# Carboxylic acids and Their derivatives

## Self Evaluation Test -28

1. Which of the following is the weakest acid[CPMT 2001]

(a) OH

(b)  $CH_3COOH$ COOH

(c) HCOOH



2. Pyruvic acid is obtained by

[AFMC 1995]

- (a) Oxidation of acetaldehyde cyanohydrin
- (b) Oxidation of acetone cyanohydrin
- (c) Oxidation of formaldehyde cyanohydrin
- (d) None of these
- **3.** The product obtained by dry distillation of calcium formate on reacting with ammonia yields
  - (a) Formamide
- (b) Acetamide
- (c) Acetaldehyde ammonia (d)

Urotropine

4. In the reaction

$$C_8H_6O_4 \xrightarrow{\Delta} X \xrightarrow{NH_3}$$

The compound X is

[Roorkee Qualifying 1998]

- (a) Phthalic anhydride
- (b) Phthalic acid
- (c) o-xylene
- (d) Benzoic acid
- 5. Ethyl acetate reacts with  $CH_3MgBr$  to form

[MP PET 1999]

- (a) Secondary alcohol
- (b) Tertiary alcohol
- (c) Primary alcohol and acid (d)

Acid

- 6. In quick vinegar process of acetic acid, the temperature of mixture is [RPMT 2003]
  - (a) 300 K
- (b) 427 K
- (c) 500 K
- (d) 350 K
- 7. Formic acid can reduce
- [CPMT 1987]
- (a) Tollen's reagent
- (b) Mercuric chloride
- (c)  $KMnO_4$
- (d) All of these
- **8.** 0.2 *gm* of fine animal charcoal is mixed with half litre of acetic acid (-SM) solution and shaken for 30 minutes

[BHU 1998]

- (a) The concentration of the solution decreases
- (b) Concentration increases

- (c) Concentration remains same
- (d) None of these
- 9. Dimerisation in carboxylic acid is due to [KCET 2002]
  - (a) Ionic bond
  - (b) Covalent bond
  - (c) Coordinate bond
  - (d) Intermolcular hydrogen bond
- **10.** A colourless organic compound gives brisk effervescences with a mixture of sodium nitrite and dil. *HCl.* It could be

[CPMT 1978]

- (a) Glucose
- (b) Oxalic acid
- (c) Urea
- (d) Benzoic acid
- 11. What is formed when benzoyl chloride reacts with aniline in presence of sodium hydroxide[BHU 1996]
  - (a) Acetanilide
- (b) Benzanilide
- (c) Benzoic acid
- (d) Azobenzene
- 12. Strong acid among the following is

[CBSE PMT 1992; AFMC 1998; BHU 2000]

- (a)  $CF_3COOH$
- (b)  $CBr_3COOH$
- (c) CH<sub>3</sub>COOH
- (d) CCl<sub>3</sub>COOH
- 13. Aspirin is obtained by the reaction of salicylic acid with

[AFMC 1998]

- (a) Acetone
- (b) Acetaldehyde
- (c) Acetyl chloride
- (d) Acetic anhydride
- 14. Oxalic acid when reduced with zinc and  $H_2 SO_4$  gives

[Tamil Nadu CET 2001]

- (a) Glyoxallic acid
- (b) Glyoxal
- (c) Glycollic acid
- (d) Glycol
- **15.** A distinctive and characteristic functional group in fats is

[NCERT 1981; MP PET 1995]

- (a) A ketonic group
- (b) An ester group
- (c) A peptide group
- (d) An alcoholic group

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Which substance will give amide when heated 16. with  $NH_3$ 

17. Which acid has least  $pK_a$  value [CPMT 1982]

(a)  $Cl_3C.COOH$ [CPMT 1997]

- (b) Cl<sub>2</sub>CH.COOH
- (c) Cl.CH<sub>2</sub>COOH
- (d) CH<sub>3</sub>COOH

(c) Ethane

(a) Potassium

(b) Hydrogen (d) Nitrogen

### Answers and Solutions

(SET -28)

- (a) Phenol is a weaker acid than carboxylic acids. 1.
- (a)  $CH_3 C H + HCN \rightarrow CH_3 C H \xrightarrow{H_2O}$ 2.  $CH_3 - CH - COOH \xrightarrow{[O]} CH_3 - C - COOH$ Pyruvic acid
- (d)  $2(HCOO)_2Ca \xrightarrow{\text{Dry distillation}} 2HCHO + 2CaCO_3$ 3.  $6HCHO + 4NH_3 \rightarrow (CH_2)_6N_4 + 6H_2O$
- 0 CONH<sub>2</sub> O-MgBr
- (b)  $CH_3 C O C_2H_5 + CH_3MgBr \rightarrow CH_3 C O C_2H_5$

$$\begin{array}{c} CH_3 \\ \xrightarrow{\text{Excess of}} CH_3 - \overset{C}{C} - OMgBr \xrightarrow{H_2O} CH_3 - \overset{C}{C} - OH \\ CH_3 MgBr & CH_3 \end{array}$$

- (a)  $CH_3CH_2OH + O_2 \xrightarrow{Acetobacter} CH_3COOH + H_2O$ 6.
- (d) Tollen's 7. reagent  $HCOOH + Ag_2O \rightarrow CO_2 + H_2O + 2Ag$

Fehling solution -  

$$HCOOH + 2CuO \rightarrow CO_2 + H_2O + Cu_2O$$
(Red ppt)

Mercury chloride – 
$$2HCOOH + 2HgCl_2 \rightarrow 2CO_2 + 4HCl + 2Hg$$
 (Black)

- 8. (a) Activated charcoal adsorbed the impurity of acetic acid by which the concentration of acetic acid solution decrease.
- (d) Intermolecular hydrogen bonding leads to 9. dimerisation of carboxylic acid in nonaqueous solvents.
- (c)  $NaNO_2 + HCl \rightarrow HNO_2 + NaCl$ 10.  $H_2NCONH_2 + HNO_2 \rightarrow CO_2 + NH_3 + H_2O + N_2$

CO<sub>2</sub> evolve with brisk effervescence.

(b) It is known as Schotten Baumann reaction. 11.

$$C_6H_5NH_2 + ClCOC_6H_5 \xrightarrow{NaOH} C_6H_5NHCOC_6H_5 + HCl \\ \text{Aniline} \qquad \text{Benzoyl chloride}$$

(a) Due to -I effect of three F atom  $CF_3COOH$  is 12. a strong acid.

OH 
$$COOH$$
  $COOH$   $COOH$ 

14. (c) 
$$COOH \atop COOH \atop COOH \atop Oxalic acid} + 4[H] \xrightarrow{Z_1} CH_2OH \atop H_2SO_4} + H_2O$$

- (b) Fat is the ester of higher acids & glycerol. 15.
- (a)  $2K + 2NH_3 \rightarrow 2KNH_2 + H_2$
- (a)  $(Cl_3C-COOH)$  Trichloroacetic acid has least 17. pka value and is most acidic.

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