

# KISHORE VAIGYANIK PROTSAHAN YOJANA - 2009

Date : 01-11-2009

Duration : 3 Hours

Max. Marks : 100

## STREAM - SA

### GENERAL INSTRUCTIONS

- The Test Booklet consists of **80** questions.
- There are Two parts in the question paper. The distribution of marks subjectwise in each part is as under for each correct response.

### MARKING SCHEME :

#### PART-I :

##### MATHEMATICS

Question No. **1 to 15** consist of **ONE (1)** mark for each correct response.

##### PHYSICS

Question No. **16 to 30** consist of **ONE (1)** mark for each correct response.

##### CHEMISTRY

Question No. **31 to 45** consist of **ONE (1)** mark for each correct response.

##### BIOLOGY

Question No. **46 to 60** consist of **ONE (1)** mark for each correct response.

#### PART-II :

##### MATHEMATICS

Question No. **61 to 65** consist of **TWO (2)** marks for each correct response.

##### PHYSICS

Question No. **66 to 70** consist of **TWO (2)** marks for each correct response.

##### CHEMISTRY

Question No. **71 to 75** consist of **TWO (2)** marks for each correct response.

##### BIOLOGY

Question No. **76 to 80** consist of **TWO (2)** marks for each correct response.

**PART-I (1 Mark)**  
**MATHEMATICS**

1. The real numbers  $x$  satisfying

$$\frac{\sqrt{x+5}}{1-x} > 1$$

are precisely those which satisfy

- A.  $x < 1$    B.  $0 < x < 1$    C.  $-5 < x < 1$    D.  $-1 < x < 1$
2. Let  $t_n$  denote the number of integral-sided triangles with distinct sides chosen from  $\{1, 2, 3, \dots, n\}$ . Then  $t_{20} - t_{19}$  equals  
A. 81   B. 153   C. 163   D. 173
3. The number of pairs of reals  $(x, y)$  such that  $x = x^2 + y^2$  and  $y = 2xy$  is  
A. 4   B. 3   C. 2   D. 1
4. How many positive real numbers  $x$  satisfy the equation

$$x^3 - 3|x| + 2 = 0.?$$

- A. 1   B. 3   C. 4   D. 6
5. Let  $(1 + 2x)^{20} = a_0 + a_1x + a_2x^2 + \dots + a_{20}x^{20}$ . Then

$$3a_0 + 2a_1 + 3a_2 + 2a_3 + 3a_4 + 2a_5 + \dots + 2a_{19} + 3a_{20}$$

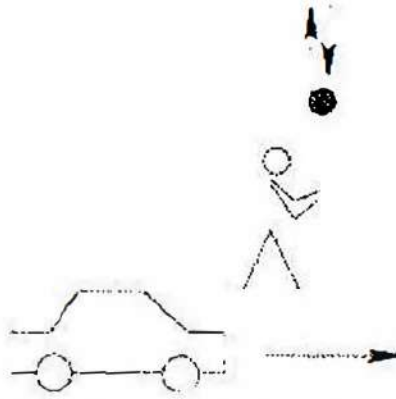
equals

- A.  $\frac{5 \cdot 3^{20} - 3}{2}$    B.  $\frac{5 \cdot 3^{20} + 3}{2}$    C.  $\frac{5 \cdot 3^{20} + 1}{2}$    D.  $\frac{5 \cdot 3^{20} - 1}{2}$
6. Let  $P_1, P_2, P_3, P_4, P_5$  be five equally spaced points on the circumference of a circle of radius 1, centred at  $O$ . Let  $R$  be the set of points in the plane of the circle that are closer to  $O$  than any of  $P_1, P_2, P_3, P_4, P_5$ . Then  $R$  is a  
A. circular region   C. pentagonal region  
B. rectangular region   D. oval region that is not circular
7. A company situated at  $(2, 0)$  in the  $xy$ -plane charges Rs. 2 per km for delivery. A second company at  $(0, 3)$  charges Rs. 3 per km for delivery. The region of the plane where it is cheaper to use the first company is  
A. the inside of the circle  $(x + 5.4)^2 + y^2 = 18.72$   
B. the outside of the circle  $(x + 1.6)^2 + (y - 5.4)^2 = 18.72$   
C. the inside of the circle  $(x - 1.6)^2 + (y + 5.4)^2 = 18.72$   
D. the outside of the circle  $(x - 5.4)^2 + (y + 1.6)^2 = 18.72$

8. In a right triangle  $ABC$ , the incircle touches the hypotenuse  $AC$  at  $D$ . If  $AD = 10$  and  $DC = 3$ , the inradius of  $ABC$  is  
A. 5   B. 4   C. 3   D. 2
9. The sides of a quadrilateral are all positive integers and three of them are 5, 10, 20. How many possible values are there for the fourth side?  
A. 29   B. 31   C. 32   D. 34
10. If the volume of a sphere increases by 72.8%, then its surface area increases by  
A. 20%   B. 44%   C. 24.3%   D. 48.6%
11. If the decimal  $0.d25d25d25\dots$  is expressible in the form  $n/27$ , then  $d + n$  must be  
A. 9   B. 28   C. 30   D. 34
12. At what time between 10 O'clock and 11 O'clock are the two hands of a clock symmetric with respect to the vertical line (give the answer to the nearest second)?  
A. 10h 9m 13s   B. 10h 9m 14s   C. 10h 9m 22s   D. 10h 9m 50s
13. A woman has 10 keys out of which only one opens a lock. She tries the keys one after the another (keeping aside the failed ones) till she succeeds in opening the lock. What is the chance that it is the seventh key that works?  
A.  $\frac{7}{10}$    B.  $\frac{1}{2}$    C.  $\frac{3}{10}$    D.  $\frac{1}{10}$
14. In a certain school, 74% students like cricket, 76% students like football and 82% like tennis. Then all the three sports are liked by at least  
A. 68%   B. 32%   C. 77%   D. 36%
15. Let  $S_n$  be the sum of all integers  $k$  such that  $2^n < k < 2^{n+1}$ , for  $n \geq 1$ . Then 9 divides  $S_n$  if and only if  
A.  $n$  is odd   C.  $n$  is of the form  $3k + 1$   
B.  $n$  is even   D.  $n$  is of the form  $3k + 2$

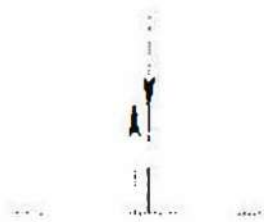
## PHYSICS

- 16 A boy standing on the footpath tosses a ball straight up and catches it. The driver of a car passing by moving with uniform velocity sees this.

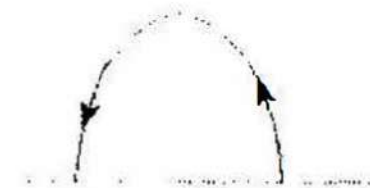


The trajectory of the ball as seen by the driver will be

A.



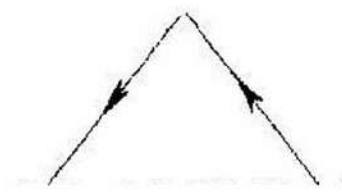
C.



B.



D.



- 17 Consider two spherical planets of same average density. Planet 2 is 8 times as massive as planet 1. The ratio of the acceleration due to gravity of the second planet to that of the first is

A. 1

B. 2

C. 4

D. 8

- 18 Two immiscible liquids, A and B are kept in a U-tube. If the density of liquid A is smaller than the density of liquid B, then the equilibrium situation is

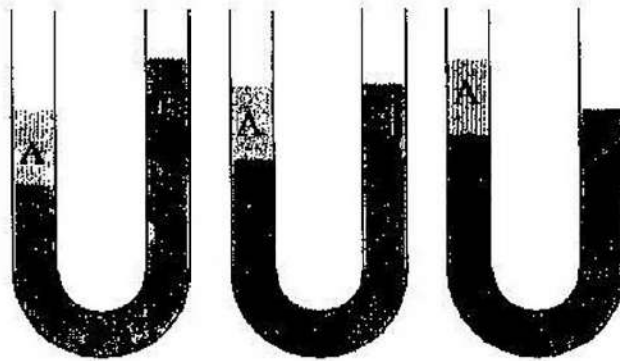
A.

B.

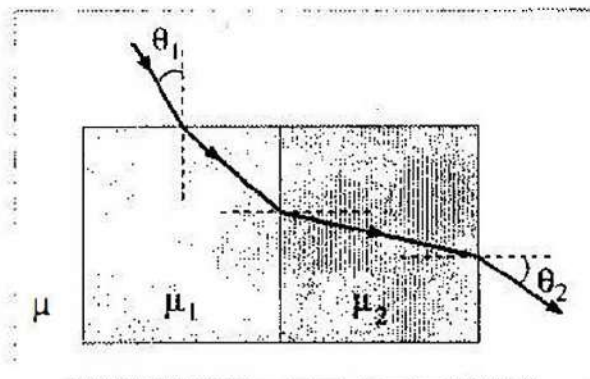
C.

D.

None of these



- 19 In the figure below, a ray of light travelling in a medium of refractive index  $\mu$  passes through two different connected rectangular blocks of refractive indices  $\mu_1$  and  $\mu_2$  ( $\mu_2 > \mu_1$ ).



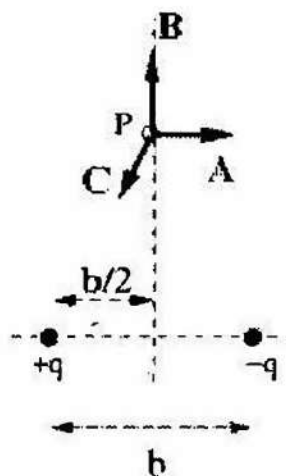
The angle of incidence  $\theta_1$  is increased slightly. The angle  $\theta_2$

- A. increases.
- B. decreases.
- C. remains the same.
- D. increases or decreases depending on the value of  $(\mu_1/\mu_2)$ .

- 20 Two charges of same magnitude move in two circles of radii  $R_1 = R$  and  $R_2 = 2R$  in a region of constant uniform magnetic field  $\vec{B}_0$ . The work  $W_1$  and  $W_2$  done by the magnetic field in the two cases, respectively, are such that

- A.  $W_1 = W_2 = 0$                       C.  $W_1 = W_2 \neq 0$   
 B.  $W_1 > W_2$                       D.  $W_1 < W_2$

- 21 Two charges  $+q$  and  $-q$  are placed at a distance  $b$  apart as shown in the figure below.



The electric field at a point  $P$  on the perpendicular bisector as shown is:

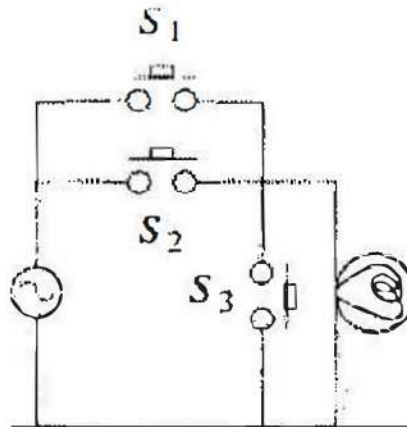
- A. along vector  $\vec{A}$                       C. along vector  $\vec{C}$   
 B. along vector  $\vec{B}$                       D. zero.
- 22 A block of mass  $M$  is at rest on a plane surface inclined at an angle  $\theta$  to the horizontal. The magnitude of force exerted by the plane on the block is

- A.  $Mg \cos \theta$                       C.  $Mg \tan \theta$   
 B.  $Mg \sin \theta$                       D.  $Mg$

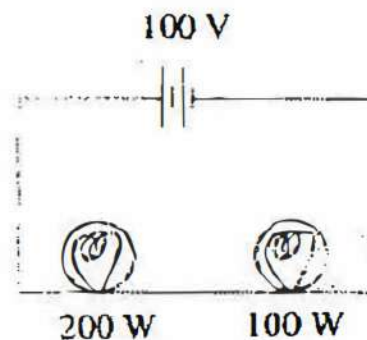


- 23 We are able to squeeze snow and make balls out of it because of
- anomalous behaviour of water.
  - large latent heat of ice.
  - large specific heat of water.
  - low melting point of ice.
- 24 Which of the following phenomena can be demonstrated by light, but not with sound waves in an air column?
- Reflection
  - Refraction
  - Diffraction
  - Polarization
- 25 The temperature of a metal coin is increased by  $100^{\circ}\text{C}$  and its diameter increases by 0.15%. Its area increases by nearly
- 0.15%
  - 0.60%
  - 0.30%
  - 0.0225%
- 26 The note "Saa" on the Sarod and the Sitar have the same pitch. The property of sound that is most important in distinguishing between the two instruments is
- fundamental frequency
  - intensity
  - displacement amplitude
  - waveform
- 27  $^{235}_{92}\text{U}$  atom disintegrates to  $^{207}_{82}\text{Pb}$  with a half-life of  $10^9$  years. In the process it emits 7 alpha particles and  $n$   $\beta^-$  particles. Here  $n$  is
- 7
  - 3
  - 4
  - 14

- 28 Consider the circuit below. The bulb will light up if



- A.  $S_1$ ,  $S_2$  and  $S_3$  are all closed.
  - B.  $S_1$  is closed but  $S_2$  and  $S_3$  are open.
  - C.  $S_2$  and  $S_3$  are closed but  $S_1$  is open.
  - D.  $S_1$  and  $S_3$  are closed but  $S_2$  is open.
- 29 Two bulbs, one of 200W and the other of 100W, are connected in series with a 100V battery which has no internal resistance. Then,



- A. the current passing through the 200W bulb is more than that through the 100W bulb.
- B. the power dissipation in the 200W bulb is more than that in the 100W bulb.



- C. the voltage drop across the 200W bulb is more than that across the 100W bulb.
- D. the power dissipation in the 100W bulb is more than that in the 200W bulb.
- 30 A solid cube and a solid sphere of identical material and equal masses are heated to the same temperature and left to cool in the same surroundings. Then
- A. the cube will cool faster because of its sharp edges.
- B. the cube will cool faster because it has a larger surface area.
- C. the sphere will cool faster because it is smooth.
- D. the sphere will cool faster because it has a larger surface area.

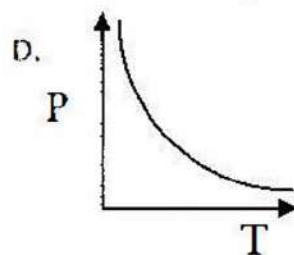
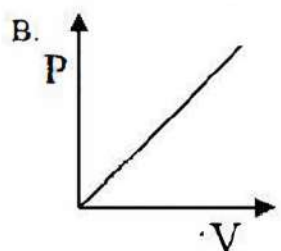
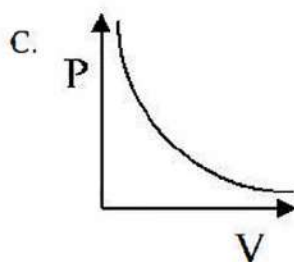
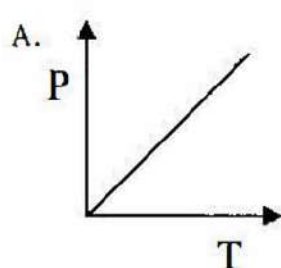
### CHEMISTRY

- 31 The element X which forms a stable product of the type  $\text{XCl}_4$  is
- A. Al                      B. Na                      C. Ca                      D. Si
- 32 A mixture of  $\text{NH}_4\text{Cl}$  and  $\text{NaCl}$  can be separated by
- A. Filtration                      C. Sublimation
- B. Distillation                      D. Decantation
- 33 The pair in which the first compound is ionic and the second compound is covalent, is
- A.  $\text{Fe}(\text{OH})_2$ ,  $\text{CH}_3\text{OH}$                       C.  $\text{CH}_3\bullet\text{H}$ ,  $\text{CH}_3\text{CH}_2\text{OH}$
- B.  $\text{Fe}(\text{OH})_2$ ,  $\text{Cu}(\text{OH})_2$                       D.  $\text{Ca}(\text{OH})_2$ ,  $\text{Cu}(\text{OH})_2$
- 34 In the reaction  $\text{SO}_2 + 2\text{H}_2\text{S} \rightarrow 3\text{S} + 2\text{H}_2\text{O}$ , the substance that is oxidized is
- A.  $\text{SO}_2$                       B.  $\text{H}_2\text{O}$                       C. S                      D.  $\text{H}_2\text{S}$

35 Sodium oxide dissolves in water to give sodium hydroxide which indicates its

- A. acidic character
- B. basic character
- C. amphoteric character
- D. ionic character

36 For an ideal gas, Boyle's law is best described by



37 The pH values of (i) 0.1 M  $\text{HCl}_{\text{aq}}$ , (ii) 0.1 M  $\text{KOH}$ , (iii) tomato juice and (iv) pure water follow the order

- A. (i) < (iii) < (iv) < (ii)
- B. (iii) < (i) < (iv) < (ii)
- C. (i) < (ii) < (iii) < (iv)
- D. (iv) < (iii) < (ii) < (i)

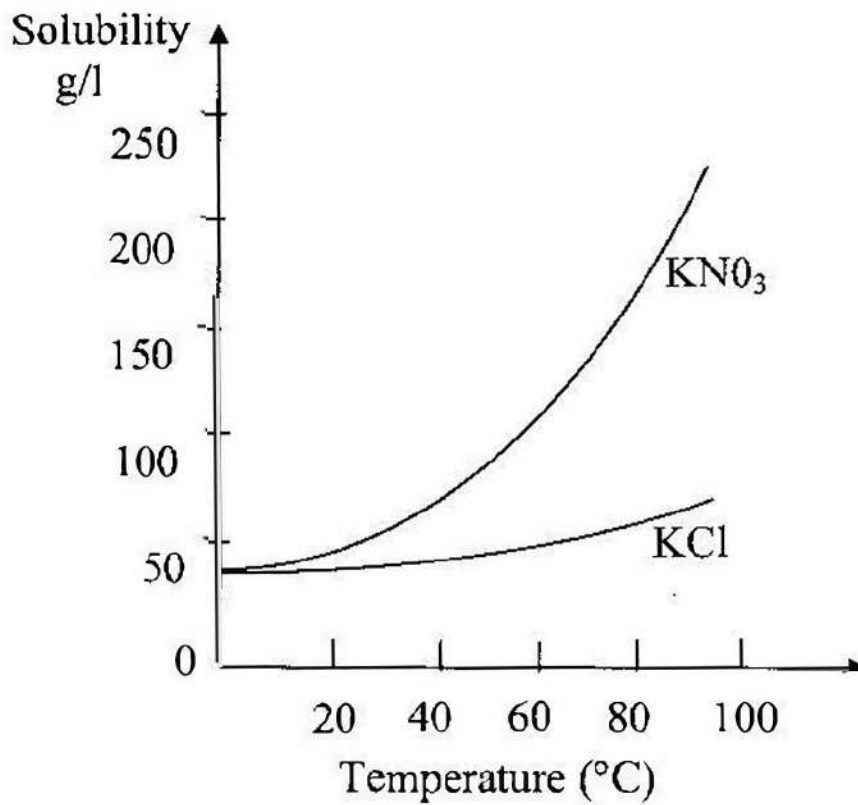
38 When calcium carbide is added to water, the gas that is evolved is

- A. carbon dioxide
- B. hydrogen
- C. acetylene
- D. methane

39 The atomic radii of the alkali metals follow the order

- A.  $\text{Li} > \text{Na} > \text{K} > \text{Cs}$
- B.  $\text{K} > \text{Cs} > \text{Li} > \text{Na}$
- C.  $\text{Na} > \text{K} > \text{Cs} > \text{Li}$
- D.  $\text{Cs} > \text{K} > \text{Na} > \text{Li}$

- 40 The number of possible structural isomers of  $C_3H_4$  is
- A. 1                      B. 2                      C. 3                      D. 4
- 41 Among the four compounds: (i) acetone, (ii) propanol, (iii) methyl acetate and (iv) propionic acid, the two that are isomeric are
- A. methyl acetate and acetone  
B. methyl acetate and propanol  
C. propionic acid and methyl acetate  
D. propionic acid and acetone
- 42 One mole of nitrogen gas on reaction with  $3.01 \times 10^{23}$  molecules of hydrogen gas produces
- A. one mole of ammonia  
B.  $2.0 \times 10^{23}$  molecules of ammonia  
C. 2 moles of ammonia  
D.  $3.01 \times 10^{23}$  molecules of ammonia
- 43 Saponification is
- A. hydrolysis of an ester  
B. hydrolysis of an amide  
C. hydrolysis of an ether  
D. hydrolysis of an acid chloride
- 44 A concentrated solution of lead nitrate in water can be stored in
- A. an iron vessel                      C. a zinc vessel  
B. a copper vessel                      D. a magnesium vessel



Given the solubility curves of KNO<sub>3</sub> and KCl, which of the following statements is not true?

- A. At room temperature the solubility of KNO<sub>3</sub> and KCl are not equal
- B. The solubilities of both KNO<sub>3</sub> and KCl increase with temperature
- C. The solubility of KCl decreases with temperature
- D. The solubility of KNO<sub>3</sub> increases much more compared to that of KCl with increase in temperature

## BIOLOGY

- 46 Which one of the following is the smallest in size?
- A. Bacteria
  - B. Mitochondrion
  - C. Mammalian cell
  - D. Virus
- 47 If birds are moved from  $30^{\circ}\text{C}$  to  $10^{\circ}\text{C}$ , their body temperature
- A. changes from  $30^{\circ}\text{C}$  to  $10^{\circ}\text{C}$
  - B. increases by  $10^{\circ}\text{C}$
  - C. does not change at all
  - D. decreases by  $10^{\circ}\text{C}$
- 48 Ascorbic acid is a/an,
- A. Strong inorganic acid
  - B. Hormone
  - C. Vitamin
  - D. Enzyme
- 49 Bile salts,
- A. break down polypeptide chains
  - B. emulsify fats and solubilize them
  - C. digest fats
  - D. help breakdown of polysaccharides
- 50 Dietary fibers are composed of,
- A. Cellulose
  - B. Proteins
  - C. Amylose
  - D. Unsaturated fats
- 51 'On the origin of species, by means of Natural selection' was written by,
- A. Hugo de Vries
  - B. Charles Darwin
  - C. Charles Dickens
  - D. Alfred Russell Wallace
- 52 Unlike humans, dogs cannot perspire to get rid of excess metabolic heat. They lose metabolic heat by,

- A. panting  
B. taking a bath  
C. running in windy conditions  
D. rolling in the mud
- 53 Haemodialysis is a treatment option for patients with malfunctions of,
- A. Kidney  
B. Liver  
C. Heart  
D. Lungs
- 54 An individual has O blood group if his/her blood sample,
- A. Clumps only when antiserum A is added  
B. Clumps only when antiserum B is added  
C. Clumps when both antiserum A and antiserum B are added  
D. Does not clump when either antiserum A or antiserum B is added
- 55 In warmer weather, curds from milk forms faster because,
- A. Bacteria diffuse better in warmer milk  
B. The rate of bacterial multiplication increases  
C. Lactogen is better dissolved  
D. It is easier to separate protein from water
- 56 Seedlings grown in dark are
- A. similar to those grown in light  
B. taller than those grown in light  
C. shorter than those grown in light  
D. they don't grow at all
- 57 In humans, Rhesus condition can arise when,
- A. father is Rh+ and mother is Rh-  
B. father is Rh- and mother is Rh+  
C. either father or mother is Rh+



D. either father or mother is Rh-

58 The part of the human brain that governs memory and intelligence is,

A. Cerebrum

C. Hypothalamus

B. Medulla

D. Cerebellum

59 Saturated dietary fats increase the risk of heart disease by,

A. widening arteries by thinning their walls

B. narrowing veins by carbohydrate deposition

C. narrowing arteries by fat deposition

D. narrowing arteries by carbohydrate deposition

60 Rotation of crops is carried out to,

A. increase variation in the mineral content of the soil

B. increase diversity of plant habitats

C. increase in nitrogen content of the soil

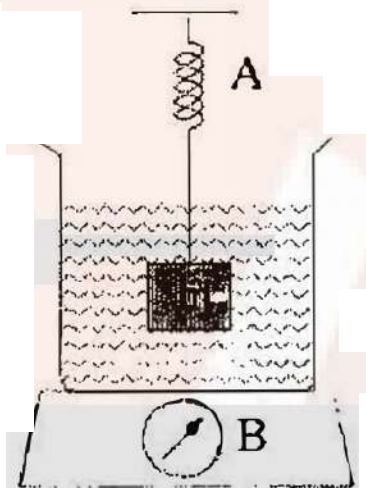
D. increase convenience for the farmer

**PART-II (2 Marks)**  
**MATHEMATICS**

61. Let  $\log_a b = 4$ ,  $\log_c d = 2$  where  $a, b, c, d$  are natural numbers. Given that  $b - d = 7$ , the value of  $c - a$  is  
A. 1   B. -1   C. 2   D. -2
62. Let  $P(x) = 1 + x + x^2 + x^3 + x^4 + x^5$ . What is the remainder when  $P(x^{12})$  is divided by  $P(x)$ ?  
A. 0   B. 6   C.  $1 + x$    D.  $1 + x + x^2 + x^3 + x^4$
63. In a triangle  $ABC$ , the altitudes from  $B$  and  $C$  on to the opposite sides are not shorter than their respective opposite sides. Then one of the angles of  $ABC$  is  
A.  $30^\circ$    B.  $45^\circ$    C.  $60^\circ$    D.  $72^\circ$
64. In a triangle  $ABC$ ,  $AB = AC = 37$ . Let  $D$  be a point on  $BC$  such that  $BD = 7$ ,  $AD = 33$ . The length of  $CD$  is  
A. 7   B. 11   C. 40   D. not determinable
65. A line segment  $l$  of length  $a$  cm is rotated about a vertical line  $L$  keeping the line  $l$  in one of the following three positions: (I)  $l$  is parallel to  $L$  and is at a distance of  $r$  cm. from  $L$ ; (II)  $l$  is perpendicular to  $L$  and its mid-point is at a distance  $r$  cm. from  $L$ ; (III)  $l$  and  $L$  are in the same plane and  $l$  is inclined to  $L$  at an angle  $30^\circ$  with its mid-point at a distance  $r$  cm. from  $L$ . Let  $A_1, A_2, A_3$  be the areas so generated. If  $r > (a/2)$ , then  
A.  $A_1 < A_3 < A_2$    C.  $A_2 < A_1 < A_3$   
B.  $A_1 = A_3 < A_2$    D.  $A_1 = A_2 = A_3$

## PHYSICS

- 66 A spring balance A reads 2 kg with a block of mass  $m$  suspended from it. Another balance B reads 3 kg when a beaker with a liquid is put on its pan. The two balances are now so arranged that the hanging mass  $m$  is fully immersed inside the liquid in the beaker as shown in the figure. In this situation



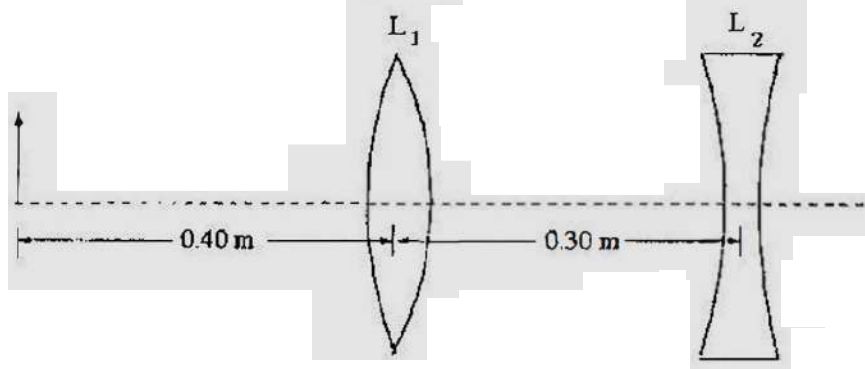
- A. the balance A will read 2 kg and B will read 5 kg.  
 B. the balance A will read 2 kg and B will read 3 kg.  
 C. the balance A will read less than 2 kg and B will read between 3 kg and 5 kg.  
 D. the balance A will read less than 2 kg and B will read 3 kg.
- 67 According to the quantum theory, a photon of electromagnetic radiation of frequency  $\nu$  has energy  $E = h\nu$  where  $h$  is known as Planck's constant. According to the theory of relativity, a particle of mass  $m$  has equivalent energy  $E = mc^2$ , where  $c$  is speed of light. Thus a photon can be treated as a particle having effective mass  $m = \frac{h\nu}{c^2}$ . If a flash of light is sent horizontally in earth's gravitational field, then photons while traveling a horizontal distance  $d$  would fall through a distance given by

- A.  $\frac{gd^2}{2c^2}$       B.  $\frac{h}{mc}$       C.  $\frac{mcd^2}{h}$       D. zero

68 A solid square plate is spun around different axes with the same angular speed. In which of the following choice of axis of rotation will the kinetic energy of the plate be the largest?

- A. Through the center normal to the plate.  
 B. Along one of the diagonals of the plate.  
 C. Along one of the edges of the plate.  
 D. Through one corner normal to the plate.

69 An object is placed 0.40 m from one of the two lenses  $L_1$  and  $L_2$  of focal lengths 0.20 m and 0.10 m respectively as depicted in the figure. The separation between the lenses is 0.30 m.

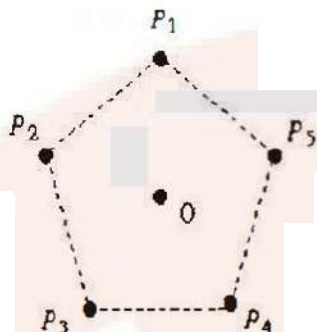


The final image formed by this two lens system is at

- A. 0.13 m to the right of the second lens.  
 B. 0.05 m to the right of the second lens.  
 C. 0.13 m to the left of the second lens.  
 D. Infinity.

70 5 charges each of magnitude  $10^{-5}\text{C}$  and mass 1 kg are placed (fixed) symmetrically about a movable central charge of magnitude  $5 \times 10^{-5}\text{C}$

and mass 0.5 kg as shown. The charge at  $P_1$  is removed. The acceleration of the central charge is



[Given:  $OP_1 = OP_2 = OP_3 = OP_4 = OP_5 = 1\text{m}$ ;  $\frac{1}{4\pi\epsilon_0} = 9 \times 10^9$  in SI units.]

A.  $9\text{ m s}^{-2}$  upwards.

B.  $4.5\text{ m s}^{-2}$  upwards.

C.  $9\text{ m s}^{-2}$  downwards.

D.  $4.5\text{ m s}^{-2}$  downwards.

### CHEMISTRY

- 71 Reaction of  $\text{NaCl}$  with conc.  $\text{H}_2\text{SO}_4$  liberates a gas, **X** that turns moist blue litmus paper red. When gas **X** is passed into a test tube containing egg shell powder suspended in water another gas, **Y** is generated which when passed through lime water makes it milky. The gases **X** and **Y**, respectively, are

A.  $\text{HCl}$  and  $\text{CO}_2$

B.  $\text{SO}_2$  and  $\text{CO}_2$

C.  $\text{Cl}_2$  and  $\text{CO}_2$

D.  $\text{SO}_2$  and  $\text{HCl}$

- 72 10 ml of an aqueous solution containing 222 mg of calcium chloride (mol. wt. = 111) is diluted to 100 ml. The concentration of chloride ion in the resulting solution is

A. 0.02 mol/lit

B. 0.01 mol/lit

C. 0.04 mol/lit

D. 2.0 mol/lit



- 73 Aluminium reduces manganese dioxide to manganese at high temperature. The amount of aluminium required to reduce one gram mole of manganese dioxide is
- A.  $1/2$  gram mole      C. 1 gram mole  
B.  $3/4$  gram mole      D.  $4/3$  gram mole
- 74 Ethanol on reaction with alkaline  $\text{KMnO}_4$  gives X which when reacted with methanol in the presence of an acid gives a sweet smelling compound, Y. X and Y, respectively, are
- A. acetaldehyde and acetone  
B. acetic acid and methyl acetate  
C. formic acid and methyl formate  
D. ethylene and ethyl methyl ether
- 75 The pH of a 10 ml aqueous solution of HCl is 4. The amount of water to be added to this solution in order to change its pH from 4 to 5 is
- A. 30 ml      C. 90 ml  
B. 60 ml      D. 120 ml

**BIOLOGY**

- 76 Proteins are synthesized on,
- A. Cytoskeleton      C. Ribosomes  
B. Mitochondria      D. Golgi apparatus
- 77 Which of the following allows light to focus in visual perception?
- A. Retina      C. Retinal pigment  
B. Iris      D. Cornea



- 78 During cell division if there is one round of chromosome duplication followed by one round of cell division, the number of chromosomes the daughter cells will have as compared to the mother is,
- A. equal      B. double      C. half      D. one fourth
- 79 Similar type of vegetation can be observed, in the same,
- A. latitude      B. longitude      C. country      D. continent
- 80 Which of the following ecological food chain does NOT represent an erect pyramid of numbers?
- A. Grass-Rodent-Snake      C. Grass-Deer-Tiger  
B. Tree-Bird-Avian parasite      D. Insect-Chicken-Human