प्रश्न-पत्र बोर्ड द्वारा निर्धारित पुस्तकों से ही सैट किया जाएगा। कृपया अध्यापकगण बोर्ड द्वारा निर्धारित पुस्तकें ही पढ़ाएँ।

<u>DESIGN</u> OUESTION PAPER

	2	<u>DUESTION</u>	PAPER		
	Subject :	Chemistr	y		
	Paper : Class :	Annual of 12 th	r Supplementary	2010.10	
	Time :	3 Hrs.	Year	2018-19	
	Marks :	. 60			
1. Weightage to Objecti Objective	ves: K	U		~	_
Percentage of marks	40	30	A 20	S	Tota
Marks	24	18	12	10 6	100 60
			12	V	00
2. Weightage to Form of Forms of Questions	f Questions: E	SA	VSA	0	m .
No. of Questions	03	05	VSA 09	O 12	Tota 29
Marks Allotted	15	15	18	12	60
Estimated Time	65	50	45	20	180
3. Weightage to Conte	ent:				100
	<u>Un</u>	its/Sub-Units			<u>Marks</u>
1. Solid State					4
2. Solution					4
3. Electro Chem	Electro Chemistry				
4. Chemical Kin	Chemical Kinetics				
5. Surface Chem	Surface Chemistry				
6. General Princ	General Principles & Processes of Isolation of Element				
7. p-Block Elem	p-Block Elements				
8. d&fBlock E	d & f Block Elements				
9. Co-ordination	Co-ordination Compounds				
10 Halo Alkanes and Halo Arenes					4
11. Alcohols, Phe	1. Alcohols, Phenols and Ethers				
12 Aldehydes, K	2 Aldehydes, Ketones and Carboxylic Acids				
13. Amines and D	Amines and Diazonium Salts				4
14. Biomolecules	Biomolecules				3
15. Polymers	5. Polymers				2
16. Chemistry in Every Day Life					2
				Total	60
. Scheme of Sect	ions :	<u> </u>	X		
. Scheme of Option		rnal Choice in I	long answer question	i.e. essay typ	oe in
. Difficulty level	: Diff	icult : <u>10</u>	<u>%</u> marks		
	Ave	rage : <u>50</u>	% marks		
	Easy	y : <u>40</u>	_% marks		

Abbreviations: K (Knowledge), U (Understanding), A (Application), S (Skill), E (Essay Type), SA(Short Answer Type), VSA (Very Short Answer Type), O (Objective Type)