II YEAR CHEMISTRY IPE QUESTION PAPER - JUNE 2009

SECTION - A

I. Answer all the following:

 $10 \times 2 = 20$

- 1. What is meant by Frenkel and Schottky defects?
- 2. What is Doping?
- 3. Write the systematic names for the following.

(a)
$$\Big\lceil Cr \Big(H_2O\Big)_6 \Big\rceil Cl_3$$
 (b) $K_2 \Big[PtCl_4\Big]$

- 4. What is PHBV? How is it useful to man?
- 5. Give the deficiency diseases caused by the vitamins A, C, D and K.
- 6. What is Peptide linkage? Give an example.
- 7. Define Antihistamines. Give an example.
- 8. How is paracetamol is prepared? Give equation.
- 9. What is Chloropicrin? How it is formed from chloroform? Give equation.
- 10. The reactant 'A' is reduced to Azobenzene with $LiAlH_4$. What is 'A'? Give equation.

SECTION - B

II. Answer any six of the following:

 $6 \times 4 = 24$

- 11. What is Colligative property? Give any three examples.
- 12. Define pH ? What is the pH of 0.005M $Ba(OH)_2$ solution ?
- 13. What is homogeneous and heterogeneous catalysis? Give an example for each.
- 14. State Hess's law of constant heat summation and explain with example.
- 15. Explain how superphosphate of lime is manufactured.
- 16. How is bauxite purified by Serpeck's process?
 - 17. Define first law of Faraday's electrolysis.A current of 10 amp is passed through molten $AlCl_3$ for 96.5 seconds. Calculate the mass of Al deposited at cathode. (Atomic weight of Al = 27)
 - 18. Explain any two characteristic properties of transition elements.

III. Answer any two of the following:

 $2 \times 8 = 16$

- 19. (a) How is bleaching powder prepared industrially?
 - (b) Give the reactions of ozone with the following and give equations.
 - (i) SO_2
- (ii) PbS
- (iiii) BaO_2
- (iv) Hg
- 20. Define Rate of reaction. Explain the factors which influence rate of reaction.
- 21. Explain the preparation of ethyl alcohol from molasses. What happens when ethyl alcohol is treated with
 - (a) CH_3MgI

- (b) PCl_5
- (c) Conc. H_2SO_4 at $140^{\circ}C$
- (d) CH_3COOH