frac{4}{3}$ 

C.  $tan^{-1}\frac{3}{4}$ 

D. None of the above

### Answer: B

### Solution:

```
Solution:

Let z_1 = 3 + 4i and z_2 = -3 - 4i

\therefore z = z_1 - z_2

= (3 + 4i) - (-3 - 4i)

= 3 + 4i + 3 + 4i = 6 + 8i

Let \theta be the argument of z, then

\tan \theta = \frac{8}{6}

\therefore \theta = \tan^{-1}\frac{4}{3}
```

# **Question 104**

\_\_\_\_\_

# Length of perpendicular from the point (1, 2, 3) to the line $\frac{x-2}{4} = \frac{y-3}{5} = \frac{z-4}{6}$

#### **Options:**

A. 
$$\sqrt{\frac{6}{77}}$$
  
B.  $\frac{6}{\sqrt{77}}$ 

C. 
$$\frac{6}{77}$$

D. None of these

#### **Answer:** A

### Solution:

#### Solution:

Let P(1,2,3) and Q( $\alpha,\beta,\gamma$ ), where Q is foot of perpendicular to the line  $\frac{x-2}{4} = \frac{y-3}{5} = \frac{z-4}{6} = \lambda$  from P

$$P(1, 2, 3)$$

$$\frac{x-2}{4} = \frac{y-3}{5} = \frac{z-4}{6}$$

$$Q(\alpha, \beta, \gamma)$$
Now, coordinates of Q are  $(4\lambda + 2, 5\lambda + 3, 6\lambda + 4)$ 
 $\therefore$  DR's of line joining PQ are  
 $< 4\lambda + 2 - 1, 5\lambda + 3 - 2, 6\lambda + 4 - 3 >$   
i.e.,  $< 4\lambda + 1, 5\lambda + 1, 6\lambda + 1 >$   
Again PQ is perpendicular to the given line, so  
 $4(4\lambda + 1) + 5(5\lambda + 1) + 6(6\lambda + 1) = 0$   
 $\Rightarrow 16\lambda + 4 + 25\lambda + 5 + 36\lambda + 6 = 0$   
 $\Rightarrow 77\lambda = -15 \Rightarrow \lambda = \frac{-15}{77}$   
 $\therefore$  Coordinates of Q are  
 $(4 \times \frac{-15}{77} + 2, 5 \times \frac{-15}{77} + 3, 6 \times \frac{-15}{77} + 4)$   
i.e.,  $(\frac{94}{77}, \frac{156}{77}, \frac{218}{77})$   
Now, PQ =  $\sqrt{(\frac{94}{77} - 1)^2 + (\frac{156}{77} - 2)^2 + (\frac{218}{77} - 3)^2}$   
 $= \sqrt{(\frac{17}{77})^2 + (\frac{2}{77})^2 + (\frac{-13}{77})^2}$   
 $= \sqrt{\frac{289 + 4 + 169}{77 \times 77} = \sqrt{\frac{6}{77}}$ 

\_\_\_\_\_

# **Question 105**

Barcode reader is a/an device.

#### **Options:**

A. Output

- B. Input
- C. Display
- D. Storage
- Answer: B

Solution: Barcode reader is an input device.

\_\_\_\_\_

# **Question 106**

## The function $f(x) = x^2 - x + 1$ in the interval (0, 1) is

### **Options:**

A. monotonic decreasing

- B. monotonic increasing
- C. neither decreasing nor increasing
- D. None of the above

### Answer: C

### Solution:

```
Solution:

We have,

f(x) = x^2 - x + 1

\therefore f(x) = 2x - 1

For f(x) to be increasing

f(x) > 0

2x - 1 > 0

x > \frac{1}{2} \Rightarrow x \in (\frac{1}{2}, \infty)

and f(x) to be decreasing

f(x) < 0

2x - 1 < 0

x < \frac{1}{2} \Rightarrow x \in (-\infty, \frac{1}{2})

So, f(x) is neither increasing nor decreasing in the interval (0, 1)
```

\_\_\_\_\_

# **Question 107**

If  $\gamma_1 = 3\hat{i} + \hat{j} - 4\hat{k}$  and  $\gamma_2 = 2\hat{i} + 5\hat{j} + 8\hat{k}$  are any two vectors then the unit vector parallel to  $\gamma_2 - \gamma_1$  is

### **Options:**

A. 
$$\frac{-\hat{\mathbf{i}} + 2\hat{\mathbf{j}} + 13\hat{\mathbf{k}}}{\sqrt{174}}$$
  
B. 
$$\frac{\hat{\mathbf{i}} + 4\hat{\mathbf{j}} - 12\hat{\mathbf{k}}}{\sqrt{161}}$$
  
C. 
$$\frac{\hat{\mathbf{i}} - 2\hat{\mathbf{j}} + 13\hat{\mathbf{k}}}{\sqrt{174}}$$
  
D. 
$$\frac{-\hat{\mathbf{i}} + 4\hat{\mathbf{j}} + 12\hat{\mathbf{k}}}{\sqrt{161}}$$

**Answer: D** 

### Solution:

Solution: We have,  $\gamma_1 = 3\hat{\mathbf{i}} + \hat{\mathbf{j}} - 4\hat{\mathbf{k}} \text{ and } \gamma_2 = 2\hat{\mathbf{i}} + 5\hat{\mathbf{j}} + 8\hat{\mathbf{k}}$ Now,  $\gamma_2 - \gamma_1 = (2\hat{\mathbf{i}} + 5\hat{\mathbf{j}} + 8\hat{\mathbf{k}}) - (3\hat{\mathbf{i}} + \hat{\mathbf{j}} - 4\hat{\mathbf{k}})$   $= -\hat{\mathbf{i}} + 4\hat{\mathbf{j}} + 12\hat{\mathbf{k}}$ Again,  $|\gamma_2 - \gamma_1| = \sqrt{(-1)^2 + (4)^2 + (12)^2}$   $= \sqrt{1 + 16 + 144} = \sqrt{161}$ Unit vector parallel to  $\gamma_2 - \gamma_1 = \frac{\gamma_2 - \gamma_1}{|\gamma_2 - \gamma_1|}$  $= \frac{-\hat{\mathbf{i}} + 4\hat{\mathbf{j}} + 12\hat{\mathbf{k}}}{\sqrt{161}}$ 

# **Question 108**

For the following linear differential equation, the integrating factor will be  $x(x^2 + 1) \frac{dy}{dx} = y(1 - x^2) + x^3 \log x$ 

### **Options:**

A. 
$$\left(\frac{x^2 + 1}{x}\right)$$
  
B. 
$$\frac{x^2 - 1}{x^2}$$

C. 
$$\frac{x^2 + 1}{x^2 - 1}$$

D. None of these

### Answer: A

### Solution:

We have,  $x(x^{2}+1)\frac{dy}{dx} = y(1-x^{2}) + x^{3}\log x$   $\Rightarrow x(x^{2}+1)\frac{dy}{dx} - (1-x^{2})y = x^{3}\log x$   $\Rightarrow \frac{dy}{dx} - \frac{(1-x^{2})}{x(x^{2}+1)}y = \frac{x^{2}\log x}{x^{2}+1}$   $\therefore \text{ Integrating factor } = e^{\int \frac{-(1-x^{2})}{x(x^{2}+1)}dx}$   $= e^{\int \left(\frac{2x}{x^{2}+1} - \frac{1}{x}\right)dx}$   $= e^{(\log(x^{2}+1) - \log x)}$   $= e^{\log\left(\frac{x^{2}+1}{x}\right)}$   $= \frac{x^{2}+1}{x}$ 

\_\_\_\_\_

# **Question 109**

The parametric representation y = 4t + 1,  $x = 4 + t^2$  represents

#### **Options:**

- A. a parabola with focus at (1, 4)
- B. an ellipse with centre at (4, 1)
- C. a parabola with vertex (4, 1)
- D. None of the above

**Answer: C** 

### Solution:

Solution: We have,  $y = 4t + 1, x = 4 + t^2$   $\Rightarrow t = \frac{y-1}{4}, t^2 = x - 4$   $\therefore (\frac{y-1}{4})^2 = x - 4$   $\Rightarrow (y-1)^2 = 16(x-4)$ which is a equation of parabola with vertex (4, 1)

\_\_\_\_\_

# **Question 110**

If 
$$\sin^{-1}\frac{2x}{1+x^2} - 2\cos^{-1}\frac{1-x^2}{1+x^2} + 2\tan^{-1}\frac{2x}{1-x^2} = \frac{\pi}{2}$$
, then x is equal to

### **Options:**

- B. 1 C.  $\frac{\pi}{4}$
- D.  $\frac{\pi}{6}$

### Answer: B

### Solution:

Solution:  
We have,  

$$\sin^{-1}(\frac{2x}{1+x^2}) - 2\cos^{-1}(\frac{1-x^2}{1+x^2}) + 2\tan^{-1}(\frac{2x}{1-x^2}) = \frac{\pi}{2}$$
  
 $\Rightarrow 2\tan^{-1}x - 2(2\tan^{-1}x) + 2(2\tan^{-1}x) = \frac{\pi}{2}$   
 $\Rightarrow 2\tan^{-1}x - 4\tan^{-1}x + 4\tan^{-1}x = \frac{\pi}{2}$   
 $\Rightarrow 2\tan^{-1}x = \frac{\pi}{2}$   
 $\Rightarrow \tan^{-1}x = \frac{\pi}{4}$   
 $\Rightarrow x = \tan\frac{\pi}{4}$   
 $\therefore x = 1$ 

#### ------

# **Question 111**

A line with direction ratios 2, 7, -5 is drawn to intersect the lines  $\frac{x+3}{-3} = \frac{y-3}{2} = \frac{z-6}{4}$  and  $\frac{x-5}{3} = \frac{y-7}{-1} = \frac{z+2}{1}$ . Then, the length intercepted on it is

#### **Options:**

A. 78

B. √**78** 

C. 38

D. None of these

#### **Answer: B**

### Solution:

#### Solution:

Let  $L_1: x + 3/-3 = y - 3/2 = z - 6/4 = r_1$ and  $L_2: x - 5/3 = y - 7/-1 = z + 2/1 = r_2$ 

Again, let A and B are the points of intersection of the line with L<sub>1</sub> and L<sub>2</sub>. Then, A( $-3r_1 - 3, 2r_1 + 3, 4r_1 + 6$ ) and B( $3r_2 + 5, -r_2 + 7, r_2 - 2$ ) Now, DR's of AB are  $< 3r_2 + 5 + 3r_1 + 3, -r_2 + 7 - 2r_1 - 3, r_2 - 2 - 4r_1 - 6 >$  i.e.,  $< 3r_2 + 3r_1 + 8$ ,  $-r_2 - 2r_1 + 4$ ,  $r_2 - 4r_1 - 8 >$ But DR's of AB are < 2, 7, -5 > .  $\therefore \frac{3r_2 + 3r_1 + 8}{2} = \frac{-r_2 - 2r_1 + 4}{7} = \frac{r_2 - 4r_1 - 8}{-5}$   $\Rightarrow 7(3r_2 + 3r_1 + 8) = 2(-r_2 - 2r_1 + 4)$ and  $-5(-r_2 - 2r_1 + 4) = 7(r_2 - 4r_1 - 8)$   $\Rightarrow 23r_2 + 25r_1 = -48$ and  $-2r_2 + 38r_1 = -36$   $\therefore r_1 = -1$  and  $r_2 = -1$ So, coordinates of A are (0, 1, 2)and B are (2, 8, -3)  $\therefore$  Length of intercept, AB  $= \sqrt{(2 - 0)^2 + (8 - 1)^2 + (-3 - 2)^2}$  $= \sqrt{4 + 49 + 25} = \sqrt{78}$ 

-----

# **Question 112**

If a = 5, b = 3, c = 2 in  $\triangle ABC$ , then  $\triangle ABC$  is a

### **Options:**

A. right angled triangle

B. equilateral triangle

C. acute angled triangle

D. None of the above

#### Answer: D

### Solution:

Solution: We have, a = 5, b = 3, c = 2Now, b + c = 3 + 2 = 5 = a $\therefore$ ABC, are collinear.

\_\_\_\_\_

# **Question 113**

If the forces  $F_1 = 2\hat{i} - 4\hat{j} + 4\hat{k}$  and  $F_2 = 2\hat{i} - 3\hat{j} - 6\hat{k}$  acting together displace a particle from P(4, -2, -6) to the point Q(10, -2, 2), then the work done by the forces is

#### **Options:**
- B. 0 unit
- C. 4 units

D. None of these

#### Answer: A

### Solution:

### Solution:

We have,  $\mathbf{F}_1 = 2\hat{\mathbf{i}} - 4\hat{\mathbf{j}} + 4\hat{\mathbf{k}}, \mathbf{F}_2 = 2\hat{\mathbf{i}} - 3\hat{\mathbf{j}} - 6\hat{\mathbf{k}}$   $\therefore$  Net force,  $\mathbf{F} = \mathbf{F}_1 + \mathbf{F}_2$   $= (2\hat{\mathbf{i}} - 4\hat{\mathbf{j}} + 4\hat{\mathbf{k}}) + (2\hat{\mathbf{i}} - 3\hat{\mathbf{j}} - 6\hat{\mathbf{k}})$   $= 4\hat{\mathbf{i}} - 7\hat{\mathbf{j}} - 2\hat{\mathbf{k}}$ Again,  $\mathbf{PQ} = \mathbf{OQ} - \mathbf{OP}$   $= (10\hat{\mathbf{i}} - 2\hat{\mathbf{j}} + 2\hat{\mathbf{k}}) - (4\hat{\mathbf{i}} - 2\hat{\mathbf{j}} - 6\hat{\mathbf{k}})$   $= 6\hat{\mathbf{i}} + 0\hat{\mathbf{j}} + 8\hat{\mathbf{k}}$ Now, work done  $= \mathbf{F} \cdot \mathbf{PQ}$   $(4\hat{\mathbf{i}} - 7\hat{\mathbf{j}} + 2\hat{\mathbf{k}}) \cdot (6\hat{\mathbf{i}} - 0\hat{\mathbf{j}} + 8\hat{\mathbf{k}})$  24 + 0 - 168 Units

-----

# **Question 114**

### Which of the following device is used for gaming?

#### **Options:**

A. Mic

B. Mouse

C. Joystick

D. Light Pen

Answer: C

### Solution:

**Solution:** Joystick is used for gaming purpose.

-----

# **Question 115**

# **Solution of differential equation** dv

 $\frac{dy}{dx} = e^{x - y} (e^x - e^y)$ 

#### **Options:**

A. 
$$e^{y+x} = e^{e^x} - xe^{e^x} + c$$
  
B.  $e^y e^{e^x} = e^x e^{e^x} - e^{e^x} + c$   
C.  $e^{x-y} = e^{e^x} + x^2 e^{e^x} + c$ 

D. None of the above

#### Answer: B

### Solution:

**Solution:** We have,  $\frac{dy}{dx} = e^{x-y}(e^x - e^y)$   $\Rightarrow \frac{dy}{dx} = e^{2x-y} - e^x$   $\Rightarrow e^y \frac{dy}{dx} = e^{2x} - e^x \cdot e^y$   $\Rightarrow e^y \frac{dy}{dx} + e^x e^y = e^{2x}$ Put  $e^y = v$   $\Rightarrow e^y \frac{dy}{dx} = \frac{dv}{dx}$   $\therefore \frac{dv}{dx} + e^x v = e^{2x}$ Now, IF  $= e^{\int e^x dx} = e^{e^x}$ Hence, solution is given as  $v \cdot e^{e^x} = \int e^{2x} \cdot e^{e^x} dx + C$ Put  $e^x = t$   $\Rightarrow e^x dx = dt$   $\therefore ve^{e^x} = te^t - \int e^t \cdot 1dt + c$   $ve^{e^x} = te^t - e^t + c$ which is required solution.

\_\_\_\_\_

# **Question 116**

Out of (2n + 1) tickets numbered from 1 to 2n + 1, three are drawn at random. The probability that the numbers on them are in arithmetic progression will be

**Options:** 

A.  $\frac{3n}{5n^2 - 1}$ <br/>B.  $\frac{3n}{4n^2 - 1}$ 

C. 
$$\frac{n}{4n^2+1}$$

D. None of the above

### Solution:

#### Solution:

Out of (2n + 1) tickets, we have (n + 1) tickets having odd number written on them and n tickets have even number. For three numbers to be in AP either two number from odd or two numbers from even numbers must be selected.

three numbers to be in Green end of  $\frac{n+1}{C_2 + nC_2}$  $= \frac{\frac{(n+1)n}{2 \times 1} + \frac{n(n-1)}{2 \times 1}}{\frac{(2n+1)(2n)(2n-1)}{3 \times 2 \times 1}}$   $= \frac{3[n(n+1) + n(n-1)]}{2n(2n+1)(2n-1)}$   $= \frac{3n(n+1+n-1)}{2n(2n+1)(2n-1)}$   $= \frac{3n}{4n^2 - 1}$ 

\_\_\_\_\_

## **Question 117**

A 28m long wire is broken into two pieces. One piece is folded as square and another as circle. The lengths of two pieces will be, when the combined areas made by both figures will be minimum

A.	$\frac{112}{4+\pi'}$	$\frac{28\pi}{4+\pi}$
B.	$\frac{124}{8+\pi}$	$\frac{26\pi}{8+\pi}$
	224	14-

C.  $\frac{224}{6+\pi}$ ,  $\frac{14\pi}{6+\pi}$ 

D. None of these

#### **Answer:** A

### **Solution:**

#### Solution:

Let the length of the side of the square be x m and the radius of the circle be y metres. It is given that the length of the wire is 28m

 $\therefore 4x + 2\pi y = 28$   $\Rightarrow 2x + \pi y = 14 \dots (i)$ Let A be the combined area of the square and circle. Then,  $A = x^{2} + \pi y^{2} \dots (ii)$   $\Rightarrow A = x^{2} + \pi (\frac{14 - 2x}{\pi})^{2} \dots [\text{Using Eq}(i)]$   $= x^{2} + \frac{1}{\pi} (14 - 2x)^{2}$   $\Rightarrow \frac{dA}{dx} = 2x + \frac{2}{\pi} (14 - 2x)(-2)$   $= 2x - \frac{4}{\pi} (14 - 2x)$ and  $\frac{d^{2}A}{dx^{2}} = 2 - \frac{4}{\pi} (-2) = 2 + \frac{8}{\pi}$ 

The critical numbers of A are given by

 $\frac{dA}{dx} = 0$   $\therefore 2x - \frac{4}{\pi}(14 - 2x) = 0$   $\Rightarrow x = \frac{28}{\pi + 4}$ Clearly,  $\left(\frac{d^2A}{dx^2}\right)_{x = \frac{28}{\pi + 4}} = 2 + \frac{8}{\pi} > 0$ Thus, A is minimum when  $x = \frac{28}{\pi + 4}$ Putting  $x = \frac{28}{\pi + 4}$  in Eq.(*i*) we get  $y = \frac{14}{\pi + 4}$ So, length of the two piece of wire are  $4x = 4 \times \frac{28}{4 + \pi} = \frac{112}{4 + \pi}$ and  $2\pi y = 2\pi \times \frac{14}{\pi + 4} = \frac{28\pi}{\pi + 4}$ 

#### \_\_\_\_\_

## **Question 118**

If the ellipse  $x^2 + 9y^2 = 9$  meets the ellipse  $b^2x^2 + y^2 = b^2$ , b > 1 in *n* points, then *n* is equal to

#### **Options:**

A. 0

B. 2

C. 4

D. 9

**Answer: C** 

### Solution:

## Solution:

We have,  

$$x^2 + 9y^2 = 9$$
  
 $\Rightarrow \frac{x^2}{9} + \frac{y^2}{1} = 1$   
and  $b^2x^2 + y^2 = b^2$   
 $\Rightarrow \frac{x^2}{1} + \frac{y^2}{b^2} = 1$  and  $b > 1$ 



So, from graph it is clear both curves meet at 4points. Hence, x = 4

Which of the following natural element is the primary element in computer chips?

### **Options:**

- A. Silicon
- B. Carbon
- C. Iron
- D. Uranium
- Answer: A

### Solution:

**Solution:** Silicon is the natural elements which is used as primary element in computer chips.

#### \_\_\_\_\_

# **Question 120**

What is the name of programs that control the computer system?

A. Hardware

B. Keyboard

C. Software

D. Mouse

Answer: C

Solution:

**Solution:** Software control the computer system

\_\_\_\_\_

## **Question 121**

## When $x^{300} + 1$ is divided by $x^3 + 1$ , then remainder is

#### **Options:**

A. 2

В. **-**2

C. 3

D. None of these

#### **Answer:** A

### Solution:

#### Solution:

```
Let P(x) = x^{300} + 1 and q(x) = x^3 + 1

Now, q(x) = 0

\Rightarrow x^3 + 1 = 0

\Rightarrow x^3 = -1

Again, P(x) = x^{300} + 1

= (-1)^{100} + 1

= 1 + 1 = 2

So, when x^{300} + 1 is divided by x^3 + 1

Remainder = 2
```

# The maximum value of the function $f(x) = (\frac{1}{x})^x$ is

\_\_\_\_\_

### **Options:**

A.  $e^{e}$ B.  $(\frac{1}{e})^{e}$ 

C.  $e^{\frac{1}{e}}$ 

D. None of the above

Answer: C

## Solution:

Solution: (c) Let  $y = (\frac{1}{x})^x = x^{-x}$   $\Rightarrow \log y = -x \log x$   $\Rightarrow \frac{1}{y} \frac{dy}{dx} = -(1 + \log x)$   $\Rightarrow \frac{dy}{dx^2} = \frac{-dy}{dx}(1 + \log x) - \frac{y}{x} = y(1 + \log x)^2 - \frac{y}{x}$   $= x^{-x}(1 + \log x)^2 - \frac{x^{-x}}{x}$   $= x^{-x}(1 + \log x)^2 - x^{-x-1}$ At points of local maximum and local minimum, we must have  $\frac{dy}{dx} = 0$   $-y(1 + \log x) = 0$   $\Rightarrow 1 + \log x = 0$   $\Rightarrow \log x = -1$   $\Rightarrow x = e^{-1}$   $\Rightarrow (\frac{d^2y}{dx^2})_{x=e^{-1}} = (\frac{1}{e})^{-\frac{1}{e}}(1 + \log \frac{1}{e})^2 - (\frac{1}{e})^{-\frac{1}{e}-1}$   $= (e^{-1})^{-\frac{1}{e}}(1 - \log e)^2 - (e^{-1})^{-\frac{1}{e}-1}$   $= (e^{-1})^{-\frac{1}{e}}(1 - \log e)^2 - (e^{-1})^{-\frac{1}{e}-1}$   $= -e^{\frac{1}{e}+1} < 0$ So,  $x = \frac{1}{e}$  is a point of local maximum. The local maximum value of y is obtained by putting  $x = \frac{1}{e}$  in y and is equal to  $e^{\frac{1}{e}}$ .

# **Question 123**

The order and degree of following differential equation is  $\frac{d^2 y}{dx^2} = \left[y + \left(\frac{dy}{dx}\right)^6\right]^{1/4}$ 

#### **Options:**

- A. order 2 and degree 6
- B. order 1 and degree 6
- C. order 2 and degree 4
- D. None of the above

#### Answer: C

### Solution:

Solution: We have,  $\frac{d^2y}{dx^2} = \left[y + \left(\frac{dy}{dx}\right)^6\right]^{\frac{1}{4}}$  $\Rightarrow \left(\frac{d^2y}{dx^2}\right)^4 = y + \left(\frac{dy}{dx}\right)^6$  $\therefore \text{ Order = 2, Degree = 4}$ 

\_\_\_\_\_

# **Question 124**

The sum of all 2-digit numbers which leave remainder 3 when divided by 5 is

#### **Options:**

A. 171

B. 99

C. 999

D. 111

**Answer: C** 

### **Solution**:

#### Solution:

Required numbers are 13, 18, 23, ..., 98 These numbers from AP with  $a = 13, d = 5, a_n = 98$ Now,  $a_n = a + (n - 1) d$   $\Rightarrow 98 = 13 + (n - 1)(5)$   $\Rightarrow 85 = 5(n - 1)$   $\Rightarrow 17 = n - 1$   $\Rightarrow n = 18$ Now, required sum  $= S_n = \frac{n}{2}[a + a_n]$   $= \frac{18}{2}[13 + 98]$  $= 9 \times 111 = 999$ 

The area bounded by curves  $y = |\log_e |x||$ and y = 0 is

### **Options:**

- A. 2 sq units
- B. 4 sq units
- C. 8 sq units
- D. None of these

### Answer: A

### Solution:





2[(0-1) - (0-0)]2 sq units

# **Question 126**

If BC = a, CA = b and AB = c are three sides of a triangle with |a| = a, |b| = b and |c| = c, then  $\frac{a^2 + b^2 - c^2}{2ab}$  is

#### **Options:**

- A. cos A
- B. cos B
- C. sin C
- D. cos C

Answer: D

### Solution:

**Solution:** In a  $\triangle$  ABC, with usual notations, we know that  $\cos C = \frac{a^2 + b^2 - c^2}{2ab}$ 

------

# **Question 127**

If position vectors of A and B are 2  $\hat{i} - \hat{j} + \hat{k}$  and  $\hat{i} + \hat{j} - \hat{k}$ , then the position vector of point P dividing AB in the ratio 3:4 is

**Options:** 

A. 
$$\frac{\hat{\mathbf{i}} - 9\,\hat{\mathbf{j}} + \hat{\mathbf{k}}}{7}$$
  
B. 
$$\frac{\hat{\mathbf{i}} + \hat{\mathbf{j}} - \hat{\mathbf{k}}}{7}$$
  
C. 
$$\frac{9\,i - \hat{\hat{\mathbf{j}}} + \hat{\mathbf{k}}}{7}$$

D. None of these

**Answer: D** 

### **Solution**:

Solution:  

$$OP = \frac{3OB + 4OA}{3 + 4}$$

$$= \frac{3[\hat{i} + \hat{j} - \hat{k}] + 4[2\hat{i} - \hat{j} + \hat{k}]}{7}$$

$$\frac{3\hat{i} + 3\hat{j} - 3\hat{k} + 8\hat{i} - 4\hat{j} + 4\hat{k}}{7}$$

$$\frac{11\hat{i} - \hat{j} + \hat{k}}{7}$$

# **Question 128**

The slope of the tangent at (x, y) to a curve passing through  $(1, \frac{\pi}{4})$  is given by  $\frac{y}{x} - \cos^2(\frac{y}{x})$ , then the equation of the curve is

#### **Options:**

A.  $y = x \tan^{-1}(\log(\frac{x}{e}))$ B.  $y = x \tan^{-1}(\log(\frac{e}{x}))$ C.  $y = x^2 \tan^{-1}(\log(\frac{e}{x}))$ 

D. None of these

#### **Answer: B**

### Solution:

Solution: We have,  $\frac{dy}{dx} = \frac{y}{x} - \cos^2(\frac{y}{x}) \dots(i)$ On putting y = vx and  $\frac{dy}{dx} = v + x\frac{dv}{dx}$  in Eq.(*i*), We get,  $v + x\frac{dv}{dx} = v - \cos^2 v$   $x\frac{dv}{dx} = -\cos^2 v$   $\int \sec^2 v \, dv = \int -\frac{1}{x} \, dx$   $\tan v = -\log x + c$   $\tan \frac{y}{x} = -\log x + c$ Since, curve passes through  $(1, \frac{\pi}{4})$ , So  $\tan \frac{\pi}{4} = -\log 1 + c$   $\Rightarrow c = 1$   $\therefore \tan \frac{y}{x} = -\log x + 1$   $\Rightarrow \tan \frac{y}{x} = \log(\frac{e}{x})$  $\Rightarrow y = x \tan^{-1}(\log(\frac{e}{x}))$ 

#### -----

## **Question 129**

The area of the region in 1st quadrant bounded by the Y-axis  $y = \frac{x}{4}$ ,  $y = 1 + \sqrt{x}$  and  $y = \frac{2}{\sqrt{x}}$  is

#### **Options:**

A.  $\frac{8}{3}$  sq units

B.  $\frac{11}{3}$  sq units

C. 
$$\frac{2}{3}$$
 sq unit

D. None of these

### Answer: B

### Solution:

#### Solution:



# **Question 130**

 $\lim_{x \to 0} \frac{x}{|x| + x^2}$  is equal to

### **Options:**

A. 1

В. **-1** 

C. does not exists

D. None of the above

### Answer: C

### Solution:

### Solution:

We have, LHL =  $\lim_{x \to 0^{-}} \frac{x}{|x| + x^{2}}$ =  $\lim_{h \to 0^{-}} \frac{(0 - h)}{|0 - h| + (0 - h)^{2}}$ 

$$= \lim_{h \to 0} \frac{-h}{h+h^{2}}$$

$$= \lim_{h \to 0} \frac{-1}{1+h} = -1$$
RHL =  $\lim_{x \to 0^{+}} \frac{x}{|x|+x^{2}}$ 

$$= \lim_{h \to 0} \frac{0+h}{|0+h|+|0+h|^{2}}$$

$$= \lim_{h \to 0} \frac{h}{h+h^{2}} = \lim_{h \to 0} \frac{1}{1+h} = 1$$

$$\therefore LHL \neq RHL$$

If  $A + B = \frac{2\pi}{3}$ , where  $A, B \in R^+$ , then the maximum value of sin A + sin B is

#### **Options:**

A. √2

B. √**3** 

C.  $\frac{\sqrt{3}}{2}$ 

D. 2

### Answer: B

### Solution:

Solution: Let  $y = \sin A + \sin B$   $= \sin A + \sin (\frac{2\pi}{3} - A)$   $= \sin A + \sin \frac{2\pi}{3} \cos A - \cos \frac{2\pi}{3} \sin A$   $= \sin A + (\frac{\sqrt{3}}{2}) \cos A - (\frac{-1}{2}) \sin A$   $= \frac{3}{2} \sin A + \frac{\sqrt{3}}{2} \cos A$   $\therefore -\sqrt{(\frac{3}{2})^2 + (\frac{\sqrt{3}}{2})^2} \le \frac{3}{2} \sin A + \frac{\sqrt{3}}{2} \cos A \le \sqrt{(\frac{3}{2})^2 + (\frac{\sqrt{3}}{2})^2}$   $\Rightarrow -\sqrt{3} \le y \le \sqrt{3}$ So, maximum value of y is  $\sqrt{3}$ 

\_\_\_\_\_

# **Question 132**

..... erases letters to the left to the cursor.

### **Options:**

A. Backspace

B. Delete

C. Both

D. None of these

### Answer: A

## Solution:

Solution:

Backspace is used to erase lettes to the left of the cursor.

.....

# **Question 133**

If a line makes angles  $\alpha$ ,  $\beta$ ,  $\gamma$ ,  $\delta$  with four diagonals of a cube, then  $\sin^2 \alpha + \sin^2 \beta + \sin^2 \gamma + \sin^2 \delta$ 

### **Options:**

A.  $\frac{4}{3}$ B.  $\frac{8}{3}$ C.  $\frac{3}{4}$ D.  $\frac{3}{8}$ 

### Answer: B

## Solution:

#### Solution:

Let a be the length of an edge of the cube andlet one corner be at the origin.



Clearly, OP, AR, BS and CQ are the diagonals of the cube. The DR's of OP, AR, BS and CQ are *a* - 0, *a* - 0, *a* - 0 i.e. *a*, *a*, *a* 0 - a, a - 0, a - 0 i.e. - a, a, a *a* - 0, 0 - *a*, *a* - 0 i.e. *a*, -*a*, *a* and a - 0, a - 0, 0 - a i.e. a, a, -a respectively. Let  $\alpha$ ,  $\beta$ ,  $\gamma$ ,  $\delta$  be the angle made by a line whose DR's are *l*, *m*, *n* with OP, AR, BS and CQ. Then,  $a \cdot I + a \cdot m + a \cdot n$  $\cos \alpha = \frac{a^{2} + a^{2}}{\sqrt{a^{2} + a^{2} + a^{2} \sqrt{l^{2} + m^{2} + n^{2}}}} = \frac{l + m + n}{a^{2}}$  $\sqrt{3}\sqrt{l^2 + m^2 + n^2}$  $-a \cdot l + a \cdot m + a \cdot n$  $\cos \beta =$  $\frac{1}{\sqrt{a^2 + a^2 + a^2} \sqrt{l^2 + m^2 + n^2}}$  $\frac{\sqrt{3}\sqrt{l^2+m^2+n^2}}{al-am+an}$  $\cos \gamma = \frac{a}{\sqrt{a^2 + a^2} + a^2} \sqrt{l^2 + m^2 + n^2}$  $\sqrt{3}\sqrt{l^2 + m^2 + n^2}$ and  $\cos \delta = \frac{a_1 + a_2}{\sqrt{a^2 + a^2 + a^2}\sqrt{l^2 + m^2 + n^2}}$ al + am – an l + m - n $\sqrt{3}\sqrt{l^2+m^2+n^2}$  $= \frac{(l+m+n)^2}{3(l^2+m^2+n^2)} + \frac{(-l+m+n)^2}{3(l^2+m^2+n^2)} + \frac{(l-m+n)^2}{3(l^2+m^2+n^2)} + \frac{(l+m-n)^2}{3(l^2+m^2+n^2)}$ =  $\frac{4}{3}$ Again,  $1 - \sin^2 \alpha + 1 - \sin^2 \beta + 1 - \sin^2 \gamma + 1 - \sin^2 \delta = \frac{4}{3}$  $\Rightarrow \sin^2 \alpha + \sin^2 \beta + \sin^2 \gamma + \sin^2 \delta = 4 - \frac{4}{3} = \frac{8}{3}$ 

If  $A + B + C = \frac{\pi}{6}$ , then tan  $6A + \tan 6B + \tan 6C =$ 

### **Options:**

A.  $\cot 6A \cdot \cot 6B \cdot \cot 6C$ 

B. 0

C. 1

D. tan  $6A \cdot tan 6B \cdot tan 6C$ 

### Answer: D

### Solution:

#### Solution:

We have,  $A + B + C = \frac{\pi}{6}$   $\Rightarrow 6A + 6B + 6C = \pi$   $\Rightarrow 6A + 6B = \pi - 6C$   $\Rightarrow \tan(6A + 6B) = \tan(\pi - 6C)$   $\Rightarrow \frac{\tan 6A + \tan 6B}{1 - \tan 6A \cdot \tan 6B} = -\tan 6C$   $\Rightarrow \tan 6A + \tan 6B = -\tan 6C + \tan 6A \tan 6B \tan 6C$   $\Rightarrow \tan 6A + \tan 6B + \tan 6C = \tan 6A \tan 6B \tan 6C$ 

# **Question 135**

The function 
$$f(x) = \begin{cases} \frac{|x-3|}{x-3} & x \neq 3\\ 0 & x = 3 \end{cases}$$
. is

\_\_\_\_\_

### **Options:**

- A. discontinuous at x = -2
- B. continuous at x = -2
- C. continuous at x = 3
- D. None of the above

### Answer: B

### Solution:

Solution: We have  $f(x) = \begin{cases} \frac{|x-3|}{x-3} & x \neq 3\\ 0 & x = 3 \end{cases}$   $= \begin{cases} \frac{-(x-3)}{x-3} & x < 3\\ 0 & x = 3\\ \frac{x-3}{x-3} & x > 3 \end{cases}$   $= \begin{cases} -1 & x < 3\\ 0 & x = 3\\ 1 & x > 3 \end{cases}$ Now, at x = 3 LHL = -1, RHL = 1, f(3) = 0So, f(x) is not continuous at x = 3Again, at c = -2  $f(x) = -1[ \because \text{ when } x < 3, f(x) = -1]$ which is a polynomial function. So, f(x) is continuous at x = -2

# **Question 136**

 $\int x^{-11} (1 + x^4)^{-1/2} \, \mathrm{dx} =$ 

#### **Options:**

A. 
$$\frac{-1}{2} \left[ \frac{(1+x^{-4})^{\frac{5}{2}}}{5} - \frac{2}{3}(1+x^{-4})^{\frac{3}{2}} + (1+x^{-4}) \right] + c$$
  
B. 
$$\frac{-1}{2} \left[ \frac{(1+x^{4})^{\frac{5}{2}}}{5} - \frac{2}{3}(1+x^{4})^{\frac{3}{2}} + (1+x^{4}) \right] + c$$
  
C. 
$$\frac{1}{2} \left[ \frac{(1+x^{4})^{\frac{5}{2}}}{5} + \frac{2}{3}(1+x^{4})^{\frac{3}{2}} + (1+x^{4}) \right] + c$$

D. None of the above

#### Answer: D

### Solution:

# Solution: Let $I = \int x^{-11} (1 + x^4)^{-\frac{1}{2}} dx$ $= \int \frac{1}{x^{11} (1 + x^4)^{\frac{1}{2}}} dx$ Put $x^2 = t$ $\Rightarrow 2x dx = dt$ $\therefore I = \frac{1}{2} \int \frac{dt}{t^6 (1 + t^2)^{\frac{1}{2}}}$ Put $t = \tan u$ $\Rightarrow dt = \sec^2 u du$

$$\begin{array}{l} \therefore \ l = \frac{1}{2} \int \frac{\sec^2 u \, du}{\tan^6 u \cdot \sec u} \\ = \frac{1}{2} \int \frac{\sec u}{\tan^6 u} \, du \\ = \frac{1}{2} \int \frac{\sec^2 u}{\sin^6 u} \, du \\ = \frac{1}{2} \int \cot^5 u \csc u \, du \\ = \frac{1}{2} \int \cot^5 u \csc u \, du \\ = \frac{1}{2} \int \cot^4 u \csc u \cdot \cot u \, du \\ = \frac{1}{2} \int (\csc^2 u - 1)^2 \csc u \cot u \, du \\ \text{Put cosec } u = v \\ \Rightarrow -\csc u \cot u \, du = dv \\ \\ \therefore \ l = \frac{-1}{2} \int (v^2 - 1)^2 \, dv \\ = \frac{-1}{2} \int (v^4 - 2v^2 + 1) \, dv \\ = \frac{-1}{2} \left[ \frac{v^5}{5} - \frac{2v^3}{3} + v \right] + C \\ = \frac{-1}{2} \left[ \frac{v^5}{5} - \frac{2v^3}{3} + v \right] + C \\ = \frac{-1}{2} \left[ \frac{(t^2 + 1)^{5/2}}{5t^5} - \frac{2}{3} (t^2 + 1)^{3/2}} + \frac{(t^2 + 1)^{1/2}}{t} \right] + c \\ = \frac{-1}{2} \left[ \frac{(x^4 + 1)^{5/2}}{5x^{10}} - \frac{2}{3} \frac{(x^4 + 1)^{3/2}}{x^6} + \frac{(x^4 + 1)^{1/2}}{x^2} \right] + c \end{array}$$

If 
$$D = \begin{vmatrix} 5! & 6! & 7! \\ 6! & 7! & 8! \\ 7! & 8! & 9! \end{vmatrix}$$
, then  $\frac{D}{2(5!)^3} - 2$  is

### **Options:**

A. a multiple of 5

B. a multiple of 6

C. 0

D. **–**5

Answer: A

## Solution:

Solution: We have,  $D = \begin{vmatrix} 5! & 6! & 7! \\ 6! & 7! & 8! \\ 7! & 8! & 9! \end{vmatrix}$   $= \begin{vmatrix} 5! & 6 \times 5! & 7 \times 6 \times 5! \\ 6 \times 5! & 7 \times 6 \times 5! & 8 \times 7 \times 6 \times 5! \\ 7 \times 6 \times 5! & 8 \times 7 \times 6 \times 5! & 9 \times 8 \times 7 \times 6 \times 5! \end{vmatrix}$   $= (5!)^{3} \begin{vmatrix} 1 & 6 & 42 \\ 6 & 42 & 336 \\ 42 & 336 & 3024 \end{vmatrix}$  Common 6 from R<sub>2</sub> and 42 from R<sub>3</sub>,  $(5!)^3 \times 6 \times 42 \begin{vmatrix} 1 & 6 & 42 \\ 1 & 7 & 56 \\ 1 & 8 & 72 \end{vmatrix}$ On applying R<sub>2</sub>  $\rightarrow$  R<sub>2</sub> - R<sub>1</sub>, R<sub>3</sub>  $\rightarrow$  R<sub>3</sub> - R<sub>2</sub>, we get  $(5!)^3 \times 6 \times 42 \begin{vmatrix} 1 & 6 & 42 \\ 0 & 1 & 14 \\ 0 & 1 & 16 \end{vmatrix}$ =  $(5!)^3 \times 6 \times 42 \times 1[16 - 14]$ D =  $(5!)^3 \times 6 \times 42 \times 2$   $\Rightarrow \frac{D}{2(5!)^3} = 6 \times 42 = 252$   $\therefore \frac{D}{2(5!)^3} - 2 = 250$ which is a multiple of 5.

\_\_\_\_\_

# **Question 138**

Differential coefficient of tan<sup>-1</sup>( $\frac{2x}{1-x^2}$ ) with respect to cos<sup>-1</sup>( $\frac{1-x^2}{1+x^2}$ ) is

### **Options:**

A. 0

B. -1

C. 1

D. None of these

Answer: C

### Solution:

Solution: Let  $u = \tan^{-1} \frac{2x}{1-x^2} = 2\tan^{-1}x$ and  $v = \cos^{-1}(\frac{1-x^2}{1+x^2}) = 2\tan^{-1}x$  $\therefore u = v \Rightarrow \frac{du}{dv} = 1$ 

\_\_\_\_\_

# **Question 139**

If a line is drawn through a fixed point P(3, 4) to cut the circle  $2x^2 + 2y^2 - 32 = 0$  at *A* and *B*, then  $PA \cdot PB =$ 

### **Options:**

A. 16

B. 9

C. 41

D. 25

#### Answer: B

### Solution:



Now, length of tangent from P(3,4) = PT =  $\sqrt{(3)^2 + (4)^2 - 16} = 3$ We know that PA · PB = PT<sup>2</sup> = (3)<sup>2</sup> = 9

# **Question 140**

If  $l_1$ ,  $m_1$ ,  $n_1$  and  $l_2$ ,  $m_2$ ,  $n_2$  are direction cosines of two concurrent lines, then the direction ratios of the lines bisecting the angle between them are

#### **Options:**

- A.  $l_1 \pm m_1, l_2 \pm m_2, l_3 \pm m_3$
- B.  $m_1 \pm n_1, m_2 \pm n_2, m_3 \pm n_3$
- C.  $I_1 \pm I_2, m_1 \pm m_2, n_1 \pm n_2$
- D. *l*<sub>1</sub>*l*<sub>2</sub>, *m*<sub>1</sub>*m*<sub>2</sub>, *n*<sub>1</sub>*n*<sub>2</sub>

**Answer: C** 

### Solution:

**Solution:** Let O be the origin



Then, OA and OB are lines where DR's are  $< l_1, m_1, n_1 >$  and  $< l_2, m_2, n_2 >$  respectively. Let OA = OB = r. Again OC denotes the angle bisector of  $\angle AOB$ . Then, C will be the mid-point of AB.  $\therefore$  Coordinates of C

$$= \left(\frac{l_1r + l_2r}{2}, \frac{m_1r + m_2r}{2}, \frac{n_1r + n_2r}{2}\right)$$
  
Now, DR's of OC =  $\frac{l_1r + l_2r}{2} - 0, \frac{m_1r + m_2r}{2} - 0, \frac{n_1r + n_2r}{2} - 0$   

$$= \frac{r(l_1 + l_2)}{2}, \frac{r(m_1 + m_2)}{2}, \frac{r(n_1 + m_2)}{2}, \frac{r(n_1 + n_2)}{2}$$
  
 $\therefore$  DR's of OC =  $l_1 + l_2$ :  $m_1 + m_2$ :  $n_1 + n_2$   
Similarly, DR's of obtuse  $\angle$ AOB will be  $l_1 - l_2$ ,  $m_1 - m_2$ ,  $n_1 - n_2$   
So, DR's of angle bisector will be  $l_1 \pm l_2$ ,  $m_1 \pm m_2$ ,  $n_1 \pm n_2$ 

# **Question 141**

If 
$$\int \frac{\mathbf{1} - x^7}{x(\mathbf{1} + x^7)} dx = a \ln \left| x \right| + b \ln x^7 + \mathbf{1} \left| + c \text{ then} \right|$$

A. a = 1 and  $b = \frac{-2}{7}$ B. a = 1 and  $b = \frac{2}{7}$ C. a = -1 and  $b = \frac{-2}{7}$ 

D. None of the above

Answer: A

## Solution:

Solution:  
Let I = 
$$\int \frac{1-x^7}{x(1+x^7)} dx$$
  
=  $\int \frac{2-(1+x^7)}{x(1+x^7)} dx$   
=  $2 \int \frac{1}{x(1+x^7)} dx - \int \frac{1}{x} dx$   
=  $2 \int \frac{x^6}{x^7(1+x^7)} dx - \ln x + c_1$   
Again, let I<sub>1</sub> =  $\int \frac{x^6}{x^7(1+x^7)} dx$   
Put  $x^7 = t$  and  $7x^6 dx = dt$   
 $\therefore$  I<sub>1</sub> =  $\frac{1}{7} \int \frac{dt}{t(1+t)}$   
=  $\frac{1}{7} \int (\frac{1}{t} - \frac{1}{1+t}) dt$   
=  $\frac{1}{7} \ln t - \frac{1}{7} \ln (1+t) + c_2$   
=  $\frac{1}{7} \ln x^7 - \frac{1}{7} \ln (1+x^7) + c_2$   
=  $\ln x - \frac{1}{7} \ln (1+x^7) + c_2$   
Now, I =  $2I_1 - \ln x + c_1$   
=  $2 \ln x - \frac{2}{7} \ln (1+x^7) + c$ ,  
where,  $c = c_1 + 2c_2$   
 $\therefore a = 1, b = -\frac{2}{7}$ 

# **Question 142**

\_\_\_\_\_

The curve  $y = \sqrt{x^2 - 4}$  is defined in the interval [2, 4]. Find the coordinates of point *P*, such that the tangent at point *P* is parallel to the chord joining two points of the curve

### **Options:**

A.  $(\sqrt{2}, \sqrt{4})$ 

B. (4, 2√3)

C.  $(\sqrt{6}, \sqrt{2})$ 

D. None of these

### Answer: C

### Solution:

Solution: Let  $y = f(x) = \sqrt{x^2 - 4}$  on [2,4] According to LMVT,  $f(c) = \frac{f(b) - f(a)}{b - a}$  $= \frac{f(4) - f(2)}{4 - 2}$ 

$$= \frac{\sqrt{16 - 4} - \sqrt{4 - 4}}{2}$$
  
=  $\frac{2\sqrt{3}}{2} = \sqrt{3}$   
 $\therefore \frac{2c}{2\sqrt{c^2 - 4}} = \sqrt{3} \dots \left[ \because f(x) = \frac{2x}{2\sqrt{x^2 - 4}} \right]$   
 $\Rightarrow c = \sqrt{3} \times \sqrt{c^2 - 4}$   
 $\Rightarrow c^2 = 3(c^2 - 4)$   
 $\Rightarrow c^2 = 3c^2 - 12$   
 $\Rightarrow 2c^2 = 12$   
 $\Rightarrow c^2 = 6$   
 $\Rightarrow c = \sqrt{6} \dots \left[ \because x \in [2, 4] \right]$   
Again  $f(c) = \sqrt{6 - 4} = \sqrt{2}$ 

So, required point is  $(\sqrt{6}, \sqrt{2})$ .

\_\_\_\_\_

## **Question 143**

If  $\hat{i} - \hat{j}$ ,  $\hat{j} - \hat{k}$ ,  $\hat{k} - \hat{i}$  and  $\hat{i} + \hat{j} + \hat{k}$  are the vertices of a tetrahedron, then its volume is

**Options:** 

A.  $\frac{1}{6}$ 

B. 6

C. 1

D. None of these

#### **Answer: D**

### Solution:

Solution:

Let A, B, C, D are vertices of tetrahedron and  $\mathbf{OA} = (\hat{i} - \hat{j}), \mathbf{OB} = \hat{j} - \hat{k}, \mathbf{OC} = \hat{k} - \hat{i}$  and  $\mathbf{OD} = \hat{i} + \hat{j} + \hat{k}$ . Then,  $\mathbf{AB} = -\hat{i} + 2\hat{j} - \hat{k}, \mathbf{AC} = -2\hat{i} + \hat{j} + \hat{k}, \mathbf{AD} = 2\hat{j} + \hat{k}$   $\therefore$  Volume of tetrahedron  $= \frac{1}{6}[ABC]$  $= \frac{1}{6} \begin{vmatrix} -1 & 2 & -1 \\ -2 & 1 & 1 \\ 0 & 2 & 1 \end{vmatrix}$  $= \frac{1}{6} [-1(1-2) - 2(-2-0) - 1(-4-0)]$  $\frac{1}{6}[1+4+4]$  $=\frac{9}{6}=\frac{3}{2}$  cubic unit

## **Question 144**

The area of triangle is  $10m^2$ . If its two vertices are (0,0), (5,0) and the third vertex lie on the line x = 5, then the third vertex is

#### **Options:**

- A. (4,5)
- B. (5,4)
- C. (0,4)
- D. (5,5)

#### Answer: B

### Solution:

Solution: Let the vertices of the triangle are A(0,0), B(5,0) and C(5,y). Now, ar( $\triangle ABC$ ) = 10 sq units  $\therefore \frac{1}{2} \begin{vmatrix} 0 & 0 & 1 \\ 5 & 0 & 1 \\ 5 & y & 1 \end{vmatrix} = \pm 10$   $\Rightarrow 1[5y - 0] = \pm 20$   $\Rightarrow 5y = \pm 20$   $\Rightarrow y = \pm 4$ So, third vertex is (5, ±4).

-----

## **Question 145**

The slope of a curve, passing through (3, 4) at any point is reciprocal of twice the ordinate of that point. Then, the curve is

#### **Options:**

- A. parabola
- B. hyperbola
- C. ellipse
- D. None of the above

#### **Answer:** A

#### Solution:

```
Solution:

We have,

\frac{dy}{dx} - \frac{1}{2y}

2ydy = dx

\Rightarrow \int 2ydy - \int dx

\Rightarrow y^2 = x + c

Since, curve passes through (3,4)

\therefore (4)^2 = 3 + c

\Rightarrow c = 13

\therefore Equation of curve is :

y^2 = x + 13
```

Which is a equation of parabola.

-----

# **Question 146**

The three points (1, 2, 3), (-2, 3, 5) and (7, 0, -1)

### **Options:**

A. are collinear

B. form a right angled triangle

C. form an equilateral triangle

D. are equidistant from each other

Answer: A

## Solution:

Solution: Let A(1,2,3), B(-2,3,5) and C(7,0,-1) Then, AB = (-2-1) $\hat{i}$  + (3-2) $\hat{j}$  + (5-3) $\hat{k}$   $-3\hat{i}$  +  $\hat{j}$  = 2 $\hat{k}$ and BC = (7+2) $\hat{i}$  + (0-3) $\hat{j}$  + (-1-5) $\hat{k}$   $-9\hat{i}$  + 3j - 6 $\hat{k}$ 3(-3 $\hat{i}$  +  $\hat{j}$  + 2 $\hat{k}$ ) BC = -3AB

### \_\_\_\_\_

# **Question 147**

The product of perpendiculars drawn from any point on  $4x^2 - 9y^2 = 144$  upon its asymptotes is

### **Options:**

A. 144/3

B. 36

C. 16

D. 52

Answer: A

### Solution:

Solution: We have,

 $4x^2 - 9y^2 = 144$  $\Rightarrow \frac{x^2}{36} - \frac{y^2}{16} = 1$ which is equation of hyperbola. Now, equation of asymptotes of hyperbola  $\frac{x^2}{a^2} - \frac{y^2}{b^2} = 1 \text{ is } y = \pm \frac{b}{a}x$ ... Equation of asymptotes of given hyperbola are  $y = \pm \frac{4}{6}x$  $\Rightarrow y = \pm \frac{2}{3}x$ So, equation of asymptotes are 3y + 2x = 0 and 3y - 2x = 0Let (h, k) be any point of hyperbola. Then, Distance of (h, k) from 3y + 2x = 0 is given by  $d_1 = \left| \frac{2h + 3k}{\sqrt{4 + 0}} \right| = \left| \frac{2h + 3k}{\sqrt{2 - 0}} \right|$ and distance of (h, k) from 3y - 2x = 0 is given by  $\frac{2h+3k}{\sqrt{2}} = \left| \frac{2h+3k}{\sqrt{2}} \right|$  and distance of (h,k) from 3y - 2x = 0 is given by  $d_1 =$  $\frac{\sqrt{4+9}}{2h+3k}$  $\sqrt{13}$ d<sub>2</sub>  $\sqrt{\frac{4}{2}} \frac{9}{2h} + \frac{3}{2K}$ √13 Now,  $d_1 d_2 = \left| \frac{2h + 3k}{\sqrt{13}} \right| \times \left| \frac{-2h + 3k}{\sqrt{13}} \right|$ =  $\left| \frac{-4h^2 + 9k^2}{\sqrt{13}} \right|$ 13 -(4*h*² – 9*k*²) ...[ $\because$  (*h*, *k*) lies on hyperbola] 13  $\frac{-144}{13}$  $\frac{144}{13}$ =

# **Question 148**

If 
$$f(x) = sin |x| + cos |x|$$
, then  $\int_{-\frac{\pi}{2}}^{\frac{\pi}{2}} f(x) dx =$ 

π

#### **Options:**

A. 2

B. 4

C. 8

D. None of the above

### Answer: B

### Solution:

## Solution: $\frac{\pi}{2}$

$$I = \int_{-\frac{\pi}{2}}^{2} f(x) dx$$
  
=  $\int_{-\frac{\pi}{2}}^{\frac{\pi}{2}} (\sin |x| + \cos |x|) dx$   
=  $2 \int_{0}^{\frac{\pi}{2}} (\sin |x| + \cos |x|) dx$ 

 $\frac{\pi}{2}$  $= 2 \int_{0}^{2} (\sin x + \cos x) dx$  $= 2[-\cos x + \sin x]_0^{\pi/2}$ = 2[(0+1) - (-1+0)]= 4

In the expansion of  $(2 + 3x - 4x^3)^{2034}$  the sum of all coefficients is

\_\_\_\_\_

**Options:** 

A. 2034

B. **–**2034

C. 1

D. -1

**Answer: C** 

Solution:

Solution: Sum of all coefficient =  $(2 + 3 - 4)^{2034}$ =  $(1)^{2034} = 1$ 

\_\_\_\_\_

# **Question 150**

A bag contains 5 balls and of these it is equally likely that 0, 1, 2, 3, 4, 5 are white. A ball drawn and if found to be white. The chance that it is the only white ball, will be

**Options:** 

A.  $\frac{1}{12}$ B.  $\frac{1}{6}$ C.  $\frac{1}{15}$ D. None of these

Answer: C

### Solution:

#### Solution:

150. (c) Let  $E_1$ : Event that bag contains 0 white ball.

- $E_2$  : Event that bag contains 1 white ball.
- $E_3^{-}$ : Event that bag contains 2 white ball.
- $E_4$ : Event that bag contains 3 white ball.
- $E_5$ : Event that bag contains 4 white ball.
- $E_6$ : Event that bag contains 5 white ball.
- A : Event that drawn ball is white ball.

Now, $P(E_1)P(E_2) = P(E_3) = P(E_4) = P(E_5) = P(E_6) = \frac{1}{6}$
$P\left(\frac{A}{F_{1}}\right) = 0$
$P\left(\frac{A}{E_2}\right) = \frac{{}^{1}C_1}{{}^{5}C_1} = \frac{1}{5}$
$P\left(\frac{A}{E_3}\right) = \frac{{}^2C_1}{{}^5C_1} = \frac{2}{5}$
$P\left(\frac{A}{E_4}\right) = \frac{{}^{3}C_5}{{}^{5}C_1} = \frac{3}{5}$
$P\left(\frac{A}{E_5}\right) = \frac{{}^4C_1}{{}^5C_1} = \frac{4}{5}$
$P\left(\frac{A}{E_5}\right) = \frac{{}^5C_1}{{}^5C_1} = \frac{5}{5} = 1$
Required probability = $P(E_2/A)P(E_2)P(A/E_2)$
$= \frac{P(E_1)P(A/E_1) + P(E_2)P(A/E_2) + P(E_3)P(A/E_3)}{P(E_4)P(A/E_4) + P(E_5)P(A/E_5) + P(E_6)P(A/E_6)}$
$=\frac{\frac{1}{6}\times\frac{1}{5}}{1}$
$\frac{1}{6} \times 0 + \frac{1}{6} \times \frac{1}{5} + \frac{1}{6} \times \frac{2}{5} + \frac{1}{6} \times \frac{3}{5} + \frac{1}{6} \times \frac{4}{5} + \frac{1}{6} \times 1$
$=\frac{1}{1+2+3+4+5}=\frac{1}{15}$

**Question 63** 

## Among the following which are mismatched?

\_\_\_\_\_

Molecule Hybridisation

A. $CIO_2^-$	sp <sup>3</sup>
B. NH₃	sp <sup>2</sup>
C. SF <sub>6</sub>	sp³d²
D. SF <sub>4</sub>	sp <sup>3</sup> d
E. XeF <sub>4</sub>	sp

### **Options:**

A. A and C

B. B and E

 $C.\ C \ and \ E$ 

D. A and D

Answer: B

## Solution:



H = H HXeF<sub>4</sub> is  $sp^3d^2$ -hybrid having 2 lone pairs.



Hence, B and E are mismatched.

\_\_\_\_\_