Sl. No.

903915

N-11(E)

(MARCH, 2012)

પ્રશ્ન પેપરનો સેટ નંબર Set No. of Question Paper

9

PART - A

Time: 75 minutes]

[Maximum Marks: 50

W	-1			
In	STTI	I Kan A I	ons	-

- (1) There are 50 objective type questions in this part and all are compulsory.
- (2) The questions are serially numbered from 1 to 50 and each carries 1 mark.
- (3) You are supplied with separate OMR sheet with the alternatives (A) ○, (B)○, (C) ○, (D) against each question number. For each question, select the correct alternative and darken the circle as completely with the pen against the alphabet corresponding to that alternative in the given OMR sheet.
- From the following 1 to 50 questions, select the correct alternative from the given four answers and darken the circle with pen against the alphabet, against the number in OMR sheet.
- Each question carries 1 mark.
- 1. Who invented the simple battery first?
 - (A) Faraday

(B) Ohm

(C) Volta

- (D) Alva Edison
- 2. Which of the following oxide is not a neutral oxide?
 - (A) CO

(B) N₂O

(C) H₂O

- (D) SO_2
- 3. What is IUPAC name of Acetone?
 - (A) Propanal

(B) Propanone

(C) Propanol

- (D) Propanoic acid
- 4. What is the diameter of nano-shells which are attached only to Cancerous cells?
 - (A) 400 nm

(B) 200 nm

(C) 100 nm

(D) 50 nm addust language

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5.	Which metal causes Minamata disease?					
	(A)	Copper	(B)	Lead		
	(C)	Manganese	(D)	Mercury		
6.	Wha	at is the chemical formula of S	ilver gl	lance ?		
	(A)	AgCl	(B)	Ag ₂ S		
	(C)	SiO_2	(D)	AgNO ₃		
		mos sue lla bas frag sirl au d				
		ch organic compounds contain				
eli i	(A)	Carboxylic acid	(B)	H functional group ? Ketone		
	(C)	Aldehyde	(D)	Alcohol		
		s, soloot the correct alteriories				
8.	Wha	at is the atomic number of Tra	nsurar	nic elements?		
	(A)	Z = 92	(B)	Z < 92		
	(C)	Z > 92	(D)	$Z \leq 92$		
		mdO Si		Who invented the simple butte		
9.	Give	e the name of theory proposed	by Err			
	(A)	Theory of germplasm	(B)	Theory of mutation		
	(C)	Theory of recapitulation	(D)	Theory of natural selection		
			Į.			
10.		ich space shuttle met with an ch Indian Astronaut Kalpana		a died in February 2003?		
	(A)	Colombia	(B)	Challenger		
	(C)	Discovery	(D)	PSLV		
11.	Whi	ich compound is condensation	polyme			
	(A)	Nylon	(B)	PVC		
	(C)	Natural Rubber	(D)	Teflon 001 (0)		
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12.		t will be produced in a reducti		action of Methanal with H ₂
	gas i	in the presence of Pd catalyst? CH ₃ OH	(B)	C ₂ H ₅ OH
	(C)	C ₃ H ₇ OH	(D)	$C_4H_{10}OH$
-	-		AT .	II of III and to movide
13.		ch structure is developed in rition to foetus?	the v	
	(A)	Amnion	(B)	Fallopian tube
	(C)	Umbilical cord	(D)	Placenta
14.	Pon	s connects which two organs wees?	ith the	e help of transverse band of
	(A)	Nerves of brain and spinal co	rd.	A STATE OF THE PARTY OF THE PAR
	(B)	Both the cerebral hemisphere	es.	The state of the s
	(C)	Cerebellum and Central nerv	ous sy	stem. Junique and the state of
	(D)	Sympathetic and Parasympat	thetic.	
15.		ich of the following hormones is lants ?	s řespo	nsible for shedding of leaves
	(A)	Abscisic acid	(B)	Gibberellin
	(C)	Cytokinin	(D)	Auxin
16.	Wh	ich metal of the following meta	ıls is n	nore active ?
	(A)	Mg	(B)	Zn
	(C)	Ca	(D)	Al this design and all al
17.	Inv	which plant, vegetative propaga	ation b	y leaf takes place?
	(A)	Sweet potato	(B)	Potato
	(C)	Phalsa	(D)	Bryophyllum
18.	Wh	at is the percentage of Carbon		
	(A)	0.1 to 0.4	(B)	1.5 to 2.5
7- 4	(C)	2.5 to 3.5	(D)	0.5 to 1.5
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		is produced?	anylah	a Marak Ita ain an asaki as
	(A)	In a direction of current.	Tx	
	(B)	In the opposite direction of cu	rrent.	
	(C)	Circular around the conductor	r	
15	(D)	Perpendicular to the direction	of cu	rent.
		oda agustri (181)		
20.		ch organism normally shows as mentation ?	sexual	reproduction by
	(A)	Oscillatoria	(B)	Amoeba
	(C)	Paramoecium	(D)	Penicillium
21.	Whi	ch material provides the mecha	anical	
	(A)	Prothrombin	(B)	Cellulose
	(C)	Lignin	(D)	Pectin
22.	How	many Light-year away, the Su	n is loc	eated from the galactic centre?
	(A)	250	(B)	30,000
	(C)	22.5	(D)	15,000
23.	In w	which material, Sulphur is solul	ole?	
	(A)	Carbon disulphide	(B)	Bromine
	(C)	Heavy water	(D)	Distilled water
		(L) Brynnbyllum		
24.	Give	e the chemical formula of Calci	um su	lphate hemihydrate.
	(A)	${\rm CaSO_4 \cdot 2H_2O}$	(B)	$CaSO_4 \cdot \frac{1}{2}H_2O$
	(C)	$CaSO_4 \cdot 10H_2O$	(D)	${\rm CaSO_4 \cdot H_2O}$
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*				
			PAGE 1	

25.	Whi	ch pair of the following is comp	olemen	atary colours?	
	(A)	Blue and yellow			
	(C)	Red and magenta	(D)	Blue and magenta	
		yla und Arrhanius le ve (D		No and Second by Co.	
26.	Wha	at is the formula for Electric Po	wer?	Which satellite is launched b	.88
	(A)	$P = I^{2}Rt$ $P = VI \times t$	(B)	$P = \frac{W}{t}$	
	(C)	$P = VI \times t$	(D)	P = VQ	
		subyet is sure sure sure (d)		The state of the s	
27.		at is the name of finger like pro	ojectio	ns in the small intestine of	
	(A)	Vermiform Appendix	(B)	Villi sobset reblat (0)	
	(C)	Gizzard	(D)	Food vacuole	
		the of the exercise of the Es	or qu		
28.	Who	developed the treatment tech te water?		for industrial and sewage	
	(A)	NEERI (II)	(B)	NACO	
	(C)	WHO	(D)	ISRO	
	1.8				
29.	Whi	ch metal oxide is used to obtain	n blue	coloured glass?	
	(A)	Ferric oxide	(B)	Chromium oxide	
	(C)	Manganese oxide	(D)	Cobalt oxide	
		"sanoqa			
30.	Whi	ch mineral is necessary for bloo	od clot	ting?	
	(A)	Calcium mullydgogill (G)	(B)	Magnesium	
	(C)	Phosphorus	(D)	Iron	
				Which chromosome has satel	
31.	By v	which, hormones are circulated	?		
	(A)	Water and reporter due 100	(B)	Nerve	
	(C)	Blood	(D)	Cytoplasm	
		*			
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32.	Whi	ch catalyst is used for the indu	strial	production of Hydrogen?
	(A)	Iron (Iron)	(B)	Nickel Mark William 188
	(C)	Vanadium pentoxide	(D)	Palladium has had
33.	Whi	ch satellite is launched by Indi	a for I	Direct to Home (DTH)?
	(A)	INSAT-4A	(B)	IRS-1
	(C)	METSAT	(D)	CARTOSAT
34.	Whi	ch compound is used for bleach	ning cl	oths in laundry ?
	(A)	Bleaching Powder	(B)	Washing Powder
	(C)	Baking Powder	(D)	Plaster of Paris
35.	Whi	ch planet has atmosphere up to	1% of	the atmosphere of the Earth?
	(A)	Mars	(B)	Venus
	(C)	Jupiter	(D)	Saturn
36.	Give	e the unit of rate of reaction.		
	(A)	Molar mola summer de summe	(B)	Second
	(C)	Second/Molar	(D)	Molar/Second
				obico osamonuiM. (C)
37.	Whi	ch plant shows thigmonastic re		
	(A)			Mimosa
	(C)	Periwinkle Periwinkle	(D)	
38.		ch chromosome has satellite?		
	(A)	Telocentric	(B)	Metacentric
	(C)	Acrocentric	(D)	Sub-metacentric

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39.	Whi	ch scientist gave the law of Act	tive ma	ass? Managro augolomo P	
	(A)	Goldberg and Waag.	(B)	Lowry and Bronsted.	
	(C)	Boyle and Arrhenius.	(D)	Lewis and Sorensen.	
				(D) Structurally hamographer	
40.	Who	gave the principle of Electron			
	(A)	Volta	(B)	Ampere	
	(C)	Faraday	(D)	Oersted	
				method (0)	
41.	Whi	ch elements are present in the	alloy	of Solder?	
	(A)	Copper and Zinc	(B)	Copper and Tin	
	(C)	Nickel and Chromium	(D)	Lead and Tin	
				ne foliament massimm 80.0 (A)	
42.	In s	pectrum obtained with prism,	which		
	(A)	Red	(B)	Yellow	
	(C)	Violet		Blue and analy daldw 1A	
				fai-eightedness (hypermetropi	
43.	If 0.	3 A current passes through a	lamp,	how many electrons will pass	
	in 6	0 seconds? $(e = 1.6 \times 10^{-19})$		anisar to morning (3)	
	(A)	2.88×10^{20}	(B)	1.125×10^{20}	
	(C)	2.25×10^{20}	(D)	1.8×10^{20}	
44.				ric energy into a mechanical	
		rov?		Alovinishedophi, we got deness	
	(A)	Electric generator	(R)		
	(C)	Electric motor	(D)		
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45. Which organs perform the same function but structurally different?

	(A)	Homologus organs.		
	(B)	Analogus organs.		
	(C)	Vestigial organs.		
	(D)	Structurally homogeneous	s organs.	
		stic induction 2 V 2.1 -		Who gave the principle of The
46.		much the temperature operature of the body?	f Scrotun	n in male is lower than the
	(A)	0° C betareO	(B)	3° C
	(C)	34° C	(D)	37° C
				White departs are project to
47.	of g		ctively 1.5	lycerine. If refractive indexes and 1.47, find the refractive
	(A)	0.03	(B)	1.02
	(C)	2.20 The last web ar molecule	(D)	0.98
				(A) Rod
48.	A 4			
40.		which place in eye, ims sightedness (hypermetropia		ormed of a person having
40.		sightedness (hypermetropia On retina	(B)	ormed of a person having Behind retina
40.	far-	sightedness (hypermetropia	(B)	9 7
40.	far-s (A)	Sightedness (hypermetropia On retina	(B)	Behind retina
49.	far-s (A) (C)	Sightedness (hypermetropia On retina	(B) (D)	Behind retina On lens of eye.
	far-s (A) (C)	on retina Infront of retina	(B) (D)	Behind retina On lens of eye.
	far-s (A) (C)	on retina Infront of retina ponges, which structure is	(B) (D) used for e	Behind retina On lens of eye. excretion?
	far-s (A) (C) In S (A) (C)	On retina Infront of retina ponges, which structure is Contractile vacuole Nephridia	(B) (D) used for e (B) (D)	Behind retina On lens of eye. excretion? Flame cells
	far-s (A) (C) In S (A) (C)	On retina Infront of retina ponges, which structure is Contractile vacuole Nephridia	(B) (D) used for e (B) (D)	Behind retina On lens of eye. excretion? Flame cells Osculum
49.	far-s (A) (C) In S (A) (C)	On retina Infront of retina ponges, which structure is Contractile vacuole Nephridia	(B) (D) used for e (B) (D)	Behind retina On lens of eye. excretion? Flame cells Osculum

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PART - B

Time: 2.00 Hours] [Maximum Marks: 50 Instructions :-There are total **four** sections in this part. All questions are compulsory. (iii) Draw neat labelled diagram as per instructions. (iv) There are internal options in some questions. Pay attention to them. Figures to the right indicate marks. (v) **SECTION - A** Give short answer (in limit of 30 words) of the following questions. 1. At which temperature range; Petrol, Diesel, Kerosene and lubricating oil is obtained in fractional distillation of Petroleum? What is Molarity? Give its unit. 2. What is slow and fast reaction? Give example. 2. Why Carbon is important in development of Nano-technology? 3. Give the definition of Solar constant and its value. 4. 2 OR Write the name of various types of Coal and give the percentage of Carbon in each. Give short information about Mercury.

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SECTION - B

Write short answer (in the limit of 30 words) of the following questions.

6.	Calculate the pH of 0.04 M aqueous solution of NaOH.	
	$(\log_{10} 4 = 0.6021)$	2
7.	Write uses of Baking powder (NaHCO ₃).	2
8.	How pure water can be obtained by sewage treatment?	2
9.	Write the process of making Ethyl acetate by esterification with equation. OR	2
9.	Write difference between Soap and Detergent.	
10.	Give short information regarding Spinal cord of human.	2
	SECTION - C	
Writ	e answers in short (in the limit of 50 words) of the following questions.	
11.	Explain the work of an Electric Generator with diagram.	3
12.	Explain allotropes of Sulphur. OR	3
12.	Complete the following chemical reactions:	
	(i) $S_{(s)} + 2H_2SO_{4(aq)} \rightarrow$	
	(ii) $SO_{2(g)} + 2H_2S_{(g)} \rightarrow$	
1	(iii) $H_2S_2O_{7(l)} + H_2O_{(l)} \rightarrow$	
13.	Describe Erythrocyte (RBC) in short.	3
14.	Explain Sex determination.	3
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15. Explain Electroplating with suitable example.

3

OR

15. Explain series connections of Resistors and derive the formula of equivalent resistance.

SECTION - D

Write the answer of the following questions in detail (in the limit of 100 words).

16. Derive lens formula $\frac{1}{v} - \frac{1}{u} = \frac{1}{f}$

5

5

17. What is concentration or enrichment of Ore? Explain the method of enrichment of ore containing Sulphide with diagram.

- 17. Explain Bayer's method for obtaining Alumina from Bauxite with equations.
- 18. Write short note: Aerobic respiration and Anaerobic respiration.

5

OR

OR

18. What is Nutrition? Describe nutrition in Amoeba. (Draw diagram).

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(Space for Rough Work)

15. Explain series connections of Resistors and derive the formula of equivalent resistance to not a compact to but to 11q and and a contract.

Ste flow of Buline powder DarkCo.

. Write the answer of the following questions in detail in the limit of 100 words).

Weite the precious making lithra estate to the control with earning of the control of the contro

of surrehment of are containing Sulphide with diagram

R

17. Explain Bivor's motived for obtaining Alumna from Bauxits with equations.

18. Write short note: Accobic respiration and Ansemble respiration.

What is Nutrition Describe autrition in Amosba (Draw diagram).

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